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Physical simulations of molecular FCN technology



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"Everybody is a genius. But if you judge a fish by its ability to climb a tree it will live its whole life believing that it is stupid" Albert Einstein

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Summary

Nowadays the world is more "mobile" than ever. This mobility brings with it a very important technological problem: the miniaturization of electronic devices. Laptops and smartphones are more and more powerful meaning that bigger RAM and powerful processors must be placed within a small space. In order to achieve this goal, the miniaturization of single transistor might be the only solution. Until now the electronics industry has been based on Moore's Law which declare that "the number of transistors per square inch on integrated circuits had doubled every year since their invention". However, the ITRS roadmap predicted in 2015 that Moore's law will die in 2020, as matter of fact, we have now reached the physical limit of the semiconductor-based technology (few nanometer). There are some unmanageable problems at this level of miniaturization, for example the high power consumption due to leakage current. To overcome this problem we are aiming at other technologies, one of the most interesting is FCN (Field Coupled Nano-Computing): with this technology no more transistors, currents and voltage are used to transfer the information but alternately, via local field interactions between blocks of the circuit. In particular, the mQCA (molecular Quantum cellular automata) technology was studied in this thesis. Many solutions were proposed for QCA technology, for example semiconductor QCA, metal-dot QCA and magnetic-QCA. These technologies present some problems such as the working temperature and the working frequencies and for these reasons they have been left out. Therefore, the molecular technology is the most promising for QCA paradigms. The basic elements of this technology are the molecules, which can store information through the arrangement of the charge inside them. The basic cell is composed by two equal molecules. Each molecule has 2 or 3 redox centers, now called dots, in which is more probable that an electron is attracted or released. Furthermore, a basic cell is composed by 4 or 6 dots, and the charge displacement inside the cell determines its logic state. This cell has been designed to be bistable, then there are two states with lower energy and in these two states we encode the "0" or "1" logic. A clock field is needed to help the switching of the molecules. In literature many molecules have been proposed as candidates, in particular diallyl butane (4-dot molecule) whose behavior is considered the ideal one for this technology, but it presents two main problem: the absence of the third dot necessary for the clock and the absence of the molecular binding element necessary for physical implementation. The University of Bologna and Polytechnic of Torino synthesized a new molecule ad-hoc for QCA computation: the bis-ferrocene molecule. In this work, three main molecules (diallyl butane, decatriene and bisferrocene) has been studied and analyzed throughout several ab-initio simulations with a computational chemistry software (GAMESS). This is the first time that this software has been used, in fact the software which has been used for this kind of work was gaussian09. For this reason, the first step of this thesis was studying and writing the manual of this software: all commands used in this work has been written and explained, for example the resolution method for Schrodinger equation (RHF, UHF, DFT), the basis-set (6,31g(d, p)), the total charge and so on. Then some simulations have been done to compare the results obtained by the two software to see if they converge. In particular the electric potential and the charge displacement were compared for the diallyl butane and decatriene molecules.

After this first step, the three candidate molecules have been analyzed. The steps were the same for each molecule:

- analysis of the molecule capability to encode binary information at equilibrium;
- simulation of the molecules with biasing conditions, in particular:
 - an external electric field (switching field);
 - a write-in system made by point charges;
 - clock field;
- post-processing with Matlab and C program.

The C program was written ad-hoc for this work: it reads the output file from the simulations then finds the atomic charges in this file and sums them to evaluate the dot charges. The first molecule studied was the diallyl butane: in its oxidized form it presents the perfect behavior for QCA computation. For this reason, it can be used as reference for other simulations. Then the bis-ferrocene molecule was simulated both in its neutral and oxidized version. In its neutral form, the bis-ferrocene is perfectly symmetrical, but the problem is that the slope is very small and consequently the switch also. To obtain a bigger slope, then, the molecule was oxidized by raising an electron from the HOMO (Highest Occupied Molecular Orbital) setting ICHARG = 1 MULT = 2. In this case, unlike gaussian09 which divided the charge equally between the two active ferrocene dots, GAMESS adds extra free charge only in one of the two dots. For this reason, the molecule is polarized at the equilibrium, this means that the molecule is not symmetrical. The difference in the results obtained with gaussian09 is due to the resolution method used and to the different basis-set. In fact, by setting two identical simulations

on both software (basis-set LANL2DZ and method DFT), GAMESS cannot solve the Schrödinger equation. Precisely for this reason the base 6,31-g (d, p) and the UHF method were used. Setting the potential (positions of the nuclei) and adding a number of electrons (with ICHARG), then the program tells you where the electrons are going to by solving the Schr^dinger equation to some approximation (and under some restrictions, say total spin multiplicity). As can be seen from the figure, in these conditions the symmetry of the molecule is shifted to the right, in particular there is a switch from one state to another with an external electric field of 4.5 V/nm, this has a consequence also for simulations with the charge points. In fact, since the charge points placed at their ideal distance of 1 nm (the same distance of the two active dots) generate an electric field of just $0.5 \,\mathrm{V/nm}$, it does not allow the molecule to change the logic state. To allow this, it was found through iterative simulations that the distance from the molecule where to place the charge points is 0.3 nm. As future works, it would be interesting to study how the two software (GAMESS and gaussian09) manage the oxidation of the molecule, in particular GAMESS. Moreover, it would be interesting to study the molecules from a dynamic point of view during the switch, to simulate a QCA wire from a point of view of charge distribution and energy behavior.

Abstract

The work in this thesis is organized as follow:

• Chapter 1: Introduction:

This chapter introduces the physical principles of QCA technology, in particular the clock system which is needed for the transportation of the information.

• Chapter 2: Molecular QCA:

Many solutions were proposed for the QCA technology. For instance, metaldot solution and semiconductor-solution which had the problem of working temperature (few mK). Another idea was the magnetic-solution, it seems promising but it has the problem of working frequency (too low). For these reason, the molecular-solution is considered the most promising and realistic solution to QCA technology. 3 candidate molecules are shown, in particular the bis-ferrocene molecule in which is focused this work.

• Chapter 3: Methodology:

In this chapter the methodology of this work thesis is shown. The steps was: studying neutral molecule at the equilibrium, with an horizontal external electric field (*switching field*) and with the point charges to approximate a write-in system; then a vertical clock field is applied to the molecules; finally all these simulations are repeated with the oxidized molecole (total charge equal to 1). The results of these simulations was shown in Chapter 6.

• Chapter 4: Tools Manual for MQCA simulation:

The software used in this work was mainly 2: GAMESS and gaussian09. This chapter shows a complete manual for these 2 software which are the most famous for the computational chemistry. All commands to be given as input to the program are described in detail, complete with input examples and explanations on how to read the output files. In addition, there is also a small manual of all programs used for post-processing (eg Avogadro, MacPlot). At the end of this chapter, the reader will be able to simulate accurately the molecular systems he or she prefers.

• Chapter 5: Gamess vs Gaussian09:

As the main objective of this thesis was to find an alternative program to gaussian09, the results obtained using GAMESS were compared with the results obtained previously with gaussian. The molecules tested for this purpose were Diallyl Butane and Decatriene in different conditions (in equilibrium or in external electric field conditions).

- Chapter 6: Charge distribution: In this chapter the methodological steps explained in chapter 3 have been followed for each of the three candidate molecules. The diallyl butane molecule, which is also the simplest one, was studied for first, then the decatriene and the bis-ferrocene. For all three, both the neutral and the oxidized form were studied and a quick comparison was made with the results obtained with gaussian09. In particular the study is focused on the bis-ferrocene molecule.
- Chapter 7: Conclusions and future study: Some ideas for future studies are proposed.

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Chapter 1

Introduction

Since the invention of the first transistor in 1947 at Bell Labs, the microelectronics industry has followed *Moore's Law* [1]. This law takes its name from the engineer that enunciated it, Gordon Moore. Moore noted that the number of transistors on a chip had doubled in a time between 18 and 24 months. From there he formulated a hypothesis that semiconductor would double their capacity every 18 months, the original Moore's Law is shown in Fig. 1.1.



Figure 1.1: Original Moore's law [1]

This means that an integrated circuit would double the number of transistors it contains every 18 months and to follow this trend the area of single transistor would have to decrease. One consequence of Moore's Law is that the length of a transistor (as a channel length) decreases by 30% every two years, as shown in Fig. 1.2.





Figure 1.2: Channel lenght of a single transistor over the years [2]

Since the early '70s, the device of choice for high levels of integration has been the field effect transistor (FET). At gate lengths below 0.1 µm FETs will begin to encounter fundamental effects that make further scaling difficult. A possible way for the microelectronics industry to maintain growth in device density is to change from the FET-based paradigm to one based on nanostructures. One nanostructure paradigm, proposed by Lent in the early '90s is the quantum-dot cellular automata (QCA) [3],[4],[5],[6] which employs arrays of coupled quantum dots to implement Boolean logic functions. The advantage of QCA lies in the extremely high packing densities possible due to the small size of the dots, the simplified interconnection, and the extremely low power-delay product.

1.1 QCA

The ideal basic QCA cell is made of coupled quantum dots and a few free charges in a square array coupled by tunnel barriers. It consists of four quantum dots in the corners (called *working dots* or *active dots*) and two free charges in the middle. Let's consider half a QCA cell made by 2 active dots and one free charge, as shown in Fig. 1.3



Figure 1.3: Basic half cell for QCA.

The electrons (as the charge) can stay in one of them depending on the external conditions. So, aligning another half QCA to obtain a complete cell, the electrons

in the first half cell force the electrons in the second half cell in the dots on opposite corners. For example, in Fig. 1.4 is shown what happens if two half cells with the charge in the same active dot are aligned.



Figure 1.4: Re-arrangement of the charge between two nearby half cell.

What happens is the the charge in the second half cell re-arranges itself. This is due to the Coulomb interaction between them and to minimize the energy of the cell. It is possible to have three different configurations for one QCA cell, which are shown in Fig. 1.5. Each configuration can encode the logic "0" or "1" if the free charges are confined in two opposite dots, while if the electrons are in the middle the state encoded is a *NULL state*. Electrons are able to tunnel between the dots, but cannot leave the cell.



Figure 1.5: Basic cell and logic state encoding for QCA.

Two contiguous cells are placed at the same distance d that is between the two active dots of a half cell, as shown in Fig. 1.6. In this condition, the communication between two nearby cells is due to the Coulomb interaction between the working dots at the edge of the two nearby cells leading to a charge re-arrangement inside the second cell according to the previous cell state, as seen for two half cells.



Figure 1.6: Interaction between QCA cells: re-arrangement of the charge in the second cell due to the Coulomb force.

Aligning many QCA cells we obtain the simplest and most fundamental block for QCA: a wire. In other words, a wire is simply a line of QCA cells, as in Fig. 1.7.



Figure 1.7: QCA wire.

At first, the line is at the ground state '0'. An input, in this case '1', is applied to the left of the line and it forces the first cell to one polarization. Now the first and the second cell are of opposite polarization, '1' for the former and '0' for the latter. This means that two electrons are close together, the line is in a higher energy state and all successive cells must change their polarization to reach the new ground state.

QCA cells can implement all logic functions [7], multiplexor [8], alu [7], [9], [10] and also microprocessors [11],[12]. The basic logic functions NOT, AND and OR are shown in Fig. 1.8 and 1.9.



Figure 1.8: QCA inverter.



Figure 1.9: QCA AND or OR implemented by majority gate.

А	В	С	Output
0	0	0	0
0	0	1	0
0	1	0	0
0	1	1	1
1	0	0	0
1	0	1	1
1	1	0	1
1	1	1	1

Table 1.1: Majority gate truth table.

In the inverter, the input is split into two lines and then put together in a line placed at 45° angle with the two previous lines. This particular angle produces an opposite polarization to that in the two lines. For implementing AND and OR gates a topology called *majority gate* [7] is used. There are three inputs A, B and C. The central cell will have the polarization which prevails on the three inputs. Finally the polarization of the central cell is propagated at the output. One of the inputs can be used for selecting the function of the device, AND or OR. All combinational logic functions can be implemented by using only these 2 QCA blocks.

1.2 Clock system

In the previous paragraph we saw how a bit encoded in a molecule can be transmitted between two nearby molecules. However, this is not possible without a *clock* signal. The problem is that this clock signal can't be abrupt, in fact authors in [13] demonstrated that an abrupt clock can lead to metastability problems along the QCA circuit. For this reason, a new type of clock was developed for QCA technology, based on *Adiabatic Switching* [13],[14],[15],[16], which is the quantum version of the adiabatic theorem. Physically the clock is multi-phase [13], that is it consists of four consecutive phases with a different phase in a quarter of the period with respect to the signal in the previous zone, as shown in Fig. 1.10.



Figure 1.10: QCA clock.

The QCA circuit needs to be divided into *clock zones*, and each zone has a different phase of the clock.

These four phases of the clock are:

- switch;
- hold;
- release;
- relax.

In the *switch* phase, the inter-dot potential barrier is reduced so that the free charges in the molecule can move in one of two active dots depending on the external conditions (proximity of other molecules, external electric field, write-in system and so on) encoding the binary information; however in the second phase, *hold*, the potential between the dot is increased so that the charges are forced to stand still in the dot, so the next molecule (which will have the clock on the switch phase) will reallocate its free charge and so will all the others. In the *release* phase, the dot barrier potentials are lowered so that free charges can return to the initial position (*NULL* state) thus the molecule returns to have minimal energy and finally the *relax* phase in which the molecule remains stable in this condition.

Chapter 2

Molecular QCA

The molecular solution for the QCA technology is just the latest technology proposed and the most promising. But, many solutions were proposed and implemented before in the last two decades. In this chapter, some examples are listed.

2.1 Metal-Dot Solution

The metal-dot implementation was the first fabrication technology created to demonstrate the concept of QCA computing paradigm. The structure of this solution is based on metal Al islands, as shown in Fig. 2.1, over a silicon dioxide (SiO_2) substrate. The cell needs 2 capacitors to prevent charge exchanging.



Figure 2.1: Metal-dot solution for QCA.

As we can see in the Fig. 2.2 we have four dots: two dots D_1 and D_2 in the left that can be associated with the others in the right $(D_3 \text{ and } D_4)$ by tunnel junctions which allow electrons to move between them [14], [5], [17].



Figure 2.2: The QCA cell viewed with the scanning electron micrograph. [17].

The schematic cell structure is shown in Fig. 2.3.



Figure 2.3: Schematic diagram of metal.-dot QCA cell.

This technology has two problems: the first one is that the metal islands are very large, so this device is not a nanodevice but it is in order to micrometers, the second problem is associated to the thermal energy. The metal-dots work only at the temperature of few mK because of quantum effects and so it is not realistic. Furthermore the maximum operating frequency is in the range of MHz.

In any case the principle has been demonstrated.

2.2 Semiconductor-Dot Solution

The second solution proposed in literature for the QCA was grounded on a semiconductor structure [13] because semiconductors' technology is a well established technology. For example, a structure based on GaAs with several interleaved layers of silicon, germanium or GaAs has been tested, as shown in Fig. 2.4. In this structure we have some dots that should be able to store charges. This structure, like the metal-dot solution has been tested but the results were not particularly successful because it should work at cryogenic temperature, it's highly defective and especially if nano-dimensions should be reached it has been proven that the defects prevent the feseability of this technology.



Figure 2.4: QCA cell implemented with semiconductor solution. [13].

2.3 Magnetic Solution

The real first successful implementation that is also currently adopted is the magnetic one. Here the cell is a nano-magnet [18] with a shape anisotropy in which the magnetization assumes a stable state in one direction or in the other direction, see Fig. 2.5.



Figure 2.5: Two possibily states of magnetization for a nano-magnet.

These two directions that the nano-magnets can assume encode the logic states in the cells. Instead of electron-tunneling effects, the term "Quantum" refers to the quantum-mechanical nature of the magnetic exchange interactions. This information propagates among magnetic QCA devices due to the magnetic field coupling interaction (ferromagnetic or antiferromagnetic).

In this case we need a clock, a magnetic clock field, which is able to erase the magnetization and forces it in an horizontal state. As soon as the clock is released,

the magnetization goes up how has been proved and it is promising at least in terms of power dissipation. Because we are talking about magnetism, the speed is not very high but at least it can be implemented at room temperature.

2.4 Molecular QCA

Of all the solutions offered for QCA technology, the most promising is definitely the molecular one [19], [20], [21], [22]. This is due to several factors: the size that is on nanometers, the operating temperature (works correctly at room temperature), frequencies that can reach computing (THz) and higher density of devices.

Dots are represented by redox centers, which act as a loading container. The choice of redox centers as dots is due to the fact that it is possible to add or remove an electron (thus reduce or oxidize the molecule) from the redox centers without breaking the chemical bonds [21], in fact a redox center is a site of the molecule where is more probable to attract or to release an electron. The molecule condenses a 0 or 1 logic depending on the charge present in the two dots, as shown in the figure. The two (or three) redox centers are connected via a tunneling path that allows the electrons to pass from one point to another. As we will see later, in this thesis work we have dealt with either a molecule with two dots or three dots, the difference being basically that the third dot is needed to implement a system with a clock and to encode the NULL state. In order to obtain a complete QCA cell, two molecules must be joined together, making the cell then composed of 6 dots.



Figure 2.6: 3-dot molecule scheme: logic state encoding.

2.4.1 Candidate molecules

In literature, many ideal molecules [20], [21], [22], [23], [24], [25] have been studied as candidate for QCA device. They are ideal molecules because their behaviour has been studied only by means of simulations and they have been never been physically implemented.

This thesis only studies three different molecules for the QCA technology. The simplest molecule proposed for QCA computing is the diallyl butane [22]. As shown

in Fig. 5.1 it has two allyl groups (circled in the figure), that represent the dots (so they are the redox centers). The molecule proposed in literature has a free positive charge (an electron is missing). As said before, the single molecule represents half a QCA cell, so the complete cell could be implemented aligning two single molecules to form a square cell.



Figure 2.7: Diallyl butane molecule.

The second molecule studied in this thesis is decatriene. The decatriene molecule is a three dot molecule. These three dots, that are circled in Fig. 2.8(A), are ethylene groups. The molecule in the figure is a single molecule, this means that it represent half a QCA cell, so the two nearby molecules have to be aligned in order to have a complete QCA cell, as in the previous case. The authors in [26] used the electrostatic potential surface to identify the charge localization inside the molecule and so for the encoding of the three stable states ('1', '0' and 'NULL'), Fig. 2.8(B).



Figure 2.8: Decatriene molecule structure (a), and logical state encoding using HOMO visulazitation (b).

The results discussed in [26] reveal that the decatriene molecule is a good candidate for QCA purpose, even though also in this case the molecule has no binding element necessary for physical implementation.

2.4.2 Bis-ferrocene molecule

The work of this thesis is focused on the bis-ferrocene molecule [27], [28], [29], [30], [31], which is considered the most promising molecule for the QCA technology. In

fact, this molecule has been synthesized by the University of Bologna and Politecnico di Torino ad hoc for QCA computation.

The structure of the molecule is reported in Fig 2.9: it consists in two ferrocenes and a central carbazole bridge.



Figure 2.9: Bis-ferrocene molecule: (a) molecular structure, (b) 3-dot scheme.

The two ferrocenes are redox centers and function as working dots, while the central carbazole bridge works as third dot for the NULL state. This molecule has also an alkyl chain (a binding element), that in particular allows to attach the molecule to the thiol which is needed to put the molecule on a gold surface [27]. The complete structure is shown in Fig. 2.10.



Figure 2.10: Bis-ferrocene molecule: dots definition (a) and equivalent 3-dot scheme (b).

The distance between the two working dot, dot1 and dot2, is 1.0 nm and, since the single molecule represents half a QCA cell, the ideal complete cell could be implemented placing two bis-ferrocenes at the distance of 1.0 nm.

Chapter 3

Methodology

The aim of this chapter is to understand how to study candidate molecules (diallyl butane, decatriene and bis-ferrocene) as QCA devices. In particular, we focused on the localization of the charges (electrons) in the molecules, which encode the binary information. To do this, simulations had to be made on computational chemistry software. In detail, the Gamess (US) [32], [33] software (the manual of which is in chapter 4) has been used instead of the most well known and most widely used Gaussian09 [34] which was used in [35].

The simulations of the candidate molecules were divided into two main steps:

- analysis of the molecules capability to encode binary information at equilibrium;
- simulation of the molecules with biasing conditions, in particular:
 - an external electric field [28];
 - a write-in system made by point charges [29].
- post-processing using Matlab and a C-program.

3.1 The steps

As said before, the first step of our analysis is to simulate a neutral molecule (total charge equal to 0) in equilibrium conditions. That means without any external field and far from other molecules (no interaction among them), as in Fig. 3.1.



Figure 3.1: A 3 dot molecule scheme.

The first simulation that must be done is of the optimization type, this means that the software calculates the position of all the atoms of the molecule iteratively until it finds the geometry which minimizes the energy of the system. Once that the simulation is over, in the output file will be the new geometry of the molecule and it will be used as input for the following simulations. In particular, the molecular system for this first simulation is described by using the Z-matrix (see Chapter 4 for the description of input file). After this first simulation, some simulations with an external electric field and with two point charges have to be performed, paragraph 3.1.1 and 3.1.2. Then the output files are analyzed by using a C-program written to evaluate the dot charges, with Matlab to draw the charges characteristic (function of external electric field or polarization of the driver) and, last but not least with Avogadro to understand the behaviour of the molecules by seeing HOMO and LUMO. The workflow of this work is shown in Fig. 3.2.



Figure 3.2: Workflow

3.1.1 External field

As explained in paragraph 3.1, after the optimization of the geometry of the molecule, a simulation with a finite external electric field has to be performed. This external electric field varies from -5 V/nm to 5 V/nm to understand how the free charges inside the dots switch between them according to the electric field.

There are two types of external electric field: Switching Field [36] and Clock Field [21]. The switching field is parallel to the dot-axis. The idea is to put the molecule between two metal electrodes (yellow box in Fig. 3.3) and to apply a voltage difference between them $(V = V^+ - V^-)$ [29]. The generated electric field has a direction

parallel to the working dot axis and lets the charges switch from one working dot to another one.



Figure 3.3: Switching field application system.



Figure 3.4: Switching field application system for a MQCA wire.

For instance, the charges in the molecule of the Fig. 3.3 switch to Dot 1 because of the direction of the switching field (from Dot 1 to Dot 2). Changing the sign of the switching field, the charge localization will be mostly in Dot 2. Basically, we can change the binary information encoded in the molecule by changing the sign of the switching field. In Fig. 6.4 is shown a simple example of this phenomenon for a diallyl butane molecule.



Figure 3.5: Displacement of the charge with an external electric field of: (a) -5 V/nm; (b) 5 V/nm.

The clock field is a reference signal and it is perpendicular to the switching field, used to raise o lower the tunneling barriers during QCA computations. Schematically, two metal electrodes are placed on the top and on the bottom of the molecule, and as for the switching field a voltage difference is applied, see Fig. 3.6.



Figure 3.6: Clock field application system.

If the clock signal is negative, we are in the release and relax phases, this means that if we are considering a 3 dot molecule the charge would be into the lower central Dot3 keeping the molecule in NULL state, encoding no information. While if the clock signal is positive, we are in the switch and hold phases, this means that the charge is moving in the Dot 1 or 2 depending on the orientation of the switching field, encoding "0" or "1" logic.

3.1.2 Point charges

Authors in [29] demonstrated that setting two ideal point charges (atomic number and mass nuclear equal to 0) at distance d (the same distance from the two active dots) it is possible to polarize the molecule forcing the free charge to move in one of two dots, as in Fig. 3.7.



Figure 3.7: Driver-molecule interaction.

The driver is represented by two point charges with a value of q and 1 - q, in this way the total charge is 1.

These point charges are named *Polarized Charge Driver* and are used as input for the system in a QCA wire, Fig. 3.8.



Figure 3.8: Point charges used as polarized driver for a MQCA wire.

Also in this case the localization of the charges inside the molecule is evaluated to understand the effect of the write-in system on the molecule.

Point charges can be useful, in addition to simulate the write-in system, also to simulate the behavior of the molecule in presence of a nearby molecule. In this case, the number of point charges must be equal to the number of dots of the molecule which has to be studied. This method is mainly used to simulate a QCA wire.

3.2 Figures of merit

Since we are analyzing the molecule from an electronic point of view, we have to understand how to study the charge localization inside the molecule. This can be done in two ways: one is a "chemical approach" and is the study of HOMO and LUMO orbitals; the second method is to define a new figures of merit, defined in [35], named *aggregated charge*, which is simply the sum of the atomic charge of each atom that forms a single dot. The choice of this new parameter means that the atomic charge is not a physical quantity, but a theoretical approximation. Even though the aggregated charge is strictly related to the atomic charge, it could represent the charge distribution inside a molecule from a macroscopic point of view and could be a readable quantity from an application perspective.




Figure 3.9: Definition of the aggregated charge for a bis-ferrocene molecule.

3.2.1 HOMO and LUMO

As we said in the previous paragraph, it is possible to see the charge configuration inside a molecule through the molecular orbital (MO). In particular, we have to see HOMO, HOMO-1, LUMO, LUMO+1. HOMO stands for Highest Occupied Molecular Orbital, so HOMO-1 is the second highest occupied molecular orbitals. Instead LUMO stands for Lowest Unoccupied Molecular Orbital. In particular, the HOMO-LUMO gap is defined as the energy difference between the HOMO's e LUMO's energy.

By definition, an orbital is the mathematical description of a region around a nucleus in an atom or molecule in which is more likely to find an electron. There are a maximum of 2 electrons in each orbital with opposite spin and the first occupied orbital is the one with the lowest energy (LUMO) up to the HOMO.

In the case studied in this work thesis, when a molecule is oxidized (which means that an electron is missing), this electron is removed by the HOMO.

3.3 Post-processing C-program

A C program has been written (see Appendix for the code) to automate the evaluation of the dots charge. In fact, at the end of simulation an output file is created, and this output file has more than 10k row. The C-program, called "*Mulliken charge*" ask for first which is the molecule to analyze, then ask to write the name of the .log file, see Fig. 3.10

```
Which molecule are you analyzing?

1. Diallyl Butane 2. Decatriene 3. Bis-ferrocene no-thiol 4. Bis-ferrocene thiol 5. Exit

3

Write the filename:

rs3_6_31_g-dp_ox_0_2_03nm.log

(a)

Mulliken charges found @ row 266100

Fc6: 0.888839

Cbz: 0.171641

Fc34: -0.117970

Me6: 0.08920

Me34: -0.032327

DOT1 = 0.978659

DOT2 = -0.150297

DOT3 = 0.171641

Terminated

(b)
```

Figure 3.10: Output of the c-program: selecting molecule and the writename (a) evaluation of the mulliken charge (b).

So it finds the mulliken charges in the .log file and perform some sums to evaluate the dots mulliken charges. This the "*one-shot*" version, which means that, after the first evaluation of the dots mulliken charge, it terminates. If you have more than one .log file to analyze you can use the *loop* version, which performs the same operation in a loop until an exit command is selected.

Finally the mulliken charge of every atoms is written in a .txt file called "*mulliken* charge atoms.txt" which can be useful for a successive visualization of the charge using Matlab.

Chapter 4

Tools Manual For Molecular QCA simulation

4.1 Ab-initio simulation

If we consider a polyatomic molecule, the electronic wave function depends on more than one parameter (bond distances, bond angles, dihedral angles of rotation about single bonds). In particular, for a molecular system, it's used to solve the Schroedinger equation defined as:

$$i\hbar\frac{\partial}{\partial t}\Psi\left(\overrightarrow{r},t\right) = \hat{H}\left(\overrightarrow{r},t\right) \tag{4.1}$$

where $\Psi(\vec{r}, t)$ is the wave function that depends on the position \vec{r} and time t; h is the reduced Planck constant; \hat{H} is the Hamiltonian defined as the sum of kinetic and potential energy operators:

$$\hat{H} = T + V \tag{4.2}$$

$$T = \frac{p^2}{2m} = -\frac{1}{2m}\nabla^2$$
 (4.3)

$$\rightarrow \hat{H} = -\frac{\nabla^2}{2m} + V\left(\overrightarrow{r}\right) \tag{4.4}$$

The chemical and physical properties of a generic molecular system are described by the solutions of the equation above. These properties are, for example the optimized physical geometrical molecular structure with the minimum total energy, the interaction energies, the electronic charge distributions. The Schrödinger equation can be solved exactly for only a few particular cases, for example for the hydrogen atom. For other cases, we need some approximations. Computational chemistry provides this and we have three classes of approximation methods [37]:

- Semi-empirical method: this method uses a simpler Hamiltonian than the correct molecular Hamiltonian and use parameters whose values are adjusted to fit experimental data or the results of ab initio calculations;
- Ab-initio method: it uses the correct Hamiltonian and does not use experimental data other than the values of the fundamental physical constants.
- Density Functional Theory (DFT) method: they are similar to ab-initio methods, but they include the effects of electron correlation, which is the fact that electrons in a molecular system react to one another motion.

Each type of method is characterized by the combination of theoretical procedure (called *method*), a *basis set* and a description of the molecule which can be cartesian or by a *Z*-matrix. A basis set is a mathematical representation of the molecular orbitals within a molecule. The basis set can be interpreted as restricting each electron to a particular region of space. Larger basis sets impose more accurately approximated molecular orbitals and they require accordingly more computational resources, as more accurate methods become more computationally expensive.

4.1.1 Z-matrix

The most common way used to describe a molecular system in every chemistry computational tool is by using the Z-matrix. The Z-matrix specifies the position of atoms in a molecule relative to each other in terms of atomic types, bond lengths and bond angles and dihedral angles.

Let's consider the molecule in Fig. 4.1 It consists of three atoms, one of oxygen and two of hydrogen. If we represent the water molecule in a three-dimensional space, each atom has three coordinates P(x, y, z).



Figure 4.1: Water molecule.

The construction of the Z-matrix has to follow these simple rules:

- 1. choose the starting atom and place it conceptually at the origin of a threedimensional space;
- 2. choose another atom linked to the first and position it along the z axis, specifying the label of the atom to which it is bound and the length of the link that connects them;
- 3. choose a third atom linked to one of the previous atoms and specify the binding angle formed by the two bonds;
- 4. Define the positions of the following atoms by specifying:
 - The atom label;
 - An atom to which it is bound and the bond length;
 - A third atom to which it is linked: label and value of the resulting binding angle;
 - A fourth atom and the value of the formed dihedral angle with the previous atoms.

It is possible to have two Z-matrix for each molecular system: one with the constant and another one with the variable. In particular, for the example in Fig. 4.1 we can have these two Z-matrix:

```
O
H 1 0.96
H 1 0.96 2 109.47
O
H 1 oh
H 1 oh 2 hoh
oh=0.96
hoh=109.47
```

In this work thesis, the input file for the optimization step has been written by using the Z-matrix, while all the others simulations by using the cartesian coordinates.

An example of an input file for a dially butane molecule is shown here:

```
$BASIS GBASIS=STO NGAUSS=3 $END
$CONTRL SCFTYP=RHF RUNTYP=FFIELD COORD=ZMT $END
$STATPT OPTTOL=0.0001 NSTEP=20 $END
$EFIELD EVEC(1)=0.002,0.001,0.001 $END
$DATA
Title
C1
\mathbf{C}
\mathbf{C}
            1
                       B1
\mathbf{C}
                       B2
            1
                              2
                                          A1
```

D1 D2 D3

C H H H H H H H H C C C C H H H H H H H	2 3 4 4 1 1 2 4 13 14 15 15 3 20 20 21 21 22 22 22	$\begin{array}{c} {\rm B3} \\ {\rm B4} \\ {\rm B5} \\ {\rm B6} \\ {\rm B7} \\ {\rm B8} \\ {\rm B9} \\ {\rm B10} \\ {\rm B11} \\ {\rm B12} \\ {\rm B13} \\ {\rm B14} \\ {\rm B15} \\ {\rm B16} \\ {\rm B17} \\ {\rm B18} \\ {\rm B19} \\ {\rm B20} \\ {\rm B21} \\ {\rm B22} \\ {\rm B23} \\ {\rm B24} \\ {\rm B25} \end{array}$	$1 \\ 1 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 1 \\ 1 \\ 2 \\ 4 \\ 4 \\ 13 \\ 13 \\ 13 \\ 13 \\ 13 \\ 13 \\$	A2 A3 A4 A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24	$ \begin{array}{c} 3 \\ 2 \\ 2 \\ 1 \\ 1 \\ 4 \\ 4 \\ 3 \\ 3 \\ 1 \\ 2 \\ 4 \\ 4 \\ 4 \\ 4 \\ 2 \\ 1 \\ 1 \\ 3 \\ $	$\begin{array}{c} D1 \\ D2 \\ D3 \\ D4 \\ D5 \\ D6 \\ D7 \\ D8 \\ D9 \\ D10 \\ D11 \\ D12 \\ D13 \\ D14 \\ D15 \\ D16 \\ D17 \\ D18 \\ D19 \\ D20 \\ D21 \\ D22 \\ D23 \\ \end{array}$
B1 B2 B3 B4 B5 B6 B7 B8 B9 B10 B11 B12 B13 B14 B15 B16 B17 B18 B19 B20 B21 B22 B23 B24 B25 A1 A2 A3 A4 A5 A6 A7 A8 A9 A10 A11 A12 A13 A14		1.5421 1.5471 1.5471 1.0877 1.0877 1.0877 1.0884 1.0884 1.0884 1.0884 1.0884 1.0884 1.0884 1.0884 1.0884 1.3777 1.3777 1.3777 1.0800 1.0803 1.5358 1.3777 1.3777 1.0800 1.0803 1.0804 1.0803 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0803 1.0804 1.0804 1.0803 1.0804 1.0804 1.0803 1.0804 1.08	$\begin{array}{r} 9718\\ 5028\\ 5028\\ 5028\\ 2095\\ 1964\\ 2095\\ 1964\\ 3697\\ 3680\\ 3697\\ 3680\\ 5525\\ 0630\\ 2504\\ 1478\\ 8854\\ 1487\\ 8854\\ 1487\\ 8854\\ 1487\\ 8854\\ 1487\\ 8854\\ 1487\\ 8859\\ 01425\\ 8930\\ 0141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 22967\\ 30141\\ 30267\\ 30267\\ 30141\\ 30141\\ 30267\\ 30141\\ 301$			

A15	121.41761015
A16	121.66666901
A17	121.41779191
A18	111.24431029
A19	118.62285258
A20	118.61999817
A21	121.66720440
A22	121.41761015
A23	121.66666901
A24	121.41779191
D1	180.00000000
D2	-58.69919355
D3	58.71180363
D4	-58.69919355
D5	58.71180363
D6	-58.61861781
D7	58.61814219
D8	-58.61861781
D9	58.61814219
D10	-179.99339158
D11	-88.92015591
D12	88.87856924
D13	-2.31385877
D14	178.35611866
D15	2.31545807
D16	-178.35545484
D17	-179.99339158
D18	-88.92015591
D19	88.87856924
D20	-2.31385877
D21	178.35611866
D22	2.31545807
D23	-178.35545484

\$END

4.2 Gamess Manual

4.2.1 Introduction to Gamess

GAMESS [32] stands for General Atomic and Molecular Electronic Structure System (GAMESS (US)) and it is a computer software for computational chemistry. It is useful to solve the 4.1 by using RHF (restricted Hartee Fock), UHF (unrestricted Hartee Fock) and DFT (density functional theory). In this work the UHF has been used in all the simulations. The difference between UHF and DFT is that the latter considers the correlation between electron and electron. But it is much more expensive in terms of computational time. Because of this reason, we preferred to use the UHF, even though the results are less precise.

4.2.2 Download

Go to http://www.msg.ameslab.gov/gamess/License_Agreement.html and click on "I agree to the above terms", you will be addressed to a new website page where you can download the version of GAMESS that you need. GAMESS is available for Windows, OSX and Linux. An e-mail should arrive within one hour to one day with the link for downloading GAMESS and an username and password. If not, try again.

4.2.3 Run Gamess

• MacOS:

To run Gamess an input file is needed which contains all the information that Gamess needs. An input file is a file with ".inp" extension, for example "water.inp". You have to copy the input file in the same folder where Gamess is located and then using the shell you have to type **cd/path** where "path" is the folder containing GAMESS. Then you can run **./gms water.inp** (see Figure 4.2).



Figure 4.2: Running Gamess.

When the simulation is over, a file named "water.log" is created and it is the output file which cointains all the results from the simulation.

• Linux:

To run Gamess in Linux the steps are the same for MacOs, the only difference is that when you are in the gamess folder you have to run **rungms** by typing: "/software/gamess/gamess20170420r1/gamess/rungms xxx.inp > xxx.log" where xxx is the name of the input file.

• Windows:

• Through the server:

Before running Gamess we have to create in the folder /gamesstmp a subfolder named as the name of the user; then inside it, we need to create another folder named "scratch" where all the temporary files will be written. When a simulation is over, these files will be deleted automatically . You need again a folder named "scratch" in your own home which will contain all the .dat files created during the simulations. If a simulation terminates abnormally you have to delete the .dat file in this folder before starting a new simulation. Running Gamess through the server works exactly like on Linux, with the difference that on the server we have the possibility to run it in parallel. To do that we have to go in the folder where the .inp file is and type in the shell: "/software/gamess/gamess20170420r1/gamess/rungms xxx.inp YY Z > xxx.log" where xxx is the name of the input file, YY is the Gamess version (in our case "00") and Z is the number of processors we need. For example, there are 4 processors in the Micro&Nano server, so if we want to run Gamess in these 4 processors in parallel we have to type: /software/gamess/gamess20170420r1/gamess/rungms xxx.inp 00 4 > xxx.log.

A shell script has been written so that all files with extension .inp in the folder are identified and all sequentially simulated.

4.2.4 Input file

GAMESS [32] input takes the form of a list of groups. Each group controls some aspect of the calculation, from the choice of basis set to the kind of calculation to the format for structure input.

- 1. Each group has a name that begins with a \$ sign, which must be in column 2 of the input line (you have to put a blank space at the start of the line);
- 2. The group must be terminated with \$END;
- 3. The options within a group may be placed all on one line, or distributed over multiple lines. GAMESS recognizes the end of an input group by the \$END;
- 4. Comment lines begin with a ! in column 1.
- 5. Each group consists of one or more keywords, depending on your choice.

Every group description are described accurately in [38].

• **\$CONTRL** Group [38]

The control group handles the type of calculation, the type of SCF, the type of coordinates of the molecule and much more. A default GAMESS control group is RHF/UHF SCF, singlet state, a single point energy calculation and Cartesian coordinates in units of Angstroms. If these defaults are ok, then the \$CONTRL group does not need to be provided.

Below are some common options used in this work, specified by listing them following the \$CONTRL:

- RUNTYP is the type of computation. In this work we used OPTIMIZE (optimize geometry using analytic energy gradients), FFIELD (applies finite electric fields) and ENERGY (single point energy);
- SCF calculation: SCFTYP=RHF for Restricted Hartree Fock calculation and UHF for Unrestricted Hartree Fock calculation.
- DFTTYP calculate the DFT (density functional theory), in particular B3LYP;
- Coordinates of the molecules: cartesians (COORD=CART) or by using the Z-matrix (COORD=ZMT), as seen in 4.1.1;
- MAXIT is the maximum number of SCF iteration cycles. This pertains only to RHF, UHF, ROHF runs (default = 30, maximum 200);
- ICHARG and MULT define the charge and the multiplicity of a molecule that can be neutral, oxidized (ICHARG=1 MULT=2) o reduced (ICHARG=-1 MULT=2).

• **\$BASIS** Group [38]

This group allows certain standard basis sets to be easily requested. Use GBA-SIS= to set the general type, and NGAUSS= to set the number of Gaussians. Here is a short list of basis presents in Gamess and used:

- STO-3G: \$BASIS GBASIS=STO NGAUSS=3 \$END;
- 6-31G: \$BASIS GBASIS=N31 NGAUSS=6 \$END;
- 6-31G(d): \$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 \$END;
- 6-31G(d,p): \$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 \$END;

Sometimes it may be necessary to use some basis that are not present in Gamess by default. For example, in the bis-ferrocene simulations in gaussian09 a basis (LANL2DZ) was used which is not present in Gamess. The procedure to use an external basis set consists in changing the \$BASIS control to read an external file, and to modify rungms to read the external basis set file. Let's start with how to modify the rungms script. You have to modify the line "setenv EXTBAS" as follows:

```
set echo
setenv ERICFMT ./ericfmt.dat
setenv IRCDATA ./$JOB.irc
setenv INPUT $SCR/$JOB.F05
setenv PUNCH ./$JOB.dat
setenv EXTBAS ./XXX.txt
setenv AOINTS $SCR/$JOB.F08
```

where XXX.txt is a random name of the basis. In this case the .txt file must be in the same folder as the rungms script, while if your file is in another folder you have to specify the path. For example, if the basis file is in the desktop folder you have to write:

setenv EXTBAS / Users / .. / Desktop / gamess /XXX.txt

Regarding the input file, the only noteworthy difference with the input file seen before is the \$BASIS line: EXTFIL=.TRUE. tells GAMESS-US to use the external file and GBASIS=XXX to use external basis sets named XXX.

The external basis sets can be downloaded from https://bse.pnl.gov/bse/portal

An example of an external basis is shown here (LANL2DZ basis):

Н	L	ANL2DZ	
1	1	$\frac{2}{1.309756377}$	0.4301284980
	2	0.233135974	0.6789135310
С	L	ANI 2DZ	
s	ш	7	
~	1	4233.0000000	0.0012200
	2	634.9000000	0.0093420
	3	146.1000000	0.0454520
	4	42.5000000	0.1546570
	5	14.1900000	0.3588660
	6	5.1480000	0.4386320
	7	1.9670000	0.1459180
\mathbf{S}		2	
	1	5.1480000	-0.1683670
~	2	0.4962000	1.0600910
S		1	1
Б	1	0.1533000	1.0000000
Ρ	4	4	0.0105200
	1	18.1600000	0.0185390
	2	3.9860000	0.1154360
	3	1.1430000	0.3801880
D	4	0.5594000	0.0401140
1	1	0.1146000	1.0000000
NT	т		
IN C	L	7	
5	1	5909 000000	0 0011900
	2	887 5000000	0.0090990
	3	204.7000000	0.0441450
	4	59.8400000	0.1504640
	5	20.0000000	0.3567410
	6	7.1930000	0.4465330
	7	2.6860000	0.1456030
\mathbf{S}		2	
	1	7.1930000	-0.1604050
	2	0.7000000	1.0582150
\mathbf{S}		1	
_	1	0.2133000	1.0000000
Ρ	_	4	
	1	26.7900000	0.0182540
	2	5.9560000	0.1164610
	3	1.7070000	0.3901780
	4	0.5314000	0.6371020

P I		
1	0.1654000	1.000000
-	0.1001000	1.0000000
D. LAN	1.0D7	
Fe LAN	L2DZ	
S 3		
1	6.4220000	-0.3927882
2	1.8260000	0.7712643
3	0.7135000	0.4920228
S = 4		
1	6.4220000	0.1786877
2	1 8260000	-0.4194032
2	0.7135000	0.4568185
5	0.1001000	1 1025049
4 1	0.1021000	1.1055048
5 1		
1	0.0363000	1.0000000
P 3		
1	19.4800000	-0.0470282
2	2.3890000	0.6248841
3	0.7795000	0.4722542
Р 1		
1	0.0740000	1 000000
	0.0740000	1.0000000
P 1	0.000000	1 0000000
1	0.0220000	1.0000000
D 4		
1	37.0800000	0.0329000
2	10.1000000	0.1787418
3	3.2200000	0.4487657
4	0.9628000	0.5876361
D 1	0.0020000	010010001
1	0 2262000	1 000000
1	0.2262000	1.0000000
1	0.2262000	1.0000000
1	0.2262000	1.0000000
1 Au LAN	0.2262000 L2DZ	1.0000000
1 Au LAN S 3	0.2262000 L2DZ	1.0000000
1 Au LAN S 3 1	0.2262000 L2DZ 2.8090000	1.0000000 -1.2021556
1 Au LAN S 3 1 2	0.2262000 L2DZ 2.8090000 1.5950000	1.0000000 -1.2021556 1.6741578
1 Au LAN S 3 1 2 3	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000	1.0000000 -1.2021556 1.6741578 0.3526593
$ \begin{array}{c} 1\\ \text{Au LAN}\\ \text{S} & 3\\ 1\\ 2\\ 3\\ \text{S} & 4 \end{array} $	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000	1.0000000 -1.2021556 1.6741578 0.3526593
1 Au LAN S 3 1 2 3 S 4 1	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481
1 Au LAN S 3 1 2 3 S 4 1 2	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 1.8642846
1 Au LAN S 3 1 2 3 S 4 1 2 2	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ 1.0256320\end{array}$
1 Au LAN S 3 1 2 3 S 4 1 2 3 4 1 2 3	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.20242920\end{array}$
1 Au LAN S 3 1 2 3 S 4 1 2 3 4 2 3 4	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.5327000 0.2826000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\end{array}$
$\begin{array}{c}1\\ \text{Au LAN}\\ \text{S} & 3\\ 1\\ 2\\ 3\\ \text{S} & 4\\ 1\\ 2\\ 3\\ 4\\ \text{S} & 1\end{array}$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\end{array}$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.5327000 0.2826000 0.0598000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\\ 1.0000000\end{array}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\\ 1.0000000\\ \end{array}$
$\begin{array}{c} 1 \\ \text{Au LAN} \\ \text{S} & 3 \\ 1 \\ 2 \\ 3 \\ \text{S} & 4 \\ 1 \\ 2 \\ 3 \\ 4 \\ \text{S} & 1 \\ 1 \\ \text{P} & 3 \\ 1 \end{array}$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000	1.0000000 - 1.2021556 1.6741578 0.3526593 1.1608481 - 1.8642846 - 1.0356230 1.3064399 1.0000000 - 0.2802681
$\begin{array}{c} 1 \\ \text{Au LAN} \\ \text{S} & 3 \\ 1 \\ 2 \\ 3 \\ \text{S} & 4 \\ 1 \\ 2 \\ 3 \\ 4 \\ \text{S} & 1 \\ 1 \\ \text{P} & 3 \\ 1 \\ 2 \end{array}$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398
$\begin{array}{c} 1 \\ \text{Au LAN} \\ \text{S} & 3 \\ 1 \\ 2 \\ 3 \\ \text{S} & 4 \\ 1 \\ 2 \\ 3 \\ 4 \\ \text{S} & 1 \\ 1 \\ P & 3 \\ 1 \\ 2 \\ 3 \end{array}$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\\ 1.0000000\\ -0.2802681\\ 0.7818398\\ 0.4804776\end{array}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.5327000 0.5327000 0.5327000 0.5327000 0.598000 3.6840000 1.6660000 0.5989000 0.6828000	1.0000000 - 1.2021556 1.6741578 0.3526593 1.1608481 - 1.8642846 - 1.0356230 1.3064399 1.0000000 - 0.2802681 0.7818398 0.4804776
$\begin{array}{c} 1 \\ \text{Au LAN} \\ \text{S} & 3 \\ 1 \\ 2 \\ 3 \\ \text{S} & 4 \\ 1 \\ 2 \\ 3 \\ 4 \\ \text{S} & 1 \\ 1 \\ \text{P} & 3 \\ 1 \\ 2 \\ 3 \\ \text{P} & 2 \\ 1 \\ \end{array}$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.6838000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078
$\begin{array}{c} 1 \\ \text{Au LAN} \\ \text{S} & 3 \\ 1 \\ 2 \\ 3 \\ \text{S} & 4 \\ 1 \\ 2 \\ 3 \\ 4 \\ \text{S} & 1 \\ 1 \\ \text{P} & 3 \\ 1 \\ 2 \\ 3 \\ \text{P} & 2 \\ 1 \\ 2 \\ 1 \\ 2 \\ 2 \\ 1 \\ 2 \\ 2 \\ 2 \\$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\\ 1.0000000\\ -0.2802681\\ 0.7818398\\ 0.4804776\\ -0.0952078\\ 1.0299147\end{array}$
$\begin{array}{cccccccc} 1 \\ Au & LAN \\ S & 3 \\ 1 \\ 2 \\ 3 \\ S & 4 \\ 1 \\ 2 \\ 3 \\ 4 \\ S & 1 \\ 1 \\ P & 3 \\ 1 \\ 2 \\ 3 \\ P & 2 \\ 1 \\ 2 \\ P & 1 \end{array}$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000	$\begin{array}{c} 1.0000000\\ -1.2021556\\ 1.6741578\\ 0.3526593\\ 1.1608481\\ -1.8642846\\ -1.0356230\\ 1.3064399\\ 1.0000000\\ -0.2802681\\ 0.7818398\\ 0.4804776\\ -0.0952078\\ 1.0299147\end{array}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000 0.0279000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078 1.0299147 1.0000000
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000 0.0279000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078 1.0299147 1.0000000
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000 0.0279000 1.2870000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078 1.0299147 1.0000000 0.5844273
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000 0.0279000 1.2870000 0.4335000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078 1.0299147 1.0000000 0.5844273 0.5298161
$\begin{array}{c} 1 \\ \text{Au LAN} \\ \text{S} & 3 \\ 1 \\ 2 \\ 3 \\ \text{S} & 4 \\ 1 \\ 2 \\ 3 \\ \text{S} & 1 \\ 1 \\ \text{P} & 3 \\ 1 \\ 2 \\ 3 \\ \text{P} & 2 \\ 1 \\ 2 \\ 1 \\ \text{D} & 2 \\ 1 \\ 2 \\ 2 \\ \text{P} & 1 \\ 1 \\ \text{D} & 2 \\ 2 \\ 1 \\ 2 \\ 1 \\ 1 \\ \text{D} & 2 \\ 2 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 2 \\ 2$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000 0.0279000 1.2870000 0.4335000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078 1.0299147 1.0000000 0.5844273 0.5298161
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.2262000 L2DZ 2.8090000 1.5950000 0.5327000 2.8090000 1.5950000 0.5327000 0.2826000 0.0598000 3.6840000 1.6660000 0.5989000 0.6838000 0.0977000 0.0279000 1.2870000 0.4335000 0.1306000	1.0000000 -1.2021556 1.6741578 0.3526593 1.1608481 -1.8642846 -1.0356230 1.3064399 1.0000000 -0.2802681 0.7818398 0.4804776 -0.0952078 1.0299147 1.0000000 0.5844273 0.5298161

• **\$STATPT** Group [38]

This group is used for RUNTYP = OPTIMIZE. In particular it controls the search for optimized points. The parameter OPTTOL is the gradient convergence tolerance, in Hartree/Bohr. The simulation converges when the convergence of a geometry search requires the largest component of the gradient to be less than OPTTOL, and the root mean square gradient less than 1/3 of OPTTOL (default=0.0001). NSTEP is the maximum number of steps to take. The default is 50 steps for a minimum search, but only 20 for a transition state search, which benefit from relatively frequent Hessian re-evaluations.

• **\$EFIELD** Group [38]

This group introduces an external electric field on the system. It is composed by two parameters, but we used just one:

- EVEC = an array of the three (x, y, z) components of the applied electric field, in a.u;

• **\$SYSTEM** Group [38]

This group provides global control information for your computer's operation. If you don't specify this group, Gamess uses 1000000 words of memory by default. 1 word is 8 byte, so 8 MB of RAM \$\$YSTEM MWORDS=1 \$END

MWORDS is the maximum replicated memory which your job can use, on every node and it can only be integers (1, 2, 3, ..). This is given in units of 1000000 words (as opposed to $1024 \cdot 1024$ words). There are some simulations which need more than 8 MB of RAM, in this case there will be an error in the output file (***** ERROR: MEMORY REQUEST EXCEEDS AVAILABLE MEMORY). The memory used by Gamess can be improved with the command \$SYSTEM. Here is the simplest way to deal with memory: if your current laptop computer has 4 GB of RAM, and you use it for other things while GAMESS is running, so you can give GAMESS a maximum of roughly 2 or 3 GB of RAM. This translates to:

 $2\,\mathrm{GB} = 2.000\,\mathrm{MB} = 250\,MWORDS$

In this case, you have to add the following command in all your input files: \$\$YSTEM MWORDS=250 \$END

Here there is a simple example of a first part of an input file:

^{\$}BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 \$END \$CONTRL SCFTYP=RHF RUNTYP=OPTIMIZE COORD=ZMT \$END \$STATPT OPTTOL=0.0001 NSTEP=20 \$END \$SYSTEM MWORDS=50 \$END

• **\$ELPOT** Group [38]

This group controls electrostatic potential calculation, if IEPOT=0 Gamess skips this property, while if IEPOT=1 calculates electric potential. It is possible calculate the electric potential in several points by the keyword WHERE which can be equal to:

- COMASS: center of mass;
- NUCLEI: at each nucleus (default);
- POINTS: at points given in \$POINTS;
- GRID : at grid given in \$GRID;

This first card in the \$POINT group must contain the string ANGS or BOHR, followed by an integer NPOINT, the number of points to be used. The next NPOINT cards are read in free format, containing the X, Y, and Z coordinates of each desired point.

The \$GRID group is used to input a plane or cube on which properties will be calculated (in our case the electric potential). It is composed of:

- MODGRD = 0 orthonormalize the grid vectors or 1 normalize the grid vectors;
- ORIGIN(i) = coordinates of one corner of the grid/cube;
- XVEC(i) = vector from ORIGIN to an adjacent corner "X" of the grid (or cube); the XVEC direction need not be parallel to the X-axis of the molecule.
- YVEC(i) = vector to the adjacent corner "Y" of grid/cube;
- ZVEC(i) = vector to the adjacent corner "Z" of the cube, given if and only if MODGRD=1;
- SIZE = grid increment in all directions (default 0.25);
- UNITS = units of the above five values, it can be either ANGS (the default) or BOHR.

In this way a cube is created in a three-dimensional space, and the electrostatic potential is calculated at each point where two lines intersect. This method is used since the value of the electrostatic potential around the molecule is needed in the algorithm used in [39] for the calculation of the ESP charges.

• \$GUESS and VEC Group [38]

You may have to use the results of a simulation as the starting point for a second simulation. For example, when the SCF does not converge after 200 iterations, you can start from the two hundredth iteration to get 400 iterations

and so on. The second example is that, for particularly complex molecules (for example bis-ferrocene), it is necessary to give as input the MOs calculated in a previous optimization simulation because otherwise the calculation of the SCF oscillates. At the end of an optimization simulation, GAMESS creates a .dat file with a section called \$VEC, where all the calculated MOs are described by a list of vectors. To use these vectors in the input file you must write the command:

\$GUESS GUESS=MOREAD NORB=154 \$END

where MOREAD tells GAMESS that the molecular orbitals must read and NORB is the number of molecular orbitals to read. This number can be found in the .log file:

NUMBER OF OCCUPIED ORBITALS	(ALPHA) =	154
NUMBER OF OCCUPIED ORBITALS	(BETA) =	153
TOTAL NUMBER OF ATOMS		72

At the end of the file the \$VEC group must be written with all the orbital vectors (more than 50000 lines). If geometry optimization was calculated using SCFTYP = UHF, you have two different values for alpha and beta orbitals. In the case in the example, you have to copy the 154th alpha orbitals after the 153th beta orbitals and set NORB=154.

• Effective fragment potential (EFP) [38]

The only way to include two or more points charge in the simulations found up to now is to use EFP (Effective Fragment Potential). The basic idea behind the EFP method is to replace the chemically inert part of a system by EFPs (points charge), while performing a regular ab-initio calculation on the chemically active part (the molecule). Here "inert" means that no covalent bond breaking process occurs. This "spectator region" consists of one or more "fragments", which interact with the ab initio "active region" through nonbonded interactions, and so of course these EFP interactions affect the ab initio wavefunction. Let's consider the following molecular system:



Dot 1

Figure 4.3: Diallyl butane with 2 points charge.

To describe this molecular system we have to write the following input file:

```
$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 $END
 $CONTRL SCFTYP=RHF RUNTYP=ENERGY COORD=CART UNITS=BOHR
 MAXIT=200 $END
 $SCF CONV=1d-5 $END
 $STATPT OPTTOL=0.0001 NSTEP=20 $END
 $SYSTEM MWORDS=50 $END
$DATA
Title
C1
С
                            -0.0002526451
                                             -0.5880556204
     6.0
            1.3332368711
\mathbf{C}
     6.0
           -1.3332368711
                            -0.0002526451
                                              0.5880556204
\mathbf{C}
     6.0
            3.4358660887
                            -0.0002526451
                                              1.4434251503
С
           -3.4358660887
                            -0.0002526451
                                             -1.4434251503
     6.0
Η
     1.0
            3.2227946107
                             1.6578397290
                                              2.6394134677
            3.2230477325
                            -1.6585651191
                                              2.6391490900
Η
     1.0
                             1.6578397290
Η
     1.0
           -3.2227946107
                                             -2.6394134677
Н
           -3.2230477325
                            -1.6585651191
                                             -2.6391490900
     1.0
            1.5533761233
                            -1.6556908607
                                             -1.7887575775
Η
     1.0
Η
     1.0
            1.5533514007
                             1.6551827736
                                             -1.7887654158
Η
     1.0
           -1.5533761233
                            -1.6556908607
                                              1.7887575775
                             1.6551827736
Η
           -1.5533514007
                                              1.7887654158
     1.0
\mathbf{C}
     6.0
           -6.0717882763
                             0.0000593591
                                             -0.2287136680
С
     6.0
           -7.1866398826
                            -2.2847192418
                                              0.3325699278
С
                             2.2851711012
     6.0
           -7.1854907850
                                              0.3336581671
Η
           -6.2846283303
                            -4.0578002243
                                             -0.1234590242
     1.0
Η
     1.0
           -9.0163919699
                            -2.3834287213
                                              1.2328758142
Η
     1.0
           -6.2825836824
                             4.0579984242
                                             -0.1215847979
                             2.3843934114
Η
     1.0
           -9.0151818102
                                              1.2340302771
\mathbf{C}
            6.0717882763
                             0.0000593591
                                              0.2287136680
     6.0
\mathbf{C}
     6.0
            7.1866398826
                            -2.2847192418
                                             -0.3325699278
С
     6.0
            7.1854907850
                             2.2851711012
                                             -0.3336581671
Н
                            -4.0578002243
     1.0
            6.2846283303
                                              0.1234590242
Η
     1.0
            9.0163919699
                            -2.3834287213
                                             -1.2328758142
Η
     1.0
            6.2825836824
                             4.0579984242
                                              0.1215847979
Η
     1.0
            9.0151818102
                             2.3843934114
                                             -1.2340302771
 $END
 $EFRAG
 POSITION=FIXED COORD=CART
```

fragname=WATER1

```
W1O1 -7.1861 -14.372 -0.333
W1H2 -3.2131 5.376 -0.12
W1H3 -3.2131 4.763 -0.23
fragname=CO21
C1C1 \quad 7.1861 \quad -14.372 \quad -0.333
C1O2
      -6.3131 5.376 -0.12
C1O3
      -6.2131 4.763 -0.23
 $END
$WATER1
WATER1 as DR1
COORDINATES(BOHR)
W101 -7.1861 -14.372 -0.333 0.0 0.0
W1H2
      -3.2131 5.376 -0.12 0.0 0.0
W1H3
      -3.2131 4.763 -0.23 0.0 0.0
STOP
MONOPOLES
W1O1 0.7 0.0
W1H2 0.0 0.0
W1H3 0.0 0.0
STOP
REPULSIVE POTENTIAL
W1O1
0, 0
STOP
$END
$CO21
Carbon as DR2
COORDINATES(BOHR)
C1C1 7.1861 -14.372 - 0.333 0.0 0.0
C1O2
      -6.3131 5.376 -0.12 0.0 0.0
C1O3 \quad -6.2131 \quad 4.763 \quad -0.23 \quad 0.0 \quad 0.0
STOP
MONOPOLES
C1C1 0.3 0.0
C1O2 \ 0.0 \ 0.0
C1O3 0.0 0.0
STOP
REPULSIVE POTENTIAL
C1C1
0, 0
STOP
$END
$FRGRPL
PAIR=WATER1 CO21
W1O1 C1C1 0 0
 STOP
 $END
```

After the coordinates of the molecule we have the EFRAG group so formed:

- line 1: POSITION=FIXED COORD=CART
 This means that the position of the fragments are fixed and that we are considering cartesian coordinates;
- line 2: fragname=XXX
 This the name of the fragment for a future use;
- line 3: Gamess needs at least three atoms to define each fragment, since we need just one atom for each fragment we set the coordinates for the first atom. The second and third atoms are dummy values;

- line 4: There is a group with the name of the fragment where we specify again the coordinates and the value of charges in MONOPOLES.
- line 5: REPULSIVE POTENTIAL signals the start of the subgroup containing the fitted exchange repulsion potential, for the interaction between the fragment and the ab initio part of the system;
- line 6: This group defines the inter-fragment repulsive potential for EFP1 potentials; it accounts primarily for exchange repulsions, but also includes charge transfer. Note that the functional form used for the fragment-fragment repulsion differs from that used for the ab initio-fragment repulsion, which is defined in the \$FRAGNAME input.

• Creating an input file by using Avogadro

Avogadro is an advanced molecule editor and visualizer designed for crossplatform use in computational chemistry, you can download it from this link: https://sourceforge.net/projects/avogadro/files/latest/download and some manuals are available here: https://avogadro.cc/docs/ and here https: //www.gitbook.com/book/ghutchis/avogadro/details. Select the pencil to begin drawing, like in Fig. 4.4:



Figure 4.4: Avogadro

Note that the red, green, and blue arrows represent the x, y, and z axes respectively. You can select the element you want from "Draw settings" on the left; if you work with organic molecules you have to select "satura con idrogeni".



Figure 4.5: Drawing a molecul with Avogadro.

After you have drawn your molecule you can optimize it. To do this, go to Estensione \rightarrow Ottimizza la geometria.



Figure 4.6: Optimize geometry.

You can also change the "ordine dei legami", as in Fig. 4.7.



Figure 4.7: Bond order.

Now you can generate the Gamess input file. Go to Estensioni \to Gamess \to Generatori di input.



Figure 4.8: Gamess input generator.

Then you can choose all the parameters for your simulation, for example basis

•		Input GAMESS	
1	💠 🖭 🔄 🖡 🗢 🖳	Impostazioni base Setup Avanz	
	Display Types Anello Assi Bastoni Dipolo Etichetta Forza Fumetto Legame idrogeno Nastro Poligono OTAIM Sfera e bastoncino Sfere di Van der Waals Superfici	Calcola Energia del punto singolo o Con: RHF e STO-3G e In: Gas e Molleplioltà: Singoletto o Carioa: Neutro o	
Ordinu Re	Ad Duplicate Impostazioni di deej Elemento: Carbonio (6) e dei legami: Dopplo agola gli idrogeni	File created by the GAMESS Input Deck Generator Plugin for Avogadro SBASIS GEASIS_STO MGAUSS-S SEND SCONTAL Title Store 01 0.708041 2.07038 0.00000 02 0.07070 309641 0.00000 03 6.0.380750 0.390641 0.00000 04 0.07068 2.0663 0.00000 11 0.75068 2.0663 0.00000 11 0.75076 2.6663 0.00000 11 0.75076 3.6663 0.00000 11 0.75076 3.6663 0.00000 11 0.75076 3.6663 0.00000 11 0.750776 3.68630 0.00000 11 0.750776 3.68630 0.00000 11 0.26010 2.68682 0.00000 11 0.26010 2.68682 0.00000 11 0.470757 3.8630 0.0000 11 0.26010 2.68682 0.00000	
		Messages	

sets, what you want to calculate and the representation of the molecule (Z-Matrix or cartesian axis).

Figure 4.9: Input generator settings (A).

•	•	Input GAMESS	
1	🔶 🕼 😓 💺 🔊 🖡	Impostazioni hasa Satun Avanz	
0.0	Display Types		
• •		Base	
	Anello	SCF	
	Assi	Dati Tipo di Coordinate 🗸 Coordinate Cartesiane uniche.	
	Bastoni	Sistema Interne di Hilderbrant Guess MO Coordinate Categiana	
	Dipolo	Varie Unità: Matrico 2	
	Etichetta	Gruppo Runtor	
	Fuiza		
	Legame idrogeno	# di variabili Z-Matrix 0 0	
	Nastro		
	Poligono	Ordine degli assi principali: 2 0	
	QTAIM	Sfrutta la simmetria durante il Calcolo	
<pre></pre>	Sfera e bastoncino		
	Sfere di Van der Waals		
	Superfici		
	Add Duplicate		
	Dapirouto	I File created by the GAMESS Input Deck Generator Plugin for Avogadro	
0 0	Impostazioni di diser	SCONTEL SCFTYP=RHF RUNYP=ENERGY SEND	
•••		\$DATA	
	Elemento: Carbonio (6)	Title G1	
	Dente	C 60 -7.06241 2.07038 0.00000 2.6025 0.00000	
Ordi	ine dei legami: Doppio	C 6.0 -3.80750 -0.30641 0.00000 C 6.0 -0.75088 2.90683 0.00000	
🛛 🖓 F	Regola gli idrogeni	H 1.0 -7.64064 1.85914 -0.87517 H 1.0 - 6.78449 3.10365 0.00305	
		H 10 -7.64394 1.85480 0.87212 H 10 - 3.44331 1.33480 0.00000	
		H 1.0 -1.07575 3.886530 0.000000	
		H 1.0 0.26010 2.8962 0.00000	
		Bipristina tutto Valori predefiniti Calcola Genera Chiudi	
		Messages	

Figure 4.10: Input generator settings (B).

Finally click on "Genera...".

4.2.5 Output file

When the simulation is over, an output file is created. This file has a .log extension and it contains all the results from the simulation. Starting from the input file for Dially Butane seen before, now the output file is reported here. Some portions of the output file are reported here.

The INPUT CARD section at the start shows the first few lines of the input file. Here it shows the entire input:

ECHO OF THE FIRST FEW INPUT CARDS -INPUT CARD> \$BASIS GBASIS=STO NGAUSS=3 \$END INPUT CARD> \$CONTRL SCFTYP=RHF RUNTYP=OPTIMIZE COORD=ZMT \$END INPUT CARD> \$STATPT OPTTOL=0.0001 NSTEP=20 \$END INPUT CARD> INPUT CARD> \$DATA INPUT CARD>Title INPUT CARD> C1 INPUT CARD> C INPUT CARD> C B11 INPUT CARD> C B22 A1 1 INPUT CARD> C D1 2 B3 1 A2 3 INPUT CARD> H 3 B41A32 D2INPUT CARD> H B5 $\mathbf{2}$ D33 1 A4INPUT CARD> H 4 B6 $\mathbf{2}$ A51 D4INPUT CARD> H $\mathbf{2}$ 4B7A61 D5INPUT CARD> H **B**8 2 Α7 1 4 D6INPUT CARD> H B9 $\mathbf{2}$ A84D71 INPUT CARD> H $\mathbf{2}$ B10 1 A93 D8INPUT CARD> H A10 2 B11 1 3 D9INPUT CARD> C $\mathbf{2}$ D10 4 B12 A111 INPUT CARD> C 4 $\mathbf{2}$ 13B13 A12 D11 INPUT CARD> C 13B144 A132 D12INPUT CARD> H B1513A144D1314INPUT CARD> H A154 14 B16 13D14 INPUT CARD> H 15B1713A16 4D15INPUT CARD> H B18 A17 D16 1513 4 INPUT CARD> C -3 B19 1 A18 2 D17 INPUT CARD> C 20B20 3 A191 D18 INPUT CARD> C B21 D19 203 A20 1 INPUT CARD> H 21B22 20A213 D20INPUT CARD> H 21B2320A223 D21INPUT CARD> H B24 20D22 22 A23 3 INPUT CARD> H 22B2520A243 D23INPUT CARD> INPUT CARD> B1 1.54219718INPUT CARD> 1.54715028B2INPUT CARD> B31.54715028INPUT CARD> B41.08772095INPUT CARD> B51.08771964INPUT CARD> B6 1.08772095INPUT CARD> B71.08771964INPUT CARD> 1.08843697 **B**8 INPUT CARD> B91.08843680 INPUT CARD> B101.08843697INPUT CARD> B11 1.08843680INPUT CARD> B12 1.53585525INPUT CARD> B131.37770630 INPUT CARD> B14 1.37772504INPUT CARD> B15 1.080014781000000 WORDS OF MEMORY AVAILABLE

BASIS OPTIONS

GBASIS=STO		IGAUSS=	3	POLAR=NONE	
NDFUNC=	0	NFFUNC=	0	DIFFSP=	\mathbf{F}
NPFUNC=	0	DIFFS=	\mathbf{F}	BASNAM	

Then there is a section called "Coordinates of all atoms (angs)" that lists the coordinate system used internally by the program:

	CC	OORDINATES OF A	LL ATOMS ARE (A	NGS)
ATOM	CHARGE	Х	Y	Z
С	6.0	0.7055186200	-0.0001336940	-0.3111856556
\mathbf{C}	6.0	-0.7055186200	-0.0001336940	0.3111856556
\mathbf{C}	6.0	1.8181821656	-0.0001336940	0.7638277505
\mathbf{C}	6.0	-1.8181821656	-0.0001336940	-0.7638277505
Н	1.0	1.7054295870	0.8772910675	1.3967175585
Н	1.0	1.7055635332	-0.8776749274	1.3965776558
Н	1.0	-1.7054295870	0.8772910675	-1.3967175585
Н	1.0	-1.7055635332	-0.8776749274	-1.3965776558
Н	1.0	0.8220113040	-0.8761539353	-0.9465698144
Н	1.0	0.8219982213	0.8758850671	-0.9465739622
Н	1.0	-0.8220113040	-0.8761539353	0.9465698144
Н	1.0	-0.8219982213	0.8758850671	0.9465739622
С	6.0	-3.2130522180	0.0000314115	-0.1210300697
С	6.0	-3.8030063244	-1.2090214437	0.1759884396
С	6.0	-3.8023982480	1.2092605574	0.1765643111
Н	1.0	-3.3256823323	-2.1472955607	-0.0653317068
Н	1.0	-4.7712695007	-1.2612562545	0.6524098320
Н	1.0	-3.3246003512	2.1474004435	-0.0643399089
Н	1.0	-4.7706291117	1.2617667465	0.6530207475
С	6.0	3.2130522180	0.0000314115	0.1210300697
С	6.0	3.8030063244	-1.2090214437	-0.1759884396
С	6.0	3.8023982480	1.2092605574	-0.1765643111
Н	1.0	3.3256823323	-2.1472955607	0.0653317068
Н	1.0	4.7712695007	-1.2612562545	-0.6524098320
Н	1.0	3.3246003512	2.1474004435	0.0643399089
Н	1.0	4.7706291117	1.2617667465	-0.6530207475

The first analysis of the tools is to calculate the SCF convergence. This calculation is made by iterative steps (maximum of 200), and if it is converge means that it has calculated the MOs correctly.

UHF SCF CALCULATION

NUCLEAR ENERGY = 472.1093670933MAXIT =200 NPUNCH= 2 MULT= 2 EXTRAP=T DAMP=F SHIFT=F RSTRCT=F DIIS=F SOSCF=T DENSITY MATRIX CONV= 1.00 E - 08SOSCF WILL OPTIMIZE 7141 BETA ROTATION ANGLES. 7296 ALPHA AND SOGTOL = 2.500E - 01MEMORY REQUIRED FOR UHF/ROHF ITERS= 663739 WORDS. ITER EX TOTAL ENERGY E CHANGE DENSITY CHANGE ORB. GRAD $1 \ 0$ -386.506..-386.506...0.1588416160.000000000 -START SECOND ORDER SCF--387.552..0.0468390212 1 -1.046...0.0832710873 2 -387.636..0.033275224 -0.083...0.035754195

4	3	-387.664	-0.028	0.034107009	0.014857162
5	4	-387.672	-0.007	0.014524410	0.007080819
6	5	-387.674	-0.001	0.006631104	0.003996855
7	6	-387.674	-0.000	0.002796382	0.001823908
8	7	-387.675	-0.000	0.001608628	0.000739862
9	8	-387.675	-0.000	0.000499825	0.000252934
10	9	-387.675	-0.000	0.000369083	0.000188611
11	10	-387.675	-0.000	0.000573632	0.000193212
12	11	-387.675	-0.000	0.001555634	0.000204483
13	12	-387.675	-0.000	0.004079090	0.000207224
14	13	-387.675	-0.000	0.010511500	0.000343003
15	14	-387.675	-0.000	0.013124786	0.000625395
16	15	-387.675	-0.000	0.003925157	0.000734568
17	16	-387.675	-0.000	0.001459006	0.000479828
18	17	-387.675	-0.000	0.002360579	0.000078118
19	18	-387.675	-0.000	0.000660724	0.000030537
20	19	-387.675	-0.000	0.000153743	0.000016381
21	20	-387.675	-0.000	0.000037904	0.000005277
22	21	-387.675	-0.000	0.000026781	0.000002305
23	22	-387.675	-0.000	0.00003936	0.000001063
24	23	-387.675	-0.000	0.000002720	0.000000491
25	24	-387.675	-0.000	0.00001486	0.000000226
26	25	-387.675	-0.000	0.00000319	0.00000108
27	26	-387.675	-0.000	0.00000200	0.00000035
28	27	-387.675	-0.000	0.00000133	0.00000026
29	28	-387.675	-0.000	0.00000060	0.00000024
30	29	-387.675	0.000	0.00000086	0.00000022
31	30	-387.675	0.000	0.00000184	0.00000019
32	31	-387.675	-0.000	0.00000326	0.00000018
33	32	-387.675	0.000	0.00000435	0.00000024
34	33	-387.675	-0.000	0.00000354	0.00000024
35	34	-387.675	0.000	0.00000128	0.00000017
36	35	-387.675	0.000	0.00000031	0.000000006
37	36	-387.675	-0.000	0.00000015	0.00000002
38	37	-387.675	0.000	0.00000006	0.000000001

The Mulliken and Lowdin population analyses is needed to calculate the charge of all atoms in the given system. The atomic charge is one of the most important figure in this work. This analysis is shown above:

MULLIKEN AND LOWDIN POPULATION ANALYSES

ATOMIC MULLIKEN POPULATION IN EACH MOLECULAR ORBITAL

	1	2	3	4	5
	2.000000	2.000000	2.000000	2.000000	2.000000
1	0.002960	0.005305	0.997891	0.995717	0.000028
2	0.002960	0.005305	0.997891	0.995717	0.000028
3	0.998141	0.995793	0.002986	0.005276	-0.000097
4	0.998141	0.995793	0.002986	0.005276	-0.000097
5	0.000490	-0.000488	-0.000004	-0.000006	-0.00001
6	0.000490	-0.000488	-0.000004	-0.000006	-0.00001
7	0.000490	-0.000488	-0.000004	-0.000006	-0.00001
8	0.000490	-0.000488	-0.000004	-0.000006	-0.00001
9	0.00001	0.00001	-0.000454	-0.000535	-0.000000
10	0.00001	0.00001	-0.000454	-0.000535	-0.000000
11	0.00001	0.00001	-0.000454	-0.000535	-0.000000
12	0.00001	0.00001	-0.000454	-0.000535	-0.000000
13	0.000123	-0.000126	0.000038	0.000090	1.001317

14 15 16 17 18 19 20 21	$\begin{array}{c} 0.000001\\ 0.000001\\ 0.000000\\ 0.000000\\ 0.000000\\ 0.000000\\ 0.000123\\ 0.000001 \end{array}$	$\begin{array}{c} 0.000001\\ 0.000001\\ -0.000000\\ 0.000000\\ -0.000000\\ 0.000000\\ -0.000126\\ 0.000001\end{array}$	$\begin{array}{c} -0.000000\\ -0.000000\\ 0.000000\\ -0.000000\\ 0.000000\\ -0.000000\\ 0.000038\\ -0.000000\end{array}$	$\begin{array}{c} -0.000000\\ -0.000000\\ 0.000000\\ 0.000000\\ 0.000000\\ 0.000000\\ 0.000000\\ -0.000000\\ \end{array}$	$\begin{array}{c} -0.000623\\ -0.000623\\ 0.000000\\ 0.000001\\ 0.000000\\ 0.000001\\ 1.001317\\ -0.000623\end{array}$
22	0.000001	0.000001	-0.000000	-0.000000	-0.000623
23	0.000000	-0.000000	0.000000	0.000000	0.000000
24	0.000000	0.000000	-0.000000	0.000000	0.000001
$\frac{20}{26}$	0.000000			0.000000	0.000000
20	0.000000	0.000000	0.000000	0.000000	0.000001
	6	7	8	9	10
	2.000000	2.000000	2.000000	2.000000	2.000000
1	0.000073	0.000000	0.000000	0.000000	0.000000
2	0.000073	0.000000	0.000000	0.000000	0.000000
3	0.000090	-0.000000	-0.000000	0.000002	0.00002
4	0.000090	-0.000000	-0.000000	0.000002	0.000002
5	0.000001	-0.000000	-0.000000	-0.000000	-0.000000
6 7	0.000001	-0.000000	-0.000000	-0.000000	-0.000000
0	0.000001	-0.000000	-0.000000	-0.000000	-0.000000
0	0.000001	-0.000000	-0.000000	-0.000000	-0.000000
9 10	0.000000	-0.000000	-0.000000	-0.000000	-0.000000
11	0.000000	-0.000000	-0.000000	-0.000000	-0.000000
12	0.000000	-0.000000	-0.000000	-0.000000	-0.000000
13	1.001265	-0.000506	-0.000506	-0.000724	-0.000724
14	0.000623	0.500765	0.500773	0.500832	0.500825
15	0.000623	0.500701	0.500694	0.500897	0.500904
16	0.000000	-0.000242	-0.000242	-0.000249	-0.000249
17	0.00001	-0.000238	-0.000238	-0.000254	-0.000254
18	0.000000	-0.000242	-0.000242	-0.000249	-0.000249
19	0.000001	-0.000238	-0.000238	-0.000254	-0.000254
20	1.001265	-0.000506	-0.000506	-0.000724	-0.000724
21	0.000623	0.500765	0.500773	0.500832	0.500825
22	0.000623	0.500701	0.500694	0.500897	0.500904
23	0.000000	-0.000242	-0.000242	-0.000249	-0.000249
24	0.000001	-0.000238	-0.000238	-0.000234	-0.000234
$\frac{25}{26}$	0.000000	-0.000242 -0.000238	-0.000242 -0.000238	-0.000249 -0.000254	-0.000249
		10	10		
	11	12	13	14	15
	2.000000	2.000000	2.000000	2.000000	2.000000
1	0.255847	0.041522	0.326683	0.177713	0.000787
2	0.255847	0.041522	0.326683	0.177713	0.000787
3	0.240847	0.194306	0.060093	0.320610	0.007633
4	0.240847	0.194306	0.060093	0.320610	0.007633
о 6	0.025669	0.020014	0.002807	0.052071 0.052070	0.001905
7	0.025009	0.020014	0.002807	0.052070 0.052071	0.001910
8	0.025009 0.025660	0.020014 0.020014	0.002807	0.052071 0.052070	0.001903
9	0.028696	0.003284	0.038823	0.017743	0.000323
10	0.028697	0.003285	0.038821	0.017742	0.000324
11	0.028696	0.003284	0.038823	0.017743	0.000323
12	0.028697	0.003285	0.038821	0.017742	0.000324
13	0.212049	0.355797	0.205191	0.113051	0.160194
14	0.079224	0.155568	0.135117	0.095594	0.286542

15	0.079226	0.155564	0.135093	0.095554	0.286601
16	0.006526	0.012897	0.011470	0.008649	0.074979
17	0.005512	0.012427	0.015816	0.020284	0.051898
18	0.006527	0.012897	0.011466	0.008642	0.074991
10	0.005512	0.012001	0.015814	0.000042 0.020277	0.074001
20	0.000012	0.012420 0.255707	0.015014	0.020277	0.160104
20	0.212049 0.070224	0.333797	0.203191 0.125117	0.113031	0.100194
21	0.079224	0.155508	0.135117	0.095594	0.280342
22	0.079226	0.155564	0.135093	0.095554	0.280001
23	0.006526	0.012897	0.011470	0.008649	0.074979
24	0.005512	0.012427	0.015816	0.020284	0.051898
25	0.006527	0.012897	0.011466	0.008642	0.074991
26	0.005512	0.012426	0.015814	0.020277	0.051913
	16	17	18	19	20
	2.000000	2.000000	2.000000	2.000000	2.000000
1	0.000371	0.187315	0.426944	0.026639	0.016436
2	0.000371	0.187315	0.426944	0.026639	0.016436
3	0.007294	0.329478	0.140537	0.107598	0.114665
4	0.007294	0.329478	0.140537	0.107598	0.114665
5	0.001808	0.075705	0.042111	0.019860	0.037940
6	0.001814	0.075703	0.042099	0.019862	0.037945
7	0.001808	0.075705	0.042111	0.019860	0.037940
8	0.001814	0.075703	0.042099	0.019862	0.037945
a	0.001014	0.030761	0.148814	0.001554	0.005111
10	0.000170	0.030762	0.148814	0.001554	0.005111
11	0.000171	0.030702 0.020761	0.148814 0.148814	0.001554	0.005112
11	0.000170	0.030701	0.140014	0.001554	0.005111
12	0.000171	0.030702	0.140014	0.001334	0.005112
13	0.160203	0.102436	0.023334	0.164917	0.148500
14	0.286977	0.055694	0.008455	0.191325	0.185925
15	0.287036	0.055653	0.008439	0.191322	0.185943
16	0.075051	0.005799	0.000948	0.060174	0.065612
17	0.052013	0.022458	0.004284	0.077510	0.065594
18	0.075063	0.005793	0.000946	0.060171	0.065611
19	0.052028	0.022444	0.004276	0.077511	0.065607
20	0.160203	0.102436	0.023334	0.164917	0.148500
21	0.286977	0.055694	0.008455	0.191325	0.185925
22	0.287036	0.055653	0.008439	0.191322	0.185943
23	0.075051	0.005799	0.000948	0.060174	0.065612
24	0.052013	0.022458	0.004284	0.077510	0.065594
25	0.075063	0.005793	0.000946	0.060171	0.065611
26	0.052028	0.022444	0.004276	0.077511	0.065607
	21	22	23	24	25
	2.000000	2.000000	2.000000	2.000000	2.000000
1	0.318257	0.040583	0.052167	0.039419	0.141526
2	0.318257	0.040583	0.052167	0.039419	0.141526
3	0.186725	0.096963	0.121483	0.223542	0.007685
4	0.186725	0.096963	0.121483	0.223542	0.007685
5	0.067580	0.003959	0.024643	0.084066	0.002952
6	0.067594	0.003959	0.024604	0.084093	0.002958
7	0.067580	0.003959	0.024643	0.084066	0.002952
8	0.067594	0.003959	0.024604	0.084093	0.002958
9	0.111881	0.001456	0.007787	0.013856	0.055878
10	0 111870	0 001461	0 007770	0.013845	0.055873
11	0.111010	0 001456	0.007787	0.013856	0.055879
19	0 111970	0.001400	0.007707	0.013845	0.0000000
12 12	0.021270	0.001401	0.001119	0.0103560	0.000070
14	0.021370 0.034747	0.102000	0.121071	0.194719	0.141200
14 15	0.034747	0.204000	0.102/00	0.134713	0.100107
тı)	0.004102	0.404001	0.102110	0.104110	0.100070

16	0 000227	0 120780	0 108176	0.000005	0 001708	
17	0.000221	0.120703	0.100170	0.000000	0.001730	
17	0.022378	0.034305	0.029031	0.084092	0.111077	
18	0.000224	0.120788	0.108174	0.000005	0.001795	
19	0.022388	0.034305	0.029027	0.084092	0.111013	
20	0.021370	0.132305	0.121571	0.103560	0.141203	
21	0.034747	0.204565	0.182783	0.134713	0.183167	
21	0.024752	0.201561	0.182776	0.194719	0.100101	
22	0.004102	0.204001	0.102170	0.104710	0.103070	
23	0.000227	0.120789	0.108176	0.000005	0.001798	
24	0.022378	0.034305	0.029031	0.084092	0.111077	
25	0.000224	0.120788	0.108174	0.000005	0.001795	
26	0.022388	0.034305	0.029027	0.084092	0.111013	
	26	27	28	29	30	
	2 000000	2 000000	2,000000	2 000000	2 000000	
	2.000000	2.000000	2.000000	2.000000	2.000000	
1	0.250249	0 406154	0 110957	0 011049	0.020748	
1	0.330342	0.400134	0.110857	0.011042	0.020748	
2	0.350342	0.406154	0.110857	0.011042	0.020748	
3	0.343779	0.150858	0.163617	0.028573	0.003155	
4	0.343779	0.150858	0.163617	0.028573	0.003155	
5	0.052509	0.011375	0.072454	0.011101	0.002757	
6	0.052107	0.011337	0.072814	0.011108	0.002761	
7	0.052500	0.011375	0.072454	0.011100	0.002757	
1	0.052509	0.011373	0.072404	0.011101	0.002757	
8	0.052107	0.011337	0.072814	0.011108	0.002761	
9	0.038591	0.010713	0.047324	0.006096	0.011464	
10	0.038831	0.010755	0.047088	0.006101	0.011457	
11	0.038591	0.010713	0.047324	0.006096	0.011464	
12	0.038831	0.010755	0.047088	0.006101	0.011457	
13	0.052100	0.065637	0.043371	0.090948	0.163248	
14	0 022941	0.088716	0 122553	0 221153	0 230586	
15	0.022011	0.088776	0.122000	0.221100 0.221146	0.230503	
10	0.023133	0.000770	0.122000	0.221140 0.150642	0.230393	
10	0.004711	0.019091	0.019032	0.139043	0.149372	
17	0.008034	0.058120	0.079857	0.036722	0.012023	
18	0.004750	0.019708	0.018982	0.159642	0.149598	
19	0.008168	0.058160	0.079710	0.036726	0.012038	
20	0.052100	0.065637	0.043371	0.090948	0.163248	
21	0.022941	0.088716	0.122553	0.221153	0.230586	
22	0 023135	0.088776	0 122338	0 221146	0 230593	
22	0.020100	0.010601	0.122000	0.150642	0.200000	
23	0.004711	0.019091	0.019032	0.139043	0.149372	
24	0.008034	0.058120	0.079857	0.036722	0.012023	
25	0.004750	0.019708	0.018982	0.159642	0.149598	
26	0.008168	0.058160	0.079710	0.036726	0.012038	
	31	32	33	34	35	
	2.000000	2.000000	2.000000	2.000000	2.000000	
	2.000000	2.000000	2.000000	2.000000	2.000000	
1	0.086425	0 202655	0.008051	0.281020	0.200125	
1	0.080425	0.392033	0.008051	0.281929	0.209125	
2	0.086425	0.392655	0.008051	0.281929	0.209125	
3	0.275794	0.250811	0.281261	0.109654	0.228707	
4	0.275794	0.250811	0.281261	0.109654	0.228707	
5	0.162810	0.021951	0.023575	0.084622	0.020686	
6	0.162746	0.022116	0.023513	0.084736	0.020645	
7	0 162810	0 021951	0.023575	0 084622	0 020686	
8	0 169746	0.021301	0.020010	0.084796	0.020000	
0	0.102740 0.046701	0.044110	0.020010	0.004/00	0.020040	
9	0.040/91	0.104192	0.002427	0.200288	0.012439	
10	0.046582	0.104376	0.002390	0.200338	0.012450	
11	0.046791	0.104192	0.002427	0.200288	0.012459	
12	0.046582	0.104376	0.002390	0.200338	0.012450	
13	0.067292	0.061053	0.244286	0.010571	0.260781	
14	0.060163	0.018676	0.103033	0.011221	0.065762	
15	0.060245	0.018649	0.103016	0.011265	0.065734	
16	0.007933	0.000824	0.019342	0.000912	0.009241	
-0	0.001000	0.000024	0.010014	0.00012	0.000 FT	

17	0.007633	0.001938	0.084896	0.001770	0.042596
18	0.007918	0.000832	0.019332	0.000907	0.009238
19	0.007667	0.001926	0.084876	0.001787	0.042577
20	0.067292	0.061053	0.244286	0.010571	0.260781
21	0.060163	0.018676	0.103033	0.011221	0.065762
22	0.060245	0.018649	0.103016	0.011265	0.065734
23	0.007933	0.000824	0.019342	0.000912	0.009241
24	0.007633	0.001938	0.084896	0.001770	0.042596
25	0.007918	0.000832	0.019332	0.000907	0.009238
26	0.007667	0.001926	0.084876	0.001787	0.042577
	36	37	38		
	2.000000	2.000000	2.000000		
1	0.038418	0.125940	0.000347		
2	0.038418	0.125940	0.000347		
3	0.023647	0.081052	0.000243		
4	0.023647	0.081052	0.000243		
5	0.012027	0.006195	0.000005		
6	0.011993	0.006178	0.000005		
7	0.012027	0.006195	0.000005		
8	0.011993	0.006178	0.000005		
9	0.000975	0.000900	0.000000		
10	0.000970	0.000898	0.000000		
11	0.000975	0.000900	0.000000		
12	0.000970	0.000898	0.000000		
13	0.428541	0.345371	0.000009		
14	0.241410	0.215109	0.499662		
15	0.241371	0.215081	0.499724		
16	0.000043	0.000358	0.000001		
17	0.000281	0.001281	0.000000		
18	0.000043	0.000358	0.000001		
19	0.000281	0.001280	0.000000		
20	0.428541	0.345371	0.000009		
21	0.241410	0.215109	0.499662		
22	0.241371	0.215081	0.499724		
23	0.000043	0.000358	0.000001		
24	0.000281	0.001281	0.000000		
25	0.000043	0.000358	0.000001		
26	0.000281	0.001280	0.000000		

After this, the charge of each atom is listed:

5	TOT.	AL MULLIKEN	AND LOWDIN	ATOMIC POP	ULATIONS
A	ГОМ	MULL.POP.	CHARGE	LOW.POP.	CHARGE
1	\mathbf{C}	6.311074	-0.311074	6.182458	-0.182458
2	\mathbf{C}	6.311076	-0.311076	6.182460	-0.182460
3	\mathbf{C}	6.267563	-0.267563	6.156002	-0.156002
4	\mathbf{C}	6.267563	-0.267563	6.156002	-0.156002
5	\mathbf{C}	5.965556	0.034444	5.858663	0.141337
6	\mathbf{C}	5.965557	0.034443	5.858663	0.141337
7	\mathbf{C}	6.115265	-0.115265	6.122202	-0.122202
8	\mathbf{C}	6.115260	-0.115260	6.122203	-0.122203
9	\mathbf{C}	6.263573	-0.263573	6.203562	-0.203562
10	\mathbf{C}	6.263575	-0.263575	6.203560	-0.203560
11	Η	0.795061	0.204939	0.855144	0.144856
12	Η	0.815910	0.184090	0.872539	0.127461
13	Η	0.795060	0.204940	0.855143	0.144857
14	Η	0.815909	0.184091	0.872539	0.127461
15	Η	0.850290	0.149710	0.892064	0.107936
16	Η	0.837496	0.162504	0.885043	0.114957
17	Η	0.850291	0.149709	0.892065	0.107935

18 H	0.837496	0.162504	0.885042	0.114958
19 H	0.744216	0.255784	0.836773	0.163227
20 H	0.744216	0.255784	0.836773	0.163227
21 H	0.838142	0.161858	0.881292	0.118708
22 H	0.838143	0.161857	0.881293	0.118707
23 H	0.837857	0.162143	0.869659	0.130341
24 H	0.857995	0.142005	0.884599	0.115401
25 H	0.837857	0.162143	0.869659	0.130341
26 H	0.857996	0.142004	0.884599	0.115401

4.3 Gaussian tutorial

Gaussian09 is another computational chemistry program. Maybe it is the most wellknown and the most developed. Unlike Gamess (which is an open source program), gaussian09 requires to purchase a license in order to use it. In the second part of this chapter will be a brief introduction to the program, the main commands will be listed and the output file will be introduced. For common arguments with Gamess (for example the Z-matrix or the meaning of some specific terms, for example SCF), refer to the Gamess manual.

A brief introduction of this software has already been written in [35], [40], [41].

4.3.1 Input file

The gaussian09 input file is very similar to the gamess one. There is a first part where the commands are given, such as the basis sets, the type of calculation to perform and so on. After the command section the Z-matrix is written 4.1.1. The portion of input file commands is shown below:

```
%NProc=6
Chk=rs3_opt_b3lyp_LANL2DZ_ch1_field_x+1.chk
# b3lyp/lanl2dz geom=connectivity scf=maxcycle=1000
field=x+20 pop=(mk, readradii)
rs3 B3LYP/TZVP field
1 2
  \mathbf{C}
  \mathbf{C}
               1
                             B1
  С
               2
                             B2
                                       1
                                                       A1
  \mathbf{C}
               3
                             B3
                                       \mathbf{2}
                                                       A2
                                                                 1
                                                                                D1
  \mathbf{C}
                                       3
                                                                                D2
               4
                             B4
                                                       A3
                                                                 2
  Fε
                                       2
               1
                             B5
                                                       A4
                                                                 3
                                                                                D3
  _{\rm C}^{\rm C}
               6
                             B6
                                       1
                                                       A5
                                                                 \mathbf{2}
                                                                                D4
                             B7
               \overline{7}
                                                                 1
                                       \mathbf{6}
                                                       A6
                                                                                D5
  \mathbf{C}
               8
                             B8
                                       7
                                                       Α7
                                                                 6
                                                                                D6
  \mathbf{C}
               9
                             B9
                                       8
                                                       A8
                                                                 7
                                                                                D7
  Ċ
              10
                           B10
                                       9
                                                       A9
                                                                 8
                                                                                D8
                                       \overline{7}
                                                                 \mathbf{6}
                                                                                D9
 \begin{array}{c} C \\ C \\ C \end{array}
               8
                            B11
                                                     A10
                                                                               D10
              12
                           B12
                                       8
                                                     A11
                                                                 7
                                                                 7
              12
                           B13
                                       8
                                                     A12
                                                                               D11
  \mathbf{C}
                                                                 8
              14
                           B14
                                      12
                                                     A13
                                                                               D12
```

С	15	B15	14	A14	12	D13
С	16	B16	15	A15	14	D14
С	17	B17	16	A16	15	D15
С	18	B18	17	A17	16	D16
С	16	B19	15	A18	14	D17
С	20	B20	16	A19	15	D18
Ν	21	B21	20	A20	16	D19
С	20	B22	16	A21	15	D20
С	23	B23	20	A22	16	D21
С	24	B24	23	A23	20	D22
\mathbf{C}	21	B25	20	A24	16	D23
\mathbf{C}	24	B26	23	A25	20	D24
С	27	B27	24	A26	23	D25
С	27	B28	24	A27	23	D26
С	29	B29	27	A28	24	D27
С	30	B30	29	A29	27	D28
С	31	B31	30	A30	29	D29
С	29	B32	27	A31	24	D30
Fe	32	B33	31	A32	30	D31
С	34	B34	32	A33	31	D32
С	35	B35	34	A34	32	D33
\mathbf{C}	36	B36	35	A35	34	D34
\mathbf{C}	37	B37	36	A36	35	D35
\mathbf{C}	38	B38	37	A37	36	D36
Η	12	B39	8	A38	7	D37
Η	13	B40	12	A39	8	D38
Η	13	B41	12	A40	8	D39
Η	13	B42	12	A41	8	D40
Η	28	B43	27	A42	24	D41
Η	28	B44	27	A43	24	D42
Η	28	B45	27	A44	24	D43
Η	36	B46	35	A45	34	D44
Η	37	B47	36	A46	35	D45
Η	38	B48	37	A47	36	D46
Η	39	B49	38	A48	37	D47
Η	35	B50	34	A49	32	D48
Η	30	B51	29	A50	27	D49
Η	33	B52	29	A51	27	D50
Η	32	B53	31	A52	30	D51
Η	31	B54	30	A53	29	D52
Η	9	B55	8	A54	7	D53
Η	10	B56	9	A55	8	D54
Η	11	B57	10	A56	9	D55
Η	7	B58	6	A57	1	D56
Η	1	B59	2	A58	3	D57
Η	5	B60	4	A59	3	D58
Η	4	B61	3	A60	2	D59
Η	3	B62	2	A61	1	D60
Н	2	B63	1	A62	5	D61
H	15	B64	14	A63	12	D62
H	19	B65	18	A64	17	D63
H	18	B66	17	A65	16	D64
H	22	B67	21	A66	20	D65
H	26	B68	21	A67	20	D66
H	25	B69	24	A68	23	D67
H	23	B70	20	A69	16	D68
Н	27	B71	24	A70	23	D69
B1		1.4	4327819			
B2		1.4	4047882			
B3		1.4	4067328			
B4		1.4	4219375			

The first line of the input file specifies the number of processors to use in order to do a parallel computation using the % symbol. In the VLSI laboratory there are 14 processors, in that of Micro & Nano 4. Obviously the number of the values of # can not exceed the physicall number of processors. The second line instead the name of the file checkpoint. This file serves mainly two purposes: it is used as a starting point for a second simulation and it is the file that is used by software like Avogadro for the molecule visualization.

In the third row, however, there are all the other commands to be given to gaussian09 to obtain the required results.

• Method and basis sets:

It corresponds to \$SCFTYP and \$GBASIS group in Gamess. Gaussian09 can also perform RHF, UHF and DFT calculations. In the example, the calculations are ub3lyp (DFT) and the basis is lanl2dz but they are optional.

• Geom=connectivity :

The Geom keyword specifies the source of the molecule specification input, options related to coordinate definitions, and geometry related output.

• Scf=maxcylce=N :

This keyword specifies the maximum number of steps for SCF. Usually, N is equal to 2000.

• Pop=(mk, readradii):

It is the type of calculation for the atomic charge. In particular, this command set Merz-Singh-Kollman (MK) scheme for evaluating the atomic charge distribution with the calculation of Mulliken and ESP charges.

• Charge:

To add a point charge driver we have to specify the keyword "charge" in the

command line. The coordinates and the value of the point charge driver will be at the end of the input file (Fe 1.3 in the example): we can add as many point driver as we want.

• External electric field field:

It is possible to insert an external electric field also with gaussian09. In this case, the syntax is *field* = $D \pm N$. *D* is the direction (x, y, z), in the example is x and N is the magnitude of the field in atomic unit $(1 \text{ a } u = 5.14 \cdot 102 \text{ V/nm})$. In the input file listed above, we have an electric field with a modulus of 1.028 V/nm and oriented along the positive X axes.

4.3.2 Running Gaussian Through Gaussian User Interface

After the input file of Gaussian simulation is properly set-up, we could run the software to perform the simulation in order to solve the Schrödinger Equation mentioned in the previous section. But until now, it is needed to point out that we have two ways to perform the running procedure: one is through the Gaussian Viewer Software user interface (could be installed on personal computer) and the other is through the online server of Politecnico di Torino.

One could run the Gaussian simulation through the Gaussian user interface, in this part, a detailed procedure is described. First we run the Gaussian software, which is provided in Windows, on the personal computer where it was installed and open an input file we just defined. Then the pop-up window is shown in Fig. 4.11, from this figure we can observe that, in the pop-up window we can modify every command that was mentioned in the last section, like the method/basis and value of the uniform electric field applied to the molecule, etc. Also the Z-matrix is available in the window and can be changed freely according to the requirement of the user. After configuring the input file, we could run the software by simply clicking the icon RUN on the right of the window and the simulation is performed. As shown in Fig. 4.11, the software is running and the content of the output file is updated frequently as it runs. About the output file, we have perform the detailed analysis later on in the following section.



		ioute oet	staft						
sian-file	es\rs3_op	_b3lyp_L/	ANL2D	Z_ch1_fi	eld_×	+1.gjf 🛅	Additional Steps		0
9	% Section	%NProc=6))					*	? ?5
		%Chk=rs3	_opt_	b3lyp_LA	NL2D	Z_ch1_field_	x+1.chk	Ŧ	
		*					Þ		\$.
Route	e Section	# b3lyp/la	nl2dz	geom=co	onnec	tivity scf=ma	xcycle=1000 field	^	• UK
				Č				-	\$0 1
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Title	e Section	rs3 B3I Y	рли	P field				~	×
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č	2	B2	1	41				-	
č	3	B3	2	A2	1	D1			
С	4	B4	3	A3	2	D2			
Fe	1	B5	2	A4	3	D3			
I C	6	B6	1	A5	2	D4		÷.	

Figure 4.11: Gaussian input file configuration through Gaussian User Interface.

This method is simple and easily used without internet connection, but generally speaking, while you set at first that the number of processors is 14, usually personal computers has only one or few processors; as a consequence of that, the time needed for the simulation in this situation is large. So in order to improve the simulation procedure and save the time, we need a more powerful environment to work with, which brings us to the following section about the online server.

4.3.3 Running Gaussian through online server

• Login:

This server is run in a Linux environment and once you have got the user id and the password, you can login into the server. A screen shot of the server interface is shown in Fig 4.12. Then create a folder which contains the input Gaussian simula- tion files. We could name the input folder as the title of the simulation, for example, rs3:._x 200.gjf. In the Fig 4.13 there are some already named input file ready to be executed.

NX - ruiyu.wang@vlsi20.1007 - VLSI (GPL Edition)			
	In Construction Constructio		© © ©
	Back Forward Icons Details Columns Preview Split		Information (A) (A)
🔳 🕥 ruivu.wang@vlsi20:~			Heme
jie £dit View Search Ierminal <u>H</u> elp ruxyugvtsi20:-\>	avogadro cache cache cache dous emacs d geonf	◆ Size Date	Molified 2013 07-17 12-41 hype Felder
🔅 📦 🔜 🛤 ruiyu.wang - Dolphin	⊂ ruiyu.wang@vlsi20:~		12:43

Figure 4.12: The interface of the server after login.

😫 💽 cloci	k_signal - Dolphin			\odot		×
<u>File Edit</u>	View Go Tools Settings Help					
Back Forward	d Icons Details Columns Preview Split					86
Places 🛞 🛞	> Home > test > with_thiol > reduced > clock_sign	al		Information	۲	8
🔚 Home	Name	♥: Size	Date	clock_signal		
Network	<pre>ms rs3_6te_b3lyp_LANL2DZ_ch1_clock_x-40.gjf</pre>	15.7 KiB	2013-07-12 12:09			
Root	ns rs3_6te_b3lyp_LANL2DZ_ch1_clock_x-200.gjf	15.7 KiB	2013-07-12 12:08			
Trash	ns rs3_6te_b3lyp_LANL2DZ_ch1_clock_x+40.gjf	15.7 KiB	2013-07-12 12:09			-
0	ns rs3_6te_b3lyp_LANL2DZ_ch1_clock_x+200.gjf	15.7 KIB	2013-07-12 12:09			
					_	
				Madified: 2012 07 1	12.12	10
				Type: Folder	12 12	.10
	4 Files (62.8 KiB)	Q	() @			

Figure 4.13: Some input Gaussian file in the folder of the server.

• Initialize the Gaussian software:

After preparing the input files, we need to initialize the Gaussian software. First open the terminal on the server and type the command source /software/scripts/init_g09 to initialize the Gaussian software on the server. Then the return instruction tells us that we could use all the commands that contains in the /g09 directory.

• Run Gaussian software: After initializing Gaussian software, we could perform the simulation. Change the current directory to the folder which contains the input files that needed to be executed, mentioned previously. Then type the command g09 input file name to run the Gaussian software on the server; for example, g09 rs3 ... x-200.gjf. The Gaussian software will generate the checkpoint file (.chk) and the output Gaussian file (.log) immediately and update them frequently until the simulation is finished. And at the same time we could perform also the inspection of the content at the output file and state of our current simulation job. By typing the command tail -f file name, for example, tail -f rs3 6te b3lyp LANL2DZ ch1 clock x-200.gjf, one could see the content at the very end of the corresponding output file and it updates. Furthermore, by typing the command top, one could see the current simulation jobs existing on the server right now the occupancy of cpu processors and memories, etc. Usually, the simulation would take few hours to finish the job depending the requirements that you have put in the input file, for example, the simulation time decreases as the number of processors increases and if we have defined a more accuracy simulations with lower convergency error, and the time consumed will be larger. During simulation, Gaussian have written all the relevant information, such as the optimized structure, the charge distribution, the energy distribution, etc, into the output file. So an detailed analysis of the output Gaussian file is necessary and provided in the next section.

4.3.4 Gaussian Output file

In this section the interisting parts of the gaussian09 output file will be analyzed. The first part of the output file is a summary of all commands given in the input file. After that, the software will calculate the convergence of the SCF and, once finished, will displays the optimized geometry of the system under the heading "Standard orientation".

SCF Don A.U. af Convg	ne: E(U) ter 137 = 0.5	B3LYP) 7 сус 5178D-) = - les -05	-1934 —\	.75497408 V/T = 2.0	0627	
			S	tanda	ard orient	tation:	
Center	Atomic	Atom	ic	Co	ordinates	(Angst	roms)
Number	Number	Type	Х		Υ		Ζ
1	6	0	-4.91	8827	-6.5031	165 -	-0.858753
2	6	0	-3.57	9317	-6.9858	817 -	-0.896848
3	6	0	-3.08	1190	-7.0062	256	0.436876
4	6	0	-4.11	2678	-6.5356	503	1.299485
5	6	0	-5.24	8466	-6.2257	703	0.498685
6	26	0	-3.62	1776	-5.0613	555 -	-0.093920
7	6	0	-3.05	0836	-3.3213	389	0.899548
8	6	0	-1.96	2576	-3.7964	163	0.105018
9	6	0	-2.41	6639	-3.8223	346 -	-1.249553
10	6	0	-3.76	3476	-3.3657	730 -	-1.289245
11	6	0	-4.158567	-3.058639	0.042877		
-----------------	---------	---	------------------------	------------------------	------------------------		
12	6	0	-0.558257	-4.131058	0.553616		
13	6	0	-0.465246	-4.439280	2.058170		
14	6	0	0.444124	-3.044541	0.163909		
15	6	0	0.139553	-1.691654	0.283352		
16	6	0	1.096090	-0.729637	-0.041332		
17	6	0	2.373864	-1.140125	-0.491853		
18	6	0	2.696541	-2.486842	-0.619626		
19	6	0	1.721577	-3.418903	-0.288960		
20	6	0	1.100825	0.719432	-0.041284		
21	6	Õ	2.381274	1.121592	-0.491743		
$\frac{1}{22}$	7	Õ	3.133515	-0.011731	-0.755438		
23	6	Õ	0.150634	1.687664	0.283527		
24	6	0	0.464079	3.038546	0.164253		
25	6	Õ	1.743936	3.404577	-0.288609		
$\frac{-6}{26}$	6	Ő	2.712770	2.466172	-0.619380		
$\frac{-0}{27}$	6	Ő	-0.531171	4.131502	0.554190		
28	6	Õ	-0.436340	4 438604	2 058855		
29	6	Õ	-1.937621	3,806277	0 105337		
30	6	Õ	-2.391378	3 835498	-1249267		
31	6	0	-3741277	3 388017	-1.249207 -1.289156		
30	6	0	-4.138545	3 083330	0.042875		
32	6	0	-3.020135	3 338/16	0.042070		
34	26	0	-3.588201	5 082574	-0.093538		
35	20 6	0	-5.207203	6.957731	0.498493		
36	6	0	-3.207203 -4.875301	6 532672	-0.858028		
37	6	0	-3532675	7.006339	-0.896808		
38	6	0	-3.034728	7.000333 7.023710	0.437034		
39	6	0	-4.069553	6 560177	1 299498		
40	1	0	-0.259057	-5.036331	0.016672		
40	1	0	-0.233037 -1.146080	-5.245724	2 336003		
41	1	0	0 550087	-3.243724 -4.738987	2.330333		
42	1	0	-0.710502	-3.550207	2.515520		
40	1	0	-1.111879	5 240435	2.000141		
45	1	0	-0.687484	3 559987	2.656509		
46	1	0	0.581807	4730823	2.000000		
40	1	0	-5.520928	6.389455	-1.710657		
48	1	Õ	-2.983376	7 283984	-1.782307		
40	1	0	-2.000070 -2.043784	7.200004 7.323560	0 739444		
50	1	0	-3 999349	6 443893	2 369292		
51	1	0	-6.147953	5 869248	0.854797		
52	1	0	-1 803305	4 150818	-2 007183		
53	1	0	-3.027230	3 221106	1 971365		
54	1	0	-5.027200 -5.112605	2.740301	0 353495		
55	1	0	-4.361577	2.740301	-2 168613		
56	1	0	-1.830752	-4 141442	-2.100010		
57	1	0	-1.830752 -4.383356	-3.202640	-2.037570 -2.168648		
58	1	0	-5.130361	-2709120	0 353637		
59	1	0	-3.048081	-3.204310	1 971248		
60	1	0	-5.563611	-6.355854	-1.710355		
61	1	0	-6.186526	-5.830003	0.855138		
62	1	0	-4.041421	-6.419502	2 369229		
63	1	0	-2.041421 -2.002175	-7.312560	0 739115		
64	1	0	2 0 2 2 1 7 2	7 267228	1 789496		
65	1	0	-5.052125	-1.207558 1.386570	-1.182420		
66	1	0	-0.843990 1 053001	4 473014	0.013822		
67	1	0	3 670610	-9.80/079	-0.001001		
69	1	0	3 699001	-2.004970 9 777059	-0.311393		
60	1	0	0.000901 1.020074	4.111903 1 159059	-0.971720		
70	1	0	1.302214	4.400000	-0.30/130		
70	1	0	-0.000000 -0.225077	1.309033 5 09/077	0.014000		
11	L G	0	-0.220911	0.0349//	1 166520		
14 79	U G	0	5.072010	0.022051	-1.400038		
10	U	U	0.972919	-0.023631	-0.304428		

74	1	0	5.247605	0.856108	-2.059664
75	1	0	5.240630	-0.891176	-2.063173
76	1	0	5.809820	0.849262	0.232206
77	1	0	5.802845	-0.898023	0.228698
78	6	0	7.310461	-0.028233	-0.841252
79	1	0	7.473558	-0.901343	-1.437891
80	1	0	7.480537	0.845943	-1.434372
81	6	0	8.205850	-0.034020	0.260858
82	6	0	9.543380	-0.041005	-0.215969
83	1	0	8.034463	-0.907207	0.855056
84	1	0	8.044074	0.840068	0.856423
85	1	0	9.714769	0.832186	-0.810162
86	1	0	9.705154	-0.915090	-0.811540
87	6	0	10.438768	-0.046792	0.886140
88	1	0	10.276994	0.827298	1.481704
89	1	0	10.267380	-0.919978	1.480339
90	16	0	11.776298	-0.053777	0.409313
91	1	0	11.712547	0.943917	0.027943

After this, the software calculates both the mulliken and ESP charge for each atom:

Mulliken atomic charges:

1	\mathbf{C}	-0.221124
2	\mathbf{C}	-0.223534
3	\mathbf{C}	-0.238507
4	\mathbf{C}	-0.228625
5	\mathbf{C}	-0.225585
6	Fe	-0.199686
7	\mathbf{C}	-0.372597
8	\mathbf{C}	0.410068
9	\mathbf{C}	-0.392602
10	\mathbf{C}	-0.226502
11	\mathbf{C}	-0.254862
12	\mathbf{C}	-0.352958
13	\mathbf{C}	-0.728372
14	\mathbf{C}	0.617275
15	\mathbf{C}	-0.554318
16	\mathbf{C}	0.147861
17	\mathbf{C}	0.105610
18	\mathbf{C}	-0.369091
19	\mathbf{C}	-0.486003
20	\mathbf{C}	0.141926
21	\mathbf{C}	0.106513
22	Ν	-0.197087
23	\mathbf{C}	-0.548822
24	\mathbf{C}	0.621526
25	\mathbf{C}	-0.487745
26	\mathbf{C}	-0.368574
27	\mathbf{C}	-0.349308
28	\mathbf{C}	-0.694479
29	\mathbf{C}	0.387699
30	\mathbf{C}	-0.354662
31	\mathbf{C}	-0.192878
32	\mathbf{C}	-0.216721
33	\mathbf{C}	-0.329718
34	Fe	-0.148018
35	\mathbf{C}	-0.200374
36	\mathbf{C}	-0.198491
37	\mathbf{C}	-0.199462
38	\mathbf{C}	-0.209320
39	\mathbf{C}	-0.196901

40	Н	0.205347					
41	Η	0.227280					
42	Η	0.211820					
43	H	0.201264					
44	H	0.211844					
45	H U	0.213162					
40	п ц	0.230390					
47	H	0.303387 0.293477					
49	н	0.283917					
50	Н	0.295778					
51	Η	0.310899					
52	Η	0.300403					
53	Η	0.298723					
54	Η	0.300730					
55	H	0.301375					
56	Н	0.269161					
57	H	0.271757					
50 50	п ц	0.200403					
59 60	н	0.233300 0.279294					
61	н	0.273935					
62	Н	0.240110					
63	Н	0.245803					
64	Η	0.262697					
65	Η	0.252826					
66	Η	0.232094					
67	Η	0.213693					
68	Н	0.216624					
69 50	Н	0.228475					
70	H	0.242723 0.100205					
71	пС	0.199393					
73	c	-0.239049 -0.372663					
74	н	0 193982					
75	Н	0.197360					
76	Н	0.220536					
77	Η	0.222355					
78	\mathbf{C}	-0.352521					
79	Η	0.193135					
80	Н	0.188875					
81	С	-0.353446					
82	C	-0.356126					
83 84	п ц	0.201132 0.105466					
85	н	0.200221					
86	Н	0.223291					
87	\mathbf{C}	-0.981432					
88	Η	0.265158					
89	Η	0.290753					
90	\mathbf{S}	0.342422					
91	Η	0.004046					
Charges Charge=	fron 1	n ESP fit , RMS= .000000 Dipole=	$0.00241 \\ -14.9919$	RRMS = 0.0 17.8193	3237: -2.3172	Tot=	23.4020
1	С	-0.080041					
2	č	-0.145153					
2	$\tilde{\mathbf{C}}$	-0.117293					
4	$\tilde{\mathbf{C}}$	-0.124639					
5	\mathbf{C}	-0.095515					
6	Fe	-0.035222					
7	\mathbf{C}	-0.218018					

8	\mathbf{C}	0.087472
9	\mathbf{C}	-0.216355
10	\mathbf{C}	-0.102856
11	\mathbf{C}	-0.198681
12	\mathbf{C}	0.055317
13	\mathbf{C}	-0.478739
14	\mathbf{C}	0.116963
15	\mathbf{C}	-0.204041
16	\mathbf{C}	-0.090490
17	\mathbf{C}	0.379271
18	\mathbf{C}	-0.361242
19	\mathbf{C}	-0.237414
20	\mathbf{C}	-0.060053
21	\mathbf{C}	0.375767
22	Ν	-0.474431
23	\mathbf{C}	-0.215194
24	\mathbf{C}	0.090075
25	\mathbf{C}	-0.252744
26	\mathbf{C}	-0.343626
27	\mathbf{C}	-0.009742
28	\mathbf{C}	-0.384757
29	\mathbf{C}	0.147562
30	\mathbf{C}	-0.164260
31	\mathbf{C}	-0.039338
32	\mathbf{C}	-0.144109
33	\mathbf{C}	-0.144862
34	\mathbf{Fe}	-0.028176
35	\mathbf{C}	-0.057776
36	\mathbf{C}	-0.036953
37	\mathbf{C}	-0.098178
38	\mathbf{C}	-0.061542
39	\mathbf{C}	-0.061836
40	тт	0 0 5 0 1 1 5
40	н	0.079417
$40 \\ 41$	н Н	$0.079417 \\ 0.139449$
$40 \\ 41 \\ 42$	H H H	0.079417 0.139449 0.127097
$40 \\ 41 \\ 42 \\ 43$	H H H H	0.079417 0.139449 0.127097 0.106225
40 41 42 43 44	H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470 \end{array}$
40 41 42 43 44 45	H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470\\ 0.102009 \end{array}$
$ \begin{array}{r} 40 \\ 41 \\ 42 \\ 43 \\ 44 \\ 45 \\ 46 \\ \end{array} $	H H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470\\ 0.102009\\ 0.133709 \end{array}$
$ \begin{array}{r} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ \end{array} $	H H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470\\ 0.102009\\ 0.133709\\ 0.159295 \end{array}$
40 41 42 43 44 45 46 47 48	H H H H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470\\ 0.102009\\ 0.133709\\ 0.159295\\ 0.161503 \end{array}$
$ \begin{array}{r} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ \end{array} $	H H H H H H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470\\ 0.102009\\ 0.133709\\ 0.159295\\ 0.161503\\ 0.146875 \end{array}$
$ \begin{array}{r} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ \end{array} $	H H H H H H H H H H H	$\begin{array}{c} 0.079417\\ 0.139449\\ 0.127097\\ 0.106225\\ 0.106470\\ 0.102009\\ 0.133709\\ 0.159295\\ 0.161503\\ 0.146875\\ 0.151044 \end{array}$
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ \end{array}$	H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ \end{array}$	н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ \end{array}$	н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ \end{array}$	н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н Н	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.160822 0.174733
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 59\end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 59\\ 60\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330 0.145478
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 59\\ 60\\ 61\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330 0.145478 0.141611
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 59\\ 60\\ 61\\ 62\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330 0.145478 0.141611 0.118465
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 60\\ 61\\ 62\\ 63\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.174733 0.159330 0.145478 0.141611 0.118465 0.129092
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 59\\ 60\\ 61\\ 62\\ 63\\ 64\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330 0.145478 0.141611 0.118465 0.129092 0.148758
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 58\\ 59\\ 60\\ 61\\ 62\\ 63\\ 64\\ 65\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330 0.145478 0.141611 0.118465 0.129092 0.148758 0.171063
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 54\\ 55\\ 56\\ 57\\ 8\\ 59\\ 60\\ 61\\ 62\\ 63\\ 64\\ 65\\ 66\end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.169313 0.149563 0.166931 0.149563 0.159330 0.145478 0.141611 0.118465 0.129092 0.148758 0.171063 0.151108
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 49\\ 50\\ 51\\ 52\\ 53\\ 55\\ 56\\ 57\\ 58\\ 96\\ 61\\ 62\\ 63\\ 66\\ 67\\ \end{array}$	H H H H H H H H H H H H H H H H H H H	0.079417 0.139449 0.127097 0.106225 0.106470 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.188968 0.166931 0.149563 0.166931 0.149563 0.166931 0.149563 0.164722 0.174733 0.159330 0.145478 0.141611 0.118465 0.129092 0.148758 0.171063 0.207316
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 9\\ 50\\ 51\\ 2\\ 53\\ 55\\ 57\\ 58\\ 9\\ 60\\ 61\\ 62\\ 63\\ 64\\ 66\\ 67\\ 68\\ \end{array}$	н н н н н н н н н н н н н н н н н н н	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161303 0.188968 0.166931 0.149563 0.160822 0.174733 0.159330 0.145478 0.141611 0.118465 0.129092 0.148758 0.151108 0.207316 0.209188
$\begin{array}{c} 40\\ 41\\ 42\\ 43\\ 44\\ 45\\ 46\\ 47\\ 48\\ 9\\ 50\\ 51\\ 2\\ 53\\ 44\\ 55\\ 57\\ 58\\ 9\\ 60\\ 61\\ 62\\ 63\\ 64\\ 65\\ 66\\ 68\\ 9\end{array}$	н н н н н н н н н н н н н н н н н н н	0.079417 0.139449 0.127097 0.106225 0.106470 0.102009 0.133709 0.159295 0.161503 0.146875 0.151044 0.163618 0.161341 0.176733 0.160822 0.174733 0.160822 0.174733 0.159330 0.145478 0.141611 0.118465 0.129092 0.148758 0.151108 0.209188 0.154750

71	Н	0.087156
72	\mathbf{C}	-0.132970
73	\mathbf{C}	0.091021
74	Η	0.111644
75	Η	0.118075
76	Η	-0.003851
77	Η	0.010561
78	\mathbf{C}	0.266237
79	Η	-0.065648
80	Η	-0.062238
81	\mathbf{C}	-0.021247
82	\mathbf{C}	0.317494
83	Η	-0.069288
84	Η	-0.035506
85	Η	-0.101566
86	Η	-0.065233
87	\mathbf{C}	0.092996
88	Η	-0.006425
89	Η	0.120334
90	\mathbf{S}	-0.603377
91	Η	0.321584

4.4 Avogadro

When the simulation is over for both Gamess and Gaussian, you can analyze the simulation results using Avogadro. Avogadro is an advanced molecule editor and visualizer designed for cross-platform use in computational chemistry, molecular modeling, bio-informatics, materials science, and related areas. It offers flexible high quality rendering and a powerful plug-in architecture.

Now you have two possibilities: either install Avogadro on your laptop or use Avogadro through the server. In the first case, when the software is opened you have to click on File \rightarrow Import \rightarrow Molecule File and then select GAMESS output or the checkpoint file of Gaussian. In the second case you have to type the command **avogadro** \mathcal{C} in the terminal of the server that we have mentioned before. The molecule will be visualized in the user interface: 4.4 - Avogadro



Figure 4.14: Visualizing the molecule with Avogadro.

The tables to the right lists all molecular orbitals. If the status bar is full, you can click on the row of the orbital and a quick low quality rendition of the orbital will be created. An example of HOMO (Highest Occupied Molecular Orbital) is shown in Fig. 4.15.



Figure 4.15: HOMO orbitals.

Chapter 5

Gamess VS Gaussian09

One of the purposes of this work thesis is to compare the results obtained by using GAMESS with those obtained with Gaussian09 in the previous thesis. It is important to understand which results have to be compared. The results of a simulation depends on the input file, changing it you can have every kind of results. But there are some type of computation that a specific tools can not do. For example, you can not calculate the ESP charges in Gamess, but just the Mulliken and Lodwin population. This is an important constraint for this work and, in this case, a post-processing work has to be done for evaluating it [39].

Since we are focused on the charge localization in this thesis, just the Mulliken and Lodwin population and the electrostatic potential (for the ESP algorithm [39]) have been analyzed for both Gaussian09 and Gamess for two candidate molecules (diallyl butane and decatriene) in two different conditions:

- at the equilibrium;
- with a switching field.

The bis-ferrocene molecule was not analyzed in this first phase of the work since, given the complexity of itself and the very high computational cost especially in terms of time, would not have been convenient.

5.1 Mulliken population analysis

Before starting with the comparison between Gamess and Gaussian09 results, let's review what Mulliken population is [42]. Population analysis is the study of charge distribution within molecules. Mulliken population analysis is by default always performed both in Gamess and in Gaussian.

Let's suppose that we have N number of electrons within a molecule. In this case N can be expressed as [43]

$$N = \sum_{j}^{electrons} \left(\sum_{r} c_{jr}^2 + \sum_{r \neq s} c_{jr} c_{js} S_{rs} \right)$$
(5.1)

The problem of the Mulliken population analysis is that it splits the shared electrons from two atoms to fifty-fifty. The advantages of this model is that it is computationally cheap. It works well for comparing changes in partial charge assignment between two different geometries when the same size basis set is used. The main disadvantage is that the partial charges assigned to atoms varies significantly for the same system when different size basis sets are used, so computations using different basis sets cannot be compared.

5.2 Electrostatic potential

For the evaluation of the electrostatic potential in a nucleus, that nucleus is ignored, avoiding a singularity. All other atoms in the molecule contribute to the calculation of electrostatic potential except for the nucleus where the electrostatic potential is calculated [32]. The analysis of electrostatic potential is very important since author in [39] starts from this result for the ESP calculation.

5.3 Diallyl butane

Let's consider a Diallyl butane molecule as in Fig. 5.1.



Figure 5.1: Diallyl butane molecule.

As mentioned before, we have to study two configuration: at the equilibrium and with the switching field.

5.3.1 At the equilibrium

First, the Mulliken charges and the electrostatic potential are analyzed at the equilibrium condition. This means that the molecule is neutral (total charge equal to 0) and there is not an external electric field. The geometry of the molecule is optimized to have the minimum possible energy in the system.

The input files for both Gamess and Gaussian are listed here:

• Gamess input:

\$BASIS GBASIS=STO NGAUSS=3 \$END \$CONTRL SCFTYP=RHF RUNIYP=OPTIMIZE COORD=ZMT \$END \$STATPT OPTTOL=0.0001 NSTEP=20 \$END

• Gaussian input:

```
%chk=1-4_diallyl_butane_opt.chk
# opt rhf/sto-3g geom=connectivity
```

In Fig. 5.2 and 5.3 the results of these comparable simulation and the absolute error (evaluated as $|V_{gamess} - V_{gaussian09}|$), for all the values see Appendix A.



Figure 5.2: Comparison of Mulliken charges for a molecule of Diallyl butane at the equilibrium.



Figure 5.3: Error in the computation of Mulliken charges for a molecule of Diallyl butane at the equilibrium between Gaussian09 and Gamess.

5.3.2 Switching field

In this second case, a switching field of $0.001 \,\mathrm{au} = 0.514 \,\mathrm{V/nm}$ is applied, as shown in Fig. 5.1. Again the results are listed in Appendix B and shown in Fig. 5.4 and 5.5.



Figure 5.4: Comparison of Mulliken charges for a molecule of Diallyl butane with an external electric field of 0.514 V/nm.



Figure 5.5: Error in che computation of Mulliken charges for a molecule of Diallyl butane with an external electric field of $0.514\,\rm V/nm$ between Gaussian09 and Gamess.

Here the electrostatic potential at the nuclei is shown, table in Appendix C:



Figure 5.6: Comparison of electrostatic potential for a molecule of Diallyl butane with an external electric field of 0.514 V/nm.



Figure 5.7: Error in the computation of the electrostatic potential for a molecule of Diallyl butane with an external electric field of 0.514 V/nm between Gaussian09 and Gamess.

5.4 Decatriene

Now, let's consider the second molecule candidate for the QCA technology, Fig. 5.8.



Figure 5.8: Decatriene molecule.

In this second case, we are evaluating the *aggregate charge* instead to the charge for each atom. Tor do this, we have to define the three dots of the Decatriene. Fig. 6.2shows a molecule of Decatriene with the three dots.



Figure 5.9: Decatriene molecule and dots definition.

Dot 1 is made up of C8, C10, H22, H25 and H26 atoms, Dot 2 is made up of C7, C9, H21, H23 and H24, finally Dot 3 is made up of C5, C6, H19 and H20 atoms, see table 5.1.

	Atoms
Dot 1	C8, C10, H22, H25, H26
Dot 2	C7, C9, H21, H23, H24
Dot 3	C5, C6, H19, H20

Table 5.1: Dots definition.

5.4.1 @ the equilibrium

Once we defined the Dot for the decatriene, we can calculate the charges in every dot of the molecule, as listed in table 5.2:

	Gaussian09	Gamess	Error
Dot 1	0.011968	0.01124	0.000728
Dot 2	0.011974	0.012778	0.000804
Dot 3	0.011974	0.03182	0.00855

Table 5.2: Dot charge comparison for a Decatriene molecule at the equilibrium.



Figure 5.10: Dot charge comparison for a Decatriene molecule at the equilibrium.

5.4.2 Switching field

Now, let's focus on a Decatriene molecule inside a switching field of magnitude of 0.514 V/nm, as seen before for the Dially butane molecule, Fig. 5.11. Again, the Dot charges is listed in the table 5.3.



Figure 5.11: Decatriene molecule in presence of a switching field of 0.514 V/nm.

	Gaussian09	Gamess	Error
Dot 1	0.001929	0.003675	0.001746
Dot 2	0.022024	0.01933	0.002694
Dot 3	0.023303	0.031792	0.008489

Table 5.3: Dot charge comparison for a Decatriene molecule with a switching field of $0.514 \,\mathrm{V/nm}$.



Figure 5.12: Dot charge comparison for a Decatriene molecule with an external electric field of $0.514 \,\mathrm{V/nm}$.

5.5 Final considerations

As can be seen from a first analysis, the results obtained from this comparison between GAMESS and gaussian09 are almost identical. Since the molecules are very simple and do not need many steps for SCF convergence, the approximation introduced by the algorithm is negligible. Despite this, the results obtained with Gamess regarding the distribution of the charges in the molecule are not very precise, as we recall that the ESP charges are much more precise. Moreover, during the simulation Gamess creates temporary files much bigger than gaussian09 (in particular for the bis-ferrocene there is a ratio of 32:1) and finally there is no clear and default procedure to handle the ideal point charge, while there is one in gaussian09 and other software. The advantage of Gamess is the ease of use for simulating systems that are not too complex and the fact that it is open source.

Chapter 6

Charge distribution

As we said in Chapter 1, a single molecule represents just half a QCA cell, since it has just 2 or 3 dots instead of 4 (for the Diallyl butane) and 6 (for decatriene and bis-ferrocene molecules). In this thesis we have analyzed the properties of half a QCA cell (single molecule) through ab initio simulations.

6.1 Diallyl butane and decatriene

The first molecule that has been analyzed is the Diallyl Butane [21], shown in Fig. 6.1. The two allyl groups represent the dots (circled in the figure) and so they encode the two logic state '0' and '1' depending on where the aggregate charge is localized.

The second molecule analyzed in this work is the decatriene, Fig. 6.2. It has three dots respect of the diallyl butane, which are ethylene groups. These two molecules have been analyzed in two configurations: neutral and oxidized. By definition "a neutral molecule is a molecule in which the number of electrons surrounding the nuclei is the same as the total number of protons in the nuclei, so that there is no net electric charge" [44] while if the molecule is oxidized it has a free positive charge, that is to say:

 $Total charge = \begin{cases} 0 & neutral molecule \\ 1 & oxidized molecule \end{cases}$



Figure 6.1: Diallyl Butane molecule and dots definition.



Figure 6.2: Decatriene molecule and dots definition.

6.1.1 Neutral molecules

The diallyl butane and the decatriene neutral molecules, which are ideal molecules for the QCA technology, have been studied in order to evaluate the effects of the switching field and of the point charges on them; in this way we can evaluate the effectiveness of this method on real molecules (the bis-ferrocene).

First, in the Table 6.1, we listed the aggregated charges of Dot1 and Dot2 for the diallyl butane as function of the switching field $(\pm 3 \text{ V/nm})$. The graph in Fig. 6.3 shows that the molecule, also if it is neutral, has a strong internal charge displacement as a function of the switching field. In particular, we can notice from table 6.1 that the module of the aggregated charges in the two dots is equal and just the sign changes. This means that the diallyl butane is an idea candidate in terms of charge localization for our technology. The problem with this molecule is that we have just 2 dots instead of 3, necessary for the clock.

Switching field [V/nm]	Dot 1	Dot 2
-3.0	0.3115	-0.3476
-2.0	0.2016	-0.2378
-1.0	0.0918	-0.128
0	-0.0181	-0.0181
+1.0	-0.128	0.0918
+2.0	-0.2378	0.2016
+3.0	-0.3476	0.3115

Table 6.1: Diallyl butane molecule: dot charges as function of the switching field.



Figure 6.3: Diallyl butane: dot charges as function of the switching field.



Figure 6.4: Displacement of the charge inside the diallyl butane molecule at the equilibrium (a) and with a switching field of -3 (b) and +3 V/nm (c).



Figure 6.5: Diallyl butane with point charges configuration.

Driver 1	Driver 2	Dot 1	Dot 2
0.1	0.9	0.0512	-0.0858
0.2	0.8	0.0341	-0.0687
0.3	0.7	0.0171	-0.0517
0.4	0.6	≈ 0	-0.0346
0.5	0.5	-0.0170	-0.0175
0.6	0.4	-0.0341	≈ 0
0.7	0.3	-0.0511	0.0166
0.8	0.2	-0.0682	0.0336
0.9	0.1	-0.0852	0.0507

Table 6.2: Diallyl butane molecule : dot charges as function of the driver polarization.



Figure 6.6: Diallyl butane: dot charges as function of the polarization of the driver.

The same type of analysis was done for the neutral decatriene molecule. Also in

this case, as shown in the table 6.3 and in the Fig. 6.7 the charge values present in Dot1 and Dot2 are almost perfectly symmetrical. The difference with the butane molecule is the presence of the third dot; in fact, in this case all the charge is not in the two dots but is divided among three. This fact will however be clearer when the molecule is oxidized. As we can see in Fig. 6.7 and table 6.3 the charge on the central dot remains almost constant, this is due to the fact that the electric field is parallel to the axis that joins the two dots and therefore the third dot is not affected by it. To interest the third dot on the charge switching, you have to use a *clock field* that is perpendicular to the axis that joins dot1 and 2 and that helps to move the charge present in the third dot in the two dots during the clock phase and to move most of the charge in Dot1 and 2 in the third dot in the relax phase.

Switching field [V/nm]	Dot 1	Dot 2	Dot 3
-4.0	-0.0196	0.0430	0.0334
-3.0	-0.0119	0.0351	0.0328
-2.0	-0.0041	0.0272	0.0323
-1.0	0.0037	0.0193	0.0317
0	0.0115	0.0115	0.0313
+1.0	0.0192	0.0036	0.0308
+2.0	0.0270	-0.0042	0.0305
+3.0	0.0349	-0.0121	0.0301
+4.0	0.0427	-0.0199	0.0298

Table 6.3: Neutral decatriene molecule : dot charges as function of the switching field.



Figure 6.7: Decatriene: dot charges as function of the switching field.

6.1.2 Oxidized molecules

Since in the neutral molecules analyzed up to now the charge that could switch was small, the oxidation technique of a molecule was used. In this way, a charge has been added inside the molecule so that the total charge quantity is 1. As we can see from Fig. 6.8, the oxidized diallyl butane shows a greater charge switching than the neutral molecules and this means a greater V_{out} . The diallyl butane shows the ideal case since there are only two dots; this free charge is distributed equally between them. In this case, as we can see from the Fig. 6.8, the trend of the charge in the molecule undergoes a sharp variation between $\pm 1 \text{ V}$, this means that a field of 2 V is enough to change the molecule status.

Switching Field [V/nm]	Dot 1	Dot 2
-5.0	-0.0217	0.8651
-4.0	-0.0139	0.8550
-3.0	-0.0060	0.8449
-2.0	0.0018	0.8346
-1.0	0.0097	0.8242
0	0.4358	0.4358
+1.0	0.8242	0.0097
+2.0	0.8346	0.0018
+3.0	0.8449	-0.0060
+4.0	0.8550	-0.0139
+5.0	0.8651	-0.0217

Table 6.4: Oxidized Diallyl butane molecule : dot charges as function of the switching field.



Figure 6.8: Oxidizied diallyl butane: dot charges as function of the switching field.

For the Decatriene molecule, and as we will also see for the bis-ferrocene, the same does not happen. In this case, because of the greater complexity of the molecule due to the presence of the third dot, the extra free charge is not equally distributed. To be precise it goes almost entirely on the third dot, making the switch almost equal to the case of the neutral molecule. As can be seen from the gaussian graph instead [35], Fig. 6.9

this distributed the charge in the two active dots and therefore the molecule switch was much better for our purposes. In reality, where to put the extra charge is in no way manageable by us but it is the software, through an energy simulation, to understand where to put it to minimize the energy of the molecule and therefore make it more stable.



Figure 6.9: Oxidized decatriene: dot charges as function of the switching field [35].

Fig. 6.10 shows the comparison between the HOMO, HOMO-1 and HOMO-2 orbitals of the neutral or oxidized decatriene. As you can see, the HOMO orbital (which is where the extra charge of oxidation goes) is present only in the third dot.



Figure 6.10: HOMO visualization in a decatriene molecule.

Switching Field [V/nm]	Dot 1	Dot 2	Dot 3
-4.0	0.1629	0.0864	0.4923
-3.0	0.1522	0.0954	0.4956
-2.0	0.1421	0.1044	0.4982
-1.0	0.1324	0.1134	0.5003
0	0.1230	0.1227	0.5020
+1.0	0.1138	0.1321	0.5034
+2.0	0.1048	0.1418	0.5043
+3.0	0.0959	0.1520	0.5048
+4.0	0.0870	0.1628	0.5048

Table 6.5: Oxidized decatriene molecule : dot charges (mulliken) as function of the switching field.



Figure 6.11: Oxidized decatriene: dot charges as function of the switching field.

In Fig. 6.13 and in table 6.6 the results of the simulation are shown with the write-in system. As already explained in Chapter 3, the two ideal charge points are placed at the distance *d* equal to the distance between the two active dots of the molecule. All the possible values between 0 and 1 for the two charge points were taken into consideration so that the total charge of the driver is equal to 1. Again, the graph is symmetrical and a better switch is obtained than in the case of the switching field. This is due to the fact that, unlike the electric field which is directed parallel to the axis between the two dots, the charge points create an electric field in all directions, ??as shown in figure 22??, this means that there will be a component of this electric field that moves the charge from the third dots to one of the two active dots.



Figure 6.12: Oxidized decatriene with point charges configuration.

Driver 1	Driver 2	Dot 1	Dot 2	Dot 3
0.0	1.0	0.1050	0.8035	0.0154
0.1	0.9	0.1024	0.8054	0.0159
0.2	0.8	0.1855	0.1762	0.4712
0.3	0.7	0.1840	0.1766	0.4720
0.4	0.6	0.1827	0.1771	0.4723
0.5	0.5	0.1815	0.1778	0.4723
0.6	0.4	0.1805	0.1786	0.4718
0.7	0.3	0.1797	0.1796	0.4710
0.8	0.2	0.1791	0.1807	0.4698
0.9	0.1	0.8068	0.0980	0.0143
1.0	0.0	0.8046	0.1003	0.0136

6 – Charge distribution

Table 6.6: Oxidized decatriene molecule : dot charges as function of the driver polarization.



Figure 6.13: Oxidized decatriene: dot charges as function of the polarization of the driver.

6.2 The bis-ferrocene molecule

6.2.1 Neutral molecule

What has been done up to now for ideal but not synthesizable molecules such as diallyl butane and decatriene will be repeated now for the bis-ferrocene molecule. This molecule, as already mentioned, has already been synthesized by the PoliTo in collaboration with the University of Bologna and it is the most promising molecule for QCA technology.

Also in this case, a first optimization simulation of the geometry of the neutral molecule was performed, in this case with the base 6,31g (d, p), which is a basis with polarization functions on both heavy and light atoms. Unlike the two previous molecules, which had only 26 atoms between carbon and hydrogen, this molecule in its version without thiol has 72 atoms, including two iron atoms and one of nitrogen. This fact causes that a single optimization simulation to last from 8 to 12 days, depending on whether it is neutral or oxidized. As can be seen from table 6.7 and Fig. 6.14, in its neutral form, this molecule behaves exactly like the ideal molecules. The switch is almost perfectly symmetrical and the charge on the third dot (carbazole) is almost 0. Since even in this case the switch is not very strong we need to oxidize the molecule.

Switching Field [V/nm]	Dot 1	Dot 2	Dot 3
-5.0	0.0676	-0.0702	0.0026
-4.0	0.0540	-0.0561	0.0021
-3.0	0.0403	-0.0421	0.0018
-2.0	0.0267	-0.0282	0.0015
-1.0	0.0144	-0.0130	0.0014
0	0	0	0.0013
+1.0	-0.0144	0.0130	0.0014
+2.0	-0.0282	0.0266	0.0015
+3.0	-0.0420	0.0403	0.0018
+4.0	-0.0546	0.0546	0.0021
+5.0	-0.0686	0.0686	0.0026

Table 6.7: Neutral bis-ferrocene molecule : dot charges (mulliken) as function of the switching field.



Figure 6.14: Neutral bis-ferrocene: dot charges as function of the switching field.

6.2.2 Oxidized molecule

For the bis-ferrocene, like for the decatriene, the extra charge obtained by oxidizing the molecule is not equally distributed between the two active dots as we hoped. Indeed, in this case, the extra charge is almost completely distributed on just one of the two active dots, practically polarizing the molecule. This behavior by the simulator causes the molecule to already codify a logical state to equilibrium and an electric field of 5 V/nm is required to change the state to the molecule.

	Dot 1	Dot 2	Dot 3
neutral	0	0	0.0013
oxidized	0.0163	0.9154	0.0683

Table 6.8: Comparison between neutral and oxidized bis-ferrocene molecule : dot charges (mulliken) at the equilibrium.



Figure 6.15: Oxidized bis-ferrocene: dot charges as function of the switching field.

Switching Field [V/nm]	Dot 1	Dot 2	Dot 3
-6.0	1.0261	-0.0610	0.0349
-5.0	1.0120	-0.0485	0.0365
-4.0	0.0820	0.8449	0.0731
-3.0	0.0677	0.8620	0.0704
-2.0	0.0536	0.8777	0.0687
-1.0	0.0397	0.8926	0.0677
0	0.0163	0.9154	0.0683
+1.0	0.0120	0.9210	0.0670
+2.0	-0.0018	0.9348	0.0670
+3.0	-0.0157	0.9486	0.0672
+4.0	-0.0298	0.9625	0.0673
+5.0	-0.0441	0.9772	0.0670

Table 6.9: Oxidized bis-ferrocene molecule : dot charges (mulliken) as function of the switching field.

Author in [35] did the same simulation with gaussian09 but with the base LANL2DZ. The results of this simulation are shown in the figure. To compare the results correctly, it would have been necessary to re-propose the exact same simulation on gamess, but this base is not present in its basis-set. It was then tried to manually build this base, however with this base the SCF did not converge.



Figure 6.16: Oxidized bis-ferrocene: dot charges as function of the switching field in gaussian09 [35].

The behaviour of the molecule in presence of two point charges is shown in Fig. 6.17. Since the molecule switches with a switching field of 5 V/nm, in this case the molecule can't switch. This is due to the fact that two point charges creae a potential difference of ± 0.5 V.



Figure 6.17: Oxidized bis-ferrocene: dot charges as function of the driver's polarization.

Driver 1	Driver 2	Dot 1	Dot 2	Dot 3
0.0	1.0	0.0287	0.8940	0.0773
0.1	0.9	0.0276	0.8954	0.0770
0.2	0.8	0.0264	0.8969	0.0767
0.3	0.7	0.0253	0.8983	0.0764
0.4	0.6	0.0241	0.8998	0.0761
0.5	0.5	0.0230	0.9012	0.0758
0.6	0.4	0.0218	0.9026	0.0755
0.7	0.3	0.0207	0.9041	0.0753
0.8	0.2	0.0195	0.9055	0.0750
0.9	0.1	0.0172	0.9083	0.0744
1.0	0.0	0.0172	0.9083	0.0744

6.3 – Clocked molecule

Table 6.10: Oxidized decatriene molecule : dot charges as function of the driver polarization.

6.3 Clocked molecule

The clock field is used to move the charge present in the third dot in the two active dots to make the switch more efficient; on the contrary for the relax phase where an electric field of opposite sign is needed. In this case the clock field has a positive sign, but this obviously depends on the orientation of the molecule in the space, which changes from software to software.



Figure 6.18: Oxidized bis-ferrocene: dot charges as function of the clock field.

Clock Field [V/nm]	Dot 1	Dot 2	Dot 3
-5.0	-0.0237	0.8214	0.2023
-4.0	-0.0142	0.8456	0.1686
-3.0	-0.0043	0.8637	0.1407
-2.0	0.0057	0.8793	0.1151
-1.0	0.0157	0.8935	0.0907
0	0.0163	0.9154	0.0683
+1.0	0.0363	0.9197	0.0439
+2.0	0.0469	0.9322	0.0210
+3.0	0.0572	0.9443	-0.0015
+4.0	0.0675	0.9563	-0.0238
+5.0	0.0778	0.9680	-0.0459

Table 6.11: Oxidized bis-ferrocene molecule : dot charges (mulliken) as function of the clock field.

The behaviour of the charge in the three dots are shown in Fig. 6.18. As we can notice, the value of 5 V/nm is not sufficient to move the charge from the two active dots to the third one. In particular, in this case a clock field of 10 V/nm is needed.



Figure 6.19: Bis-ferrocene oxidized scheme: at the equilibrium (a), with a clock field of 10 V/nm (b).

6.4 V_{out}

Authors in [35], [39] defined the *output voltage* V_{out} : this parameter is calculated as a difference voltage $V_2 - V_1$ imagining to place a receiver at a distance equal to the distance between the two active dots, which is for the diallyl butane 0.7 nm, as shown in the Fig. 6.20. The algorithm for this calculation was developed in MATLAB and says that for every charges in a molecule $\{q_1, q_2, ..., q_N\}$ and their position in the space $\{P_1, P_2, ..., P_N\}$, the electric potential at distance d is equal to:



Figure 6.20: Molecule - receiver scheme.

$$V\left(\vec{d}\right) = \frac{1}{4\pi\varepsilon_0} \left(\frac{q_1}{|P_1 - \vec{d}|} + \dots + \frac{q_N}{|P_2 - \vec{d}|} \right)$$
(6.1)

In the case shown in Fig. 6.20, if we consider the active dots as point charges, we have:

$$V_1 = \frac{1}{4\pi\varepsilon_0 \cdot d} \left(q_1 + \frac{q_2}{\sqrt{2}} \right) \tag{6.2}$$

$$V_2 = \frac{1}{4\pi\varepsilon_0 \cdot d} \left(\frac{q_1}{\sqrt{2}} + q_1\right) \tag{6.3}$$



Figure 6.21: V_{out} in function of the switching field at distance of 1 nm and 0.5 nm.

The V_{out} characteristic in function of the switching field is shown in Fig. 6.21. The ideal receiver is set at a distance of 1 nm and 0.5 nm. In these cases, the V_{out} is included between $\pm 0.5 \text{ V/nm}$ and $\pm 1.5 \text{ V/nm}$. The sign of V_{out} changes for a switching field of 4.5 V/nm, as for the mulliken charges. In particular we don't have potential for a switching field of 4.5 V/nm since the simulation doesn't converge.

Chapter 7

Conclusions and future study

In conclusion, GAMESS is a good software for ab-initio simulations, also if it has some shortcomings. In particular, the absence of a default action for the point charges is really disabling for future works. In fact, in this work we used some tricks to simulate them and this produced some errors in the evaluation of the system energy.

Future studies could be to understand how GAMESS handles the oxidation of a molecule and the differences with gaussian09, to look for a new basis that can reproduce the simmetry of the bis-ferrocene molecule, as for gaussian09. It could be interesting to study the molecules in a dynamic point of wiev (introducing time also during the switch), to study a complete QCA cell and to see how the molecules interacts with each other and finally analyzing a QCA wire and its consumption in terms of power and energy.

After this, some logic component can be studied and analyzed.
Appendix A

Mulliken charge comparison

Atom	Gaussian09	Gamess	Error
C1	-0.092065	-0.096216	0.004151
C2	-0.092065	-0.096216	0.004151
C3	-0.098441	-0.102622	0.004181
C4	-0.098441	-0.102622	0.004181
H5	0.056792	0.055832	-0.00096
H6	0.056784	0.055831	-0.000953
H7	0.056792	0.055832	-0.00096
H8	0.056784	0.055831	-0.000953
H9	0.052050	0.052632	0000582
H10	0.052048	0.052632	0.000584
H11	0.052050	0.052632	0.000582
H12	0.052048	0.052632	0.000584
C13	0.000463	0.001109	0.000646
C14	-0.125606	-0.123333	-0.002273
C15	-0.125615	-0.123332	-0.002283
H16	0.053773	0.056583	0.00281
H17	0.058023	0.057151	-0.000872
H18	0.053769	0.056582	0.002813
H19	0.058026	0.057151	-0.000875
C29	0.000463	0.001109	0.000646
C21	-0.125606	-0.123333	-0.002273
C22	-0.125615	-0.123332	-0.002283
H23	.053773	0.056583	0.00281
H24	0.058023	0.057151	-0.000872
H25	0.053769	0.056582	0.002813
H26	0.058026	0.057151	-0.000875

Table A.1: Mulliken charge comparison gfor a Diallyl butane molecule at the equilibrium.

Atom	Gaussian09	Gamess	Error
C1	-0.095807	-0.095791	0.000016
C2	-0.096626	-0.096643	0.000017
C3	-0.102682	-0.102695	0.000013
C4	-0.102557	-0.102545	-0.000012
H1	0.053031	0.053162	0.000131
H2	0.053030	0.053161	0.000131
H3	0.058627	0.058496	-0.000131
H4	0.058626	0.058494	0.000132
H5	0.052926	0.052801	-0.000125
H6	0.052927	0.052802	-0.000125
H7	0.052337	0.052462	0.000125
H8	0.052335	0.052460	0.000125
C5	-0.000895	-0.000932	0.000037
C6	-0.097411	-0.097333	-0.000078
C7	-0.097408	-0.097330	-0.000078
H9	0.070150	0.070154	0.000004
H10	0.073413	0.073538	0.000125
H11	0.070147	0.070151	0.000004
H12	0.073412	0.073538	0.000126
C8	0.003113	0.003151	0.000038
C9	-0.149272	-0.149350	0.000078
C10	-0.149269	-0.149347	0.000078
H13	0.043031	0.043028	-0.000003
H14	0.040894	0.040769	0.000125
H15	0.043031	0.043028	-0.000003
H16	0.040894	0.040769	0.000125

Table A.2: Mulliken charge comparison with an external electric field of $0.5\,\mathrm{V/nm}$

Appendix B

Electric potential comparison with an external electric field of 0.5 V/nm.

Atom	Gaussian09	Gamess	Error
C1	-14.540419	-14.540452	0.00033
C2	-14.532074	-14.532042	0.000032
C3	-14.545358	-14.545338	0.00002
C4	-14.522107	-14.522128	0.000021
H1	-1.179047	-1.178929	0.000118
H2	-1.179049	-1.178931	0.000118
H3	-1.156583	-1.156700	0.000117
H4	-1.156584	-1.156701	0.000117
H5	-1.176205	-1.176340	0.000135
H6	-1.176205	-1.176340	0.000135
H7	-1.166866	-1.166731	0.000135
H8	-1.166865	-1.166730	0.000135
C5	-14.525094	-14.525053	0.000041
C6	-14.540743	-14.540626	0.000117
C7	-14.540745	-14.540627	0.000118
H9	-1.118863	-1.118802	0.000061
H10	-1.115423	-1.115252	0.000171
H11	-1.118865	-1.118804	0.000061
H12	-1.115423	-1.115252	0.000171
C8	-14.573726	-14.573768	0.000042
C9	-14.620710	-14.620828	0.000118
C10	-14.620710	-14.620828	0.000118
H13	-1.181810	-1.181870	0.00006
H14	-1.188241	-1.188412	0.000171
H15	-1.181807	-1.181867	0.00006
H16	-1.188239	96-1.188410	0.000171

Appendix C

Input file for Gamess

C.1 Diallyl butane input files

C.1.1 Neutral at the equilibrium

\$BASIS GBASIS=STO NGAUSS=3 \$END \$CONTRL SCFTYP=UHF RUNTYP=OPTIMIZE COORD=ZMT \$END \$STATPT OPTTOL=0.0001 NSTEP=20 \$END

\$DA	ТA					
Tit	le					
C1						
\mathbf{C}						
\mathbf{C}	1	B1				
\mathbf{C}	1	B2	2	A1		
\mathbf{C}	2	B3	1	A2	3	D1
Η	3	B4	1	A3	2	D2
Η	3	B5	1	A4	2	D3
Η	4	B6	2	A5	1	D4
Η	4	B7	2	A6	1	D5
Η	1	B8	2	A7	4	D6
Η	1	B9	2	A8	4	D7
Η	2	B10	1	A9	3	D8
Η	2	B11	1	A10	3	D9
\mathbf{C}	4	B12	2	A11	1	D10
\mathbf{C}	13	B13	4	A12	2	D11
\mathbf{C}	13	B14	4	A13	2	D12
Η	14	B15	13	A14	4	D13
Η	14	B16	13	A15	4	D14
Η	15	B17	13	A16	4	D15
Η	15	B18	13	A17	4	D16
\mathbf{C}	3	B19	1	A18	2	D17
\mathbf{C}	20	B20	3	A19	1	D18
\mathbf{C}	20	B21	3	A20	1	D19
Η	21	B22	20	A21	3	D20
Η	21	B23	20	A22	3	D21
Η	22	B24	20	A23	3	D22
Η	22	B25	20	A24	3	D23
1	31	1.54219718				
I	32	1.54715028				

B3	1.54715028
B4	1.08772095
B5	1.08771964
B6	1.08772095
B7	1.08771964
B8	1.08843697
B9	1.08843680
B10	1.08843697
B11	1.08843680
B12	1.53585525
B13	1.37770630
B14	1.37772504
B15	1.08001478
B16	1.08038854
B17	1.08001487
B18	1.08038819
B19	1.53585525
B20	1.37770630
B21	1.37772504
B22	1.08001478
B23	1.08038854
B24	1.08001487
B25	1.08038819
A1	112.18495797
A2	112.18495797
A3	109.25301425
A4	109.25298930
A5	109.25301425
A6	109.25298930
A7	109.48180141
A8	109.48122967
A9	109.48180141
A10	109.48122967
A11	111.24431029
A12	118.62285258
A13	118.61999817
A14	121.66720440
A15	121.41761015
A16	121.66666901
A17	121.41779191
A18	111.24431029
A19	118.62285258
A20	118.61999817
A21	121.66720440
A22	121.41761015
A23	121.66666901
A24	121.41779191
DI	
D2 D3	-58.69919355
D3	58.71180363
D4 D5	-58.69919355
D5 D6	58.71180363
D6	-58.61861781
	58.61814219
D8	-58.01801781
D9	58.01814219
D10	-179.99339158
	-88.92015591
D12 D12	88.87856924
D13 D14	-2.31385877
D14	178.35611866
D15	2.31545807
D10	-178.35545484

D17	-179.99339158
D18	-88.92015591
D19	88.87856924
D20	-2.31385877
D21	178.35611866
D22	2.31545807
D23	-178.35545484

C.1.2 Neutral with the switching field

```
$BASIS GBASIS=STO NGAUSS=3 $END
$CONTRL SCFTYP=UHF RUNTYP=FFIELD COORD=ZMT MAXIT =200 ICHARG=1
MULT=2 $END
 $SCF CONV=1d-5 $END
$STATPT OPTTOL=0.0001 NSTEP=20 $END
EFIELD EVEC(1) = -0.006, 0, 0 EVD
$DATA
Title
C1
\mathbf{C}
\mathbf{C}
         1
                  Β1
\mathbf{C}
                  B2
                            \mathbf{2}
                                     A1
         1
\mathbf{C}
         2
                  B3
                            1
                                     A2
                                               3
                                                        D1
Н
         3
                  B4
                                     A3
                                               \mathbf{2}
                                                        D2
                            1
Η
         3
                  B5
                            1
                                     A4
                                               2
                                                        D3
Η
         4
                  B6
                            \mathbf{2}
                                     A5
                                               1
                                                        D4
Η
                  B7
                            \mathbf{2}
                                     A6
                                                        D5
         4
                                               1
Η
         1
                  B8
                            2
                                     A7
                                               4
                                                        D6
Η
         1
                  B9
                            \mathbf{2}
                                     A8
                                               4
                                                        D7
Н
         \mathbf{2}
                 B10
                            1
                                     A9
                                               3
                                                        D8
Η
         \mathbf{2}
                 B11
                            1
                                    A10
                                               3
                                                        D9
\mathbf{C}
         4
                 B12
                            \mathbf{2}
                                    A11
                                               1
                                                       D10
\mathbf{C}
       13
                                               \mathbf{2}
                 B13
                            4
                                    A12
                                                       D11
\mathbf{C}
       13
                 B14
                            4
                                    A13
                                               \mathbf{2}
                                                       D12
Η
       14
                 B15
                           13
                                    A14
                                                       D13
                                               4
                 B16
Н
       14
                           13
                                    A15
                                               4
                                                       D14
Η
       15
                 B17
                           13
                                    A16
                                               4
                                                       D15
Н
       15
                 B18
                                    A17
                                                       D16
                           13
                                               4
\mathbf{C}
        3
                 B19
                            1
                                    A18
                                               \mathbf{2}
                                                       D17
\mathbf{C}
       20
                 B20
                            3
                                    A19
                                               1
                                                       D18
\mathbf{C}
       20
                 B21
                                                       D19
                                    A 20
                            3
                                               1
Η
       21
                 B22
                           20
                                    A21
                                               3
                                                       D20
Н
       21
                 B23
                           20
                                    A22
                                               3
                                                       D21
Η
       22
                 B24
                           20
                                                       D22
                                    A23
                                               3
Η
       22
                 B25
                           20
                                    A24
                                               3
                                                       D23
   B1
                        1.54219718
   B2
                        1.54715028
   B3
                        1.54715028
   B4
                        1.08772095
                        1.08771964
   B5
                        1.08772095
   B6
   B7
                        1.08771964
   B8
                        1.08843697
   B9
                        1.08843680
   B10
                        1.08843697
                        1.08843680
   B11
   B12
                        1.53585525
                        1.37770630
   B13
```

B14	1.37772504
B15	1.08001478
B16	1.08038854
B17	1.08001487
B18	1.08038819
B19	1 53585525
B20	1.37770630
B20 B21	1.377700000 1.37772504
D21 D22	1.02001472
D22	1.00001470
B23	1.08038854
B24	1.08001487
B25	1.08038819
A1	112.18495797
A2	112.18495797
A3	109.25301425
A4	109.25298930
A5	109.25301425
A6	109.25298930
A7	109.48180141
A8	109.48122967
A9	109.48180141
A10	109 48122967
Δ11	111 24/31020
Δ12	118 62285258
A12 A12	118.02285258
A15	110.01999017
A14	121.00720440
A15	121.41761015
A16	121.66666901
A17	121.41779191
A18	111.24431029
A19	118.62285258
A20	118.61999817
A21	121.66720440
A22	121.41761015
A23	121.66666901
A24	121.41779191
D1	180.00000000
D2	-58.69919355
D3	58.71180363
D4	-58,69919355
D5	58 71180363
DG	59 61961791
D0 D7	-30.01001701
	50.01014219
D8	-38.01801/81
D9 D10	58.61814219
D10	-179.99339158
D11	-88.92015591
D12	88.87856924
D13	-2.31385877
D14	178.35611866
D15	2.31545807
D16	-178.35545484
D17	-179.99339158
D18	-88.92015591
D19	88.87856924
D20	-2.31385877
D20	178 35611866
D21	9 91545907
D22	2.31343807
D23	-118.35545484

C.1.3 Neutral with the driver

\$BASIS GBASIS=STO NGAUSS=3 \$END \$CONTRL SCFTYP=UHF RUNTYP=ENERGY COORD=CART UNITS=BOHR MAXIT =200 ICHARG=1 MULT=2 \$END \$STATPT OPTTOL=0.0001 NSTEP=20 \$END \$SYSTEM MWORDS=50 \$END

\$DATA Title C1 \mathbf{C} 6.0 1.3332368711-0.0002526451-0.5880556204 \mathbf{C} -1.3332368711-0.00025264510.5880556204 6.0 \mathbf{C} 6.03.4358660887 -0.00025264511.4434251503 \mathbf{C} -3.4358660887-0.0002526451-1.44342515036.0 Η 1.03.22279461071.65783972902.63941346773.2230477325-1.65856511912.6391490900Η 1.0Η 1.0-3.22279461071.6578397290-2.6394134677Η 1.0-3.2230477325-1.6585651191-2.6391490900Н 1.01.5533761233-1.6556908607-1.7887575775Η 1.0 1.55335140071.6551827736-1.7887654158Η 1.0-1.5533761233-1.65569086071.7887575775Η 1.0-1.55335140071.65518277361.7887654158 С 6.0-6.07178827630.0000593591-0.2287136680 \mathbf{C} 6.0-7.1866398826-2.28471924180.3325699278 \mathbf{C} 6.0 -7.18549078502.28517110120.3336581671 Η 1.0-6.2846283303-4.0578002243-0.1234590242Н 1.0-9.0163919699-2.38342872131.2328758142Η 1.0-6.28258368244.0579984242-0.1215847979Η 1.0-9.01518181022.38439341141.2340302771 \mathbf{C} 6.0 6.0717882763 0.00005935910.2287136680 \mathbf{C} 6.0 7.1866398826 -2.2847192418-0.3325699278 \mathbf{C} 6.07.1854907850 2.2851711012-0.3336581671Η 6.2846283303 -4.05780022430.12345902421.0Н 1.09.0163919699-2.3834287213-1.2328758142Н 1.06.28258368244.05799842420.12158479799.01518181022.3843934114-1.23403027711.0 Н \$END \$EFRAG POSITION=FIXED COORD=CART fragname=WATER1 W101 -7.1861 -14.372 -0.333 W1H2 -3.2131 5.376 -0.12W1H3 -3.2131 4.763 -0.23fragname=CO21 C1C1 7.1861 - 14.372 - 0.333 $\begin{array}{rrrr} -6.3131 & 5.376 & -0.12 \\ -6.2131 & 4.763 & -0.23 \end{array}$ C1O2C1O3 \$END \$WATER1 WATER1 as DR1 COORDINATES (BOHR) $-7.1861 \ -14.372 \ -0.333 \ 0.0 \ 0.0$ W101 W1H2 -3.2131 5.376 -0.12 0.0 0.0-3.2131 4.763 -0.23 0.0 0.0W1H3 STOP MONOPOLES W1O1 0.7 0.0 W1H2 0.0 0.0 W1H3 0.0 0.0 STOP REPULSIVE POTENTIAL W101

```
0, 0
  STOP
  $END
  CO21
Carbon \ as \ DR2
COORDINATES(BOHR)
\begin{array}{c} \text{COORDINATES(DARK)}\\ \text{C1C1} & 7.1861 & -14.372 & -0.333 & 0.0 & 0.0 \\ \text{C1O2} & -6.3131 & 5.376 & -0.12 & 0.0 & 0.0 \\ \text{C1O3} & -6.2131 & 4.763 & -0.23 & 0.0 & 0.0 \end{array}
 STOP
 MONOPOLES
C1C1 \ 0.3 \ 0.0
C1O2 \ 0.0 \ 0.0
C1O3 0.0 0.0
 STOP
  REPULSIVE POTENTIAL
C1C1
0, 0
  STOP
  $END
  $FRGRPL
  PAIR=WATER1 CO21
W101 C1C1 0 0
STOP
  $END
```

C.2 Decatriene input files

C.2.1 Oxidized at the equilibrium

\$BASIS GBASIS=N31 NGAUSS=6 \$END \$CONTRL SCFTYP=UHF RUNTYP=OPTIMIZE COORD=ZMT MAXIT=200 ICHARG=1 MULT=2 \$END \$SCF CONV=1d-5 \$END \$STATPT OPTTOL=0.0001 NSTEP=120 \$END

\$DA	ТА					
Tit	le					
C1						
\mathbf{C}						
С	1	B1				
\mathbf{C}	1	B2	2	A1		
\mathbf{C}	2	B3	1	A2	3	D1
\mathbf{C}	1	B4	3	A3	2	D2
\mathbf{C}	5	B5	1	A4	3	D3
\mathbf{C}	3	B6	1	A5	5	D4
\mathbf{C}	4	B7	2	A6	1	D5
\mathbf{C}	7	B8	3	A7	1	D6
\mathbf{C}	8	B9	4	A8	2	D7
Η	1	B10	5	A9	6	D8
Н	1	B11	5	A10	6	D9
Н	2	B12	1	A11	5	D10
Η	2	B13	1	A12	5	D11
Η	3	B14	1	A13	5	D12
Н	3	B15	1	A14	5	D13
Η	4	B16	2	A15	1	D14
Η	4	B17	2	A16	1	D15
Η	5	B18	1	A17	3	D16
Н	6	B19	5	A18	1	D17

H 7 H 8 H 9 H 9 H 10 H 10	 B20 B21 B22 B23 B24 B25 	3 4 7 7 8 8	A19 A20 A21 A22 A23 A24	$egin{array}{c} 1 \\ 2 \\ 3 \\ 3 \\ 4 \\ 4 \end{array}$	D18 D19 D20 D21 D22 D23
$\begin{array}{c} B1\\ B2\\ B3\\ B4\\ B5\\ B6\\ B7\\ B8\\ B9\\ B10\\ B11\\ B12\\ B13\\ B14\\ B15\\ B16\\ B17\\ B18\\ B19\\ B20\\ B21\\ B22\\ B23\\ B24\\ B25\\ A1\\ A2\\ A3\\ A4\\ A5\\ A6\\ A7\\ A8\\ A9\\ A10\\ A11\\ A12\\ A13\\ A44\\ A5\\ A6\\ A7\\ A8\\ A9\\ A10\\ A11\\ A12\\ A13\\ A14\\ A15\\ A16\\ A17\\ A18\\ A19\\ A20\\ A21\\ A22\\ A23\\ A24\\ D1\\ D2\\ D3\\ D4\\ D5\\ D6\\ D7\\ \end{array}$		$\begin{array}{c} 3.\\ 1.\\ 1.\\ 1.\\ 1.\\ 1.\\ 1.\\ 1.\\ 1.\\ 1.\\ 1$	010589 532279 532173 502549 341173 501633 501728 337494 3375370 098662 097832 093406 097403 0977572 096741 086605 086825 086851 086851 0868530 084245 084473 084245 084473 084373 491563 385888 686875766881 7388455766881 7388455766881 738845766881 7388457766881 7388457766881 7388457766881 7388457766881 7388457766881 7388457766881 7388457766881 7388457766881 7388457766687 7417686716571 380422 5060600 7963500 3785426 791436 5789422 4113266 5690866 704938 375324 394007 416988 578942 4113266 5690867 704938 375324 3839651 3839631 3839413 3839413	6352397759044588379501166979155103438200023266667915510346979033692438000000000000000000000000000000000000	

D8	-149.74946798
D9	-35.69924519
D10	65.04827083
D11	144.94742397
D12	-57.29573822
D13	59.14444985
D14	2.03583153
D15	118.06137418
D16	-88.60340061
D17	-179.09463140
D18	-57.25941520
D19	-57.33234676
D20	179.45913830
D21	-0.60143634
D22	179.43625357
D23	-0.63938009

Oxidized with the switching field C.2.2

```
$BASIS GBASIS=N31 NGAUSS=6 $END
$CONTRL SCFTYP=UHF RUNTYP=ENERGY COORD=CART MAXIT=200 ICHARG=1
MULT=2 $END
SCF CONV=1d-5 END
EFIELD EVEC(1) = 0.008, 0, 0 \
¢DATA
```

\$I	J	P	ſ	T	7	ſ
		•		1		

-1 i t	ie			
C1				
\mathbf{C}	6.0	1.4077421114	-0.6099878945	-0.7211454165
\mathbf{C}	6.0	-1.4103554140	-0.6054306245	0.7234133917
С	6.0	2.6147211034	-0.3675706950	0.2667154670
\mathbf{C}	6.0	-2.6159800022	-0.3609911734	-0.2655294792
\mathbf{C}	6.0	0.6343451538	-1.7935760604	-0.3006175310
С	6.0	-0.6397955938	-1.7915315546	0.3047978910
\mathbf{C}	6.0	3.4424451357	0.7943232453	-0.2045827673
С	6.0	-3.4419129741	0.8025438757	0.2046580313
\mathbf{C}	6.0	3.5739917360	1.9350905960	0.4628050034
\mathbf{C}	6.0	-3.5707754269	1.9433248731	-0.4633542805
Η	1.0	1.8265231880	-0.7980097485	-1.7046600307
Η	1.0	0.8075305812	0.2852756252	-0.7722006204
Η	1.0	-1.8304818069	-0.7911823989	1.7067517013
Η	1.0	-0.8078804400	0.2882086579	0.7740274032
Η	1.0	2.2345003628	-0.1795078667	1.2636867785
Η	1.0	3.2186049642	-1.2681063244	0.3080741334
Η	1.0	-2.2345491097	-0.1741985975	-1.2622556257
Η	1.0	-3.2217084718	-1.2602210851	-0.3069751625
Η	1.0	1.0998560386	-2.7544354789	-0.4425950370
Η	1.0	-1.1076917379	-2.7510434523	0.4481717254
Η	1.0	3.9567118324	0.6570273353	-1.1401541884
Η	1.0	-3.9574865198	0.6665620630	1.1396979959
Η	1.0	4.1889445163	2.7315273049	0.0926872726
Η	1.0	3.0869574988	2.1082293798	1.4050135547
Η	1.0	-4.1847549979	2.7409927201	-0.0940955756
Η	1.0	-3.0824886896	2.1151702588	-1.4051841977

\$END

Oxidized with the driver C.2.3

\$BASIS GBASIS=N31 NGAUSS=6 \$END

\$CO MU \$SC \$SYS	NTRL S LT=2 \$ F CONV STEM M	SCFTYP=UHF RUNT SEND V=1d-5 \$END MVORDS=50 \$END	YP=ENERGY COORD	ECART MAXIT =200	ICHARG=1
\$DA	ГА				
Title	е				
CI	6 0	1 4077491114	0 6000878045	0 7011454165	
C	6.0	1.4077421114 1.4102554140	-0.6099878945 0.6054206245	-0.7211404100 0.7024122017	
c	6.0	-1.4103334140 2 6147211034	-0.3675706950	0.7254155517 0.2667154670	
č	6.0	-2.6159800022	-0.3609911734	-0.2655294792	
C	6.0	0.6343451538	-1.7935760604	-0.3006175310	
\mathbf{C}	6.0	-0.6397955938	-1.7915315546	0.3047978910	
\mathbf{C}	6.0	3.4424451357	0.7943232453	-0.2045827673	
С	6.0	-3.4419129741	0.8025438757	0.2046580313	
С	6.0	3.5739917360	1.9350905960	0.4628050034	
C	6.0	-3.5707754269	1.9433248731	-0.4633542805	
н u	1.0	1.8265231880	-0.7980097485	-1.7040000307 0.7722006204	
н	1.0 1.0	-1.8304818069	-0.7911823989	-0.7722000204 1 7067517013	
Н	1.0	-0.8078804400	0.2882086579	0.7740274032	
Н	1.0	2.2345003628	-0.1795078667	1.2636867785	
Η	1.0	3.2186049642	-1.2681063244	0.3080741334	
Η	1.0	-2.2345491097	-0.1741985975	-1.2622556257	
Η	1.0	-3.2217084718	-1.2602210851	-0.3069751625	
Н	1.0	1.0998560386	-2.7544354789	-0.4425950370	
H	1.0	-1.1076917379	-2.7510434523	0.4481717254	
H H	1.0	3.9567118324	0.6570273353	-1.1401541884	
п Н	1.0 1.0	-3.9574605196 4.1880445163	0.0005020050 2.7315273049	1.1390979939	
Н	1.0	3.0869574988	2.1082293798	1.4050135547	
Н	1.0	-4.1847549979	2.7409927201	-0.0940955756	
Н	1.0	-3.0824886896	2.1151702588	-1.4051841977	
\$EN	D				
\$EFF	RAG				
POS	ITION=	FIXED COORD=CA	RT		
frag	name=	=WATER1			
WIOI	3.57	$(-5.21 \ 0.46$	1.1.0		
W1H2	-10	$-2131 \ 30.570 \ -0$).12		
frag	name=	-CO21	7.20		
C1C1	-3.5	57 - 5.21 - 0.46			
C1O2	-10	.3131 30.376 -0	0.12		
C1O3	-10	.2131 30.763 -0	0.23		
\$ENI	D				
\$WA	TER1				
WATE	R1 as	DR1			
W1O1		ES(BOHR)			
W101 W1H9	3.37 _10	-5.21 0.40 0.0			
W1H3	-10	$.2131 \ 30.763 \ -0$	0.12 0.0 0.0		
STO	P				
MON	OPOLE	S			
W1O1	0.3	0.0			
W1H2	0.0	0.0			
W1H3	0.0	0.0			
STO	P III CH T				
REP	ULSIVI	E POTENTIAL			
WIOI					
STO	Р				
\$EN	D				

```
$CO21
Carbon as DR2
COORDINATES(BOHR)
\begin{array}{ccccccc} \text{C0Cl} & -3.57 & -5.21 & -0.46 & 0.0 & 0.0 \\ \text{C1C2} & -10.3131 & 30.376 & -0.12 & 0.0 & 0.0 \\ \text{C1O3} & -10.2131 & 30.763 & -0.23 & 0.0 & 0.0 \\ \end{array}
  STOP
  MONOPOLES
C1C1 0.7 0.0
C1O2 0.0 0.0
C1O3 0.0 0.0
 STOP
  REPULSIVE POTENTIAL
C1C1
0, 0
  STOP
  $END
  $FRGRPL
  PAIR=WATER1 CO21
W1O1 C1C1 0 0
 STOP
  $END
```

C.3 Bis-ferrocene input file

C.3.1 Neutral at the equilibrium

\$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 \$END \$CONTRL SCFTYP=UHF RUNTYP=OPTIMIZE COORD=ZMT MAXIT=200 \$END \$STATPT OPTTOL=0.0001 NSTEP=120 \$END \$SYSTEM MWORDS=125 \$END

\$DA	ГА					
Τit	le					
C1						
С						
\mathbf{C}	1	B1				
С	2	B2	1	A1		
\mathbf{C}	3	B3	2	A2	1	D1
\mathbf{C}	4	B4	3	A3	2	D2
Fe	1	B5	5	A4	4	D3
\mathbf{C}	6	B6	1	A5	5	D4
\mathbf{C}	7	B7	6	A6	1	D5
\mathbf{C}	8	B8	7	A7	6	D6
\mathbf{C}	9	B9	8	A8	7	D7
\mathbf{C}	10	B10	9	A9	8	D8
\mathbf{C}	8	B11	7	A10	6	D9
\mathbf{C}	12	B12	8	A11	7	D10
\mathbf{C}	12	B13	8	A12	7	D11
\mathbf{C}	14	B14	12	A13	8	D12
\mathbf{C}	15	B15	14	A14	12	D13
\mathbf{C}	16	B16	15	A15	14	D14
\mathbf{C}	17	B17	16	A16	15	D15
\mathbf{C}	18	B18	17	A17	16	D16
\mathbf{C}	16	B19	15	A18	14	D17
\mathbf{C}	20	B20	16	A19	15	D18
Ν	17	B21	16	A20	15	D19
\mathbf{C}	20	B22	16	A21	15	D20
\mathbf{C}	23	B23	20	A22	16	D21
С	24	B24	23	A23	20	D22

С	25	B25	24	A24	23	D23
C	24	B26	23	A25	20	D24
С	27	B27	24	A26	23	D25
\mathbf{C}	27	B28	24	A27	23	D26
\mathbf{C}	29	B29	27	A28	24	D27
\mathbf{C}	30	B30	29	A29	27	D28
\mathbf{C}	31	B31	30	A30	29	D29
\mathbf{C}	32	B32	31	A31	30	D30
Fe	32	B33	31	A32	30	D31
\mathbf{C}	34	B34	32	A33	31	D32
\mathbf{C}	35	B35	34	A34	32	D33
C	36	B36	35	A35	34	D34
C	37	B37	36	A36	35	D35
C	35	B38	34	A37	32	D36
H	12	B39 D40	8	A38	1	D37
п	13	B40 D41	12	A39	8	D38
п u	13	D41 D49	12	A40	0	D39
п ц	13	D42 B43	12	A41 A42	24	D40
н	28	B43 B44	21	A42 A43	24	D41 D42
н	28	B45	27	A44	24	D42 D43
н	36	B46	35	A45	34	D44
Н	37	B47	36	A46	35	D45
Н	38	B48	37	A47	36	D46
Η	39	B49	35	A48	34	D47
Η	35	B50	34	A49	32	D48
Η	30	B51	29	A50	27	D49
Η	33	B52	32	A51	31	D50
Н	32	B53	31	A52	30	D51
Н	31	B54	30	A53	29	D52
H	9	B55	8	A54	7	D53
H H	10	B50 D57	10	A55	8	D54 DF5
п п	11	DST	10	A50	9	D55
н	1	B59	5	A58	4	D50
н	5	B60	4	A59	3	D58
Н	4	B61	3	A60	2	D59
Н	3	B62	2	A61	1	D60
Н	2	B63	1	A62	5	D61
Н	15	B64	14	A63	12	D62
Η	19	B65	18	A64	17	D63
Η	18	B66	17	A65	16	D64
Η	22	B67	17	A66	16	D65
Η	26	B68	25	A67	24	D66
Η	25	B69	24	A68	23	D67
Η	23	B70	20	A69	16	D68
Н	27	B71	24	A70	23	D69
B	1		1.4	243214	10	
В	2		1.4	238570)4 75	
B	3 1		1.4	240301	17	
B	- 5		2.0	845985	51	
В	6		2.0	835375	53	
В	7		1.4	287348	35	
В	8		1.4	288827	78	
В	9		1.4	226889	92	
B	10		1.4	230074	10	
B	11		1.5	0117223	38	
B	12		1.5	386148	52 16	
р В	14		1.0	20112210	10 13	
B	15		1.3	3949403	33	

B16	1.41568887
B17	1 30071610
D17 D19	1.39071010
D10	1.30073239
B19	1.44907700
B20	1.41569620
B21	1.38557527
B22	1.39493330
B23	1.39188889
B24	1 40609326
B25	1.38875051
D20 D26	1.50075551
D20	1.52870034
B27	1.53861066
B28	1.51173418
B29	1.42888093
B30	1.42269326
B31	1.42301251
B32	1.42477709
B33	2 07800077
D00 D04	2.01003011
D04	2.08030370
B35	1.42417847
B36	1.42431745
B37	1.42386359
B38	1.42384347
B39	1.09423424
B40	1.09161746
B41	1 09116815
D41 D49	1.00120227
D42 D42	1.09109207
B43	1.09101847
B44	1.09189298
B45	1.09116870
B46	1.07827234
B47	1.07838995
B48	1.07857541
B49	1 07838288
B50	1.07837063
D50	1.07800004
D51 D50	1.07899004
B52	1.07807904
B53	1.07848088
B54	1.07839866
B55	1.07898980
B56	1.07839841
B57	1.07847960
B58	1.07808027
B50	1.07827208
D60	1.07826004
D00	1.07030994
B01	1.07838285
B62	1.07857378
B63	1.07838864
B64	1.08332259
B65	1.08458937
B66	1.08359593
B67	1.00464467
B68	1 08359663
DCO	1.000000000
D09 D70	1.00400000
D/U D71	1.08332472
B71	1.09423262
Al	107.98832088
A2	108.00649939
A3	107.99419900
A4	70.09766200
A5	157.37258679
AG	70 47889197
Δ7	106 097/0/90
L1	100.34143439

A8	108.75166886
A9	107.81760282
A10	128.03571982
A11	112.99631494
A12	112.09494024
A13	121.69854015
A14	120.02011915
A15	119.53931805
A16	121,29465590
A17	117 72001719
A18	133 77882278
A 19	106 67933896
A 20	108 61749139
A 21	133 77047148
A21 A22	100.00020208
A 22	120.02032303 110.02714625
A20 A94	119.02714023
A24	122.39844197
A25	121.69533968
A26	110.61033671
A27	112.09462869
A28	124.92639957
A29	108.75131922
A30	107.81771939
A31	107.89077475
A32	70.03962299
A33	108.52747933
A34	69.96906657
A35	108.01020745
A36	107.98846118
A37	70.07751394
A38	106.94805730
A39	111.10363706
A40	110.22845254
A41	111.12988947
A42	111.10263583
A43	111.13050995
A44	110.22817317
A45	125 99759718
A46	125.99017943
A47	125.96907648
Δ / 8	125.95907040
A40	124 50850256
A49 A50	124.39830230
A50 A 51	125.30226934
A51 A59	120.42000194
A52	120.14/80303
A53	126.06430118
A54	125.30079771
A55	126.06287414
A56	126.14718431
A57	124.42996574
A58	125.99739697
A59	125.99241800
A60	126.04151755
A61	125.96936989
A62	125.98903171
A63	119.93178095
A64	118.77656718
A65	121.51664779
A66	125.29231743
A67	120.76122417
A68	118.82449094
A69	120.04155665
A70	106.47738774

D1	0 02620520
DI	-0.02030339
D2	0.05761447
D3	-60.01192633
D4	-48.53390094
D5	163 88864144
D0 D0	-105.88804144
D6	59.55195094
D7	0.18507789
D8	-0.27879090
DO	194 18200408
D9	-124.18300498
D10	22.08131956
D11	-103.72104496
D12	$41 \ 48335946$
D12	177 01112020
D13	177.91113020
D14	0.01685779
D15	0.07690624
D16	-0.04136208
D17	170 26281550
DII	179.30281330
D18	-179.40764070
D19	179.61525020
D20	0.00403170
D91	170 26810207
D21	-179.30810327
D22	0.14227371
D23	-0.18335598
D24	-177.90806203
D25	85 59612133
D26	41 40644779
D20	-41.49044772
D27	-71.91729979
D28	176.23869852
D29	0.27802356
D30	-0.26216300
D91	50 08810612
D31	39.98819012
D32	121.83857251
D33	-118.04554140
D34	-60.06362139
D35	0.01630894
D26	102 12785780
D30	120.00010076
D37	139.93912076
D38	55.95915065
D39	176.25827364
D40	-64.81624883
D41	177 44421629
D41 D49	C1 70020705
D42	-61.78038705
D43	57.14602374
D44	118.86515404
D45	178.88926613
D46	$179 \ 32636937$
D47	110 12026020
D41	-119.12020929
D48	2.51618449
D49	-4.03547809
D50	178.67415052
D51	178 53923948
D52	178 66885262
D52	178.00885202
D53	-179.54332556
D54	-178.66713159
D55	-178.53703408
D56	75.31266505
D57	-178 88030134
	170.0000000000
D98	-178.91097519
D59	-179.06827109
D60	-179.31952094
D61	-178.89008234
D62	-3.04587300
D63	179 65804780
L00	113.00004100

D64	179.48068500
D65	-179.11727176
D66	179.61536151
D67	179.56466154
D68	-0.32400050
D69	-158.12256424

C.3.2 Oxidized at the equilibrium

\$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 \$END \$CONTRL SCFTYP=UHF RUNTYP=OPTIMIZE COORD=ZMT MAXIT=200 ICHARG=1 MULT=2 \$END \$STATPT OPTTOL=0.0001 NSTEP=300 \$END \$SCF CONV=1d-5 \$END \$SYSTEM MWORDS=125 \$END

\$DATA Title C1 \mathbf{C} Č B1 1 \mathbf{C} 2 B21 A1 \mathbf{C} B3 $\mathbf{2}$ D13 A2 1 \mathbf{C} 4 B43 A32 D2Fe B5D3 1 5A4 $\mathbf{4}$ \mathbf{C} 6 B6 1 A55D4 \mathbf{C} 7B7 $\mathbf{6}$ A61 D5 \mathbf{C} 8 B87 A7 $\mathbf{6}$ D6 $\tilde{\mathbf{C}}$ B979 8 A8D7 \mathbf{C} 10B109 A98 D8 \mathbf{C} B11 $\overline{7}$ A10 6 D9 8 \mathbf{C} 12B128 A117D10 \mathbf{C} 12B138 A12 7D11 \mathbf{C} 1214B14 A138 D12 \mathbf{C} 15B1514A1412D13 \mathbf{C} 16B16 15A15 D14 14 $\tilde{\mathbf{C}}$ 17 B17 16A1615D15 \mathbf{C} 18B1817A17 16D16 \mathbf{C} B19 D17 1615A18 14 \mathbf{C} 20B2016A19 15D18Ν 17B2116A20 15D19 C C B22 D20 2016 A21 1523B2320A2216D21 \mathbf{C} B242324A23 20D22 $\tilde{\mathbf{C}}$ 25B2524D23A2423 \mathbf{C} 24B2623A2520D24 \mathbf{C} 27B2724A26 23D25 \mathbf{C} 27B2824A2723D26 \mathbf{C} 29B2927A28 D2724 \mathbf{C} 30**B**30 29 A29 27 D28 \mathbf{C} 31B3130A3029D29 \mathbf{C} 32B3231A31 30 D30 Fe B33D31 3231A3230 \mathbf{C} 34B3432A33 31D32 \mathbf{C} 35B3534A3432D33 \mathbf{C} 36B3635A3534D34 \mathbf{C} 37B3736A36 35D35 \mathbf{C} B38 D36 35 34 A37 32Η 12B398 A387D37 Н B40 128 D38 13A39

H H H H H H H	13 13 28 28 36 37 38 39	B41 B42 B43 B44 B45 B46 B47 B48 B49	$ \begin{array}{r} 12 \\ 12 \\ 27 \\ 27 \\ 35 \\ 36 \\ 37 \\ 35 \\ 35 \\ \end{array} $	A40 A41 A42 A43 A44 A45 A46 A47 A48		D39 D40 D41 D42 D43 D44 D45 D46 D47
H H H H H H H	$35 \\ 30 \\ 33 \\ 32 \\ 31 \\ 9 \\ 10 \\ 11 \\ 7$	 B50 B51 B52 B53 B54 B55 B56 B57 B58 	$34 \\ 29 \\ 32 \\ 31 \\ 30 \\ 8 \\ 9 \\ 10 \\ 6$	A49 A50 A51 A52 A53 A54 A55 A56 A57	$32 \\ 27 \\ 31 \\ 30 \\ 29 \\ 7 \\ 8 \\ 9 \\ 1$	D48 D49 D50 D51 D52 D53 D54 D55 D56
H H H H H H H H H	$ \begin{array}{r} 1 \\ 5 \\ 4 \\ 3 \\ 2 \\ 15 \\ 19 \\ 18 \\ 22 \\ 26 \\ \end{array} $	$\begin{array}{c} {\rm B59} \\ {\rm B60} \\ {\rm B61} \\ {\rm B62} \\ {\rm B63} \\ {\rm B63} \\ {\rm B64} \\ {\rm B65} \\ {\rm B66} \\ {\rm B67} \\ {\rm B68} \end{array}$	$5 \\ 4 \\ 3 \\ 2 \\ 1 \\ 14 \\ 18 \\ 17 \\ 17 \\ 25$	A58 A59 A60 A61 A62 A63 A64 A65 A66 A67	$egin{array}{c} 4 \\ 3 \\ 2 \\ 1 \\ 5 \\ 12 \\ 17 \\ 16 \\ 16 \\ 24 \end{array}$	$\begin{array}{c} {\rm D57} \\ {\rm D58} \\ {\rm D59} \\ {\rm D60} \\ {\rm D61} \\ {\rm D62} \\ {\rm D63} \\ {\rm D64} \\ {\rm D65} \\ {\rm D66} \end{array}$
H H B B B B B B B B B B B B B B B B B B	25 23 27 1 2 3 4 5 6	B69 B70 B71	$24 \\ 20 \\ 24 \\ 1.4 \\ 1.4 \\ 1.4 \\ 2.0 \\ 2.0 \\ 2.0 \\$	A68 A69 A70 243214 238570 246307 238444 845985 835375	23 16 23 40 04 75 47 51 53	D67 D68 D69
B B B B B B B B B B B B B B B B B B B	7 8 9 10 11 12 13 14 15 16 17		1.4 1.4 1.4 1.4 1.5 1.5 1.5 1.3 1.3 1.4 1.3	287348 288827 226889 230074 117223 386148 918810 918810 949403 156888 907161	35 78 92 40 38 32 16 03 33 37 10	
B B B B B B B B B B B B B	 18 19 20 21 22 23 24 25 26 27 28 29 		1.3 1.4 1.4 1.3 1.3 1.3 1.4 1.3 1.5 1.5 1.5 1.5 1.4	887523 490770 156962 855752 949333 918888 060932 887595 287665 386106 117341 288809	 39 30 20 27 30 39 26 51 54 56 18 33 	
B B B B B B B B B B B B	21 22 23 24 25 26 27 28 29 30 31		1.3 1.3 1.3 1.4 1.3 1.5 1.5 1.5 1.4 1.4 1.4	855752 949333 918888 060932 887595 287665 386106 117341 288809 226932 230125	27 30 39 26 51 54 66 18 93 26 51	

B32	1.42477709
B33	2.07809077
B34	2.08630376
B35	1.42417847
B36	1.42431745
B37	1.42386359
B38	1.42384347
B39	1.09423424
B40	1.09161746
B41	1.09116815
B42	1.09189287
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B44	1.09189298
B45	1.09116870
B46	1.07827234
B47	1.07838995
B48	1.07857541
B49	1.07838288
B50	1.07837063
B51	1.07899004
B52	1.07807904
B53	1.07848088
B54	1.07839866
B55	1.07898980
B56	1.07839841
B57	1.07847960
B58	1.07808027
B59	1.07827298
B60	1.07836994
B61 DC0	1.07838285
B62	1.07857378
B03 D64	1.07838804
D04 D65	1.00332239
B00 DCC	1.08438937
D00 D67	
D07 B68	1.00404407 1.08350663
B60	1.08359005
B70	1.0843883 1.08339479
B70 B71	1.00332472
A1	107 98832088
A2	108.00649939
A2 A3	107 99419900
A4	70 09766200
A5	157.37258679
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A9	107.81760282
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A11	112.99631494
A12	112.09494024
A13	121.69854015
A14	120.02011915
A15	119.53931805
A16	121.29465590
A17	117.72001719
A18	133.77882278
A19	106.67933896
A20	108.61749139
A21	133.77947148
A22	120.02032308
A23	119.02714625

A24	122.39844197
A25	121.69533968
A26	110.61033671
A27	112.09462869
A28	124.92639957
A29	108.75131922
A30	107.81771939
A31	107 89077475
A 32	70 03962299
A 33	108 52747033
A 94	60.06006657
A34 A 25	108 01020745
A30 A90	108.01020740
A36	107.98846118
A37	70.07751394
A38	106.94805730
A39	111.10363706
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A41	111.12988947
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A43	111.13050995
A44	110.22817317
A45	125.99759718
A46	125,99017943
A47	125.96907648
A48	125 95822504
A 40	124 50850256
A49 A50	124.09000200
A50 A51	125.30226934
A51 A59	120.42000194
A52	120.14780303
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A61	125.96936989
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A63	119,93178095
A64	118.77656718
A 65	121 51664779
A 66	125 29231743
A67	120.25201140 120.76199417
A68	118 89440004
A00 A60	120.04155665
A09	120.04100000
A70	100.47738774
DI	-0.02630539
D2	0.05761447
D3	-60.01192633
D4	-48.53390094
D5	-163.88864144
D6	59.55195094
D7	0.18507789
D8	-0.27879090
D9	-124.18300498
D10	22.08131956
D11	-103.72104496
D12	41.48335946
D13	177.91113020
D14	0.01685770
D15	0.07600694
D10	0.07090024
D10	-0.04130208

D17	179.36281550
D18	-179.40764070
D19	179.61525020
D20	0.00403170
D21	-179.36810327
D22	0 14227371
D22 D23	-0.18335598
D20	-17790806203
D24 D25	85 50612133
D26	-41 49644772
D20 D27	-41.43044172 71.01720070
D21 D28	176 23860852
D20	0.27802356
D23	0.21002000
D30 D21	-0.20210300
D31 D22	101 02057051
D32 D22	1121.03037231
D33	
D34	-00.00302139
D30	0.01030094 192 12785780
D30 D37	120.13703700
D38	55 05015065
D30	$176 \ 25827364$
D33 D40	-64 81624883
D40 D41	$177 \ 44421629$
D41 D42	
D42 D43	57 14602374
D43 D44	$118\ 86515404$
D44 D45	178 88926613
D46	179.32636937
D47	-119.12026929
D48	2.51618449
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D50	178.67415052
D51	178.53923948
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D55	-178.53703408
D56	75.31266505
D57	-178.88039134
D58	-178.91097519
D59	-179.06827109
D60	-179.31952094
D61	-178.89008234
D62	-3.04587300
D63	179.65804780
D64	179.48068500
D65	-179.11727176
D66	179.61536151
D67	179.56466154
D68	-0.32400050
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C.3.3 Oxidized with the switching field

\$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 \$END \$CONTRL SCFTYP=UHF RUNTYP=ENERGY COORD=CART MAXIT=200 ICHARG=1 MULT=2 \$END \$SCF CONV=1d-5 \$END

\$SYS \$GUH	TEM MW ESS GUE	VORDS=125 \$END SS=MOREAD NORB=	154 \$END	
SEF1	ELD EV.	EC(1) = 0.012, 0, 0	\$EIND	
\$DAT	^T A			
Ci	e			
C	6.0	7 016/199199	2 2076022018	0 5072077971
C	6.0	-7.0104136163 7.4340771815	3.2970923910 2.0256357170	-0.3073977271
C	6.0	7 6020874282	1 2066451046	-0.9525252400
C	0.0 6.0	7 2024502528	1.2000431940 1.0749191924	1 2000565027
C	0.0 6.0	-7.5224095056	2 2662205172	1.3200303237
C FF	0.0	-0.9471010149 5 2010607225	3.2003003170 1.7794114596	0.902080071
гь С	20.0	- 0.3919097223	1.7724114060	1 1525056060
C	6.0 6.0		0.9933743009	1.1030900900
č	6.0 C 0	-3.8397291022	0.0309030307	0.1/09132103
č	6.0 C 0	-3.7300118080	0.0044309455	-1.0853271419
C	6.0	-3.3090857503	1.9970151124	-0.8874008254
C	6.0	-3.1651263368	2.2031902716	0.4980489998
C	6.0	-4.1825291179	-1.4323827937	0.3857713847
C	6.0	-4.6449895372	-1.7394846179	1.8187057221
C	6.0	-3.0349096885	-2.3613731012	-0.0101268594
C	6.0	-1.7026698709	-2.0304007736	0.2382634963
C	6.0	-0.6910355827	-2.9426404074	-0.0952921515
C	6.0	-1.0193547461	-4.1832452497	-0.6753915221
C	6.0	-2.3402489430	-4.5334346940	-0.9322290570
C	6.0	-3.3305196182	-3.6105304849	-0.5924414059
C	6.0	0.7470956660	-2.9115041801	0.0182150217
С	6.0	1.2261565834	-4.1389084714	-0.4981827053
Ν	7.0	0.1513307286	-4.8847051730	-0.9158704840
\mathbf{C}	6.0	1.6523068119	-1.9637850016	0.5022732126
\mathbf{C}	6.0	3.0224420195	-2.2410223927	0.4692922306
С	6.0	3.4716334434	-3.4666583804	-0.0478372440
\mathbf{C}	6.0	2.5846911645	-4.4286020315	-0.5336788540
\mathbf{C}	6.0	4.0289289159	-1.2256766570	1.0066541542
\mathbf{C}	6.0	4.0015990509	-1.1082050710	2.5348268861
\mathbf{C}	6.0	3.8103408451	0.0964498039	0.3107374146
\mathbf{C}	6.0	3.8242564936	0.2482520302	-1.1780554296
\mathbf{C}	6.0	3.4917670902	1.5485110280	-1.4896045088
\mathbf{C}	6.0	3.2341390979	2.2392065602	-0.2573239150
\mathbf{C}	6.0	3.4914027300	1.3420773883	0.8307489367
\mathbf{FE}	26.0	5.5325414044	1.6799706686	-0.3172045050
\mathbf{C}	6.0	7.0703131731	3.3892489570	-0.3373824131
\mathbf{C}	6.0	7.2967943242	2.5446026660	-1.4464127167
\mathbf{C}	6.0	7.6334514836	1.2584859453	-0.9621685910
\mathbf{C}	6.0	7.5948216759	1.3038764328	0.4520676530
\mathbf{C}	6.0	7.2404084175	2.6198243277	0.8351589625
Η	1.0	-5.0121480360	-1.6674302799	-0.2756822195
Η	1.0	-5.4978591682	-1.1283156865	2.0957103062
Η	1.0	-4.9342894684	-2.7809290929	1.9048210945
Η	1.0	-3.8500363991	-1.5617271633	2.5355328633
Η	1.0	4.7561958075	-0.4164253809	2.8981597299
Η	1.0	3.0313148486	-0.7810689963	2.8910471314
Η	1.0	4.2003350598	-2.0783484152	2.9717129544
Η	1.0	7.2709938506	2.8395873181	-2.4771055680
Η	1.0	7.9207503091	0.4165945825	-1.5607575043
Η	1.0	7.8510648324	0.5032585646	1.1176273446
Η	1.0	7.1659244341	2.9816591216	1.8418532862
Н	1.0	6.8384358054	4.4352227018	-0.3780433307
Н	1.0	4.0137730917	-0.5517477593	-1.8664813614
Н	1.0	3.3614179790	1.5816061870	1.8665814483
Н	1.0	2.9014834716	3.2552259239	-0.1662515367
Н	1.0	3.3720372394	1.9589171524	-2.4723755406
Н	1.0	-3.9070406727	0.1952101872	-2.0343293871

н		1 0	21	3266	85791	9 79	000/1	7754	1 6570	718170
11 11		1.0	-3.1	5400	00721	2.12	2241	0150	-1.0378	047965
H		1.0	-2.8	5913	37705	3.11	3789	2158	0.9756	947365
Н		1.0	-3.4	7025	62079	0.83	37342	2782	2.2143	923092
Н		1.0	-6.8	1838	35639	4.14	8949	6653	-1.1290	0736915
Η		1.0	-6.6	8674	10254	4.08	89078	5663	1.5388	374938
Η		1.0	-7.3	9795	01662	1.64	3995	4421	2.3465	089600
Η		1.0	-7.9	7028	55327	0.19	01845	2882	0.1760	687250
Η		1.0	-7.6	0916	21795	1.74	10437	3673	-1.9712	2140124
Η		1.0	-1.4	5729	80942	-1.0	78093	8204	0.6722	033015
Η		1.0	-4.3	5835	51000	-3.8	64152	1594	-0.7863	8771279
Η		1.0	-2.5	9776	08888	-5.4	76429	3826	-1.3800	0560981
Η		1.0	0.20	0052	69185	-5.80	06956	4576	-1.2760	0495704
Η		1.0	2.98	5141	39808	-5.3	61807	1998	-0.9214	1062983
Н		1.0	4.52	2674	57220	-3.6	79714	3631	-0.0642	2377320
Н		1.0	1.28	8856	57454	-1.0	32900	6146	0.9032	498809
Н		1.0	5.0	1770	75177	-1.5'	72751	9512	0.7142	901192
\$EN	ND									
\$VI	EC									
1	1	1.541	80469	E-08		-1.011'	75600	E-08		
1	2	3.383	23391	E-07		1.1293	32931	E-07		
1	3	1.280	07430	E-07		-6.1532	23336	E-08		
	-									
•										
•										
•										
•										
5/10	50	7 9 9 9	22144	F 05		7 7940	09119	F 04		
5416 5416	59 59	1.400	47062	ы—00 Б ос	• • • •	1 2665	77594	E-04 E 04		
0410 0410	- UU	-9.030	947003	<u>в</u> -00	• • •	1.3007	1054	Ľ-04		
) JEI	ND									

C.3.4 Oxidized with the clock field

С	6.0	-7.0164138183	3.2976923918	-0.5073977271
С	6.0	-7.4340771815	2.0256357170	-0.9523252460
С	6.0	-7.6229874382	1.2066451946	0.1819875317
С	6.0	-7.3224593538	1.9742121234	1.3288565237
\mathbf{C}	6.0	-6.9471510149	3.2663805178	0.9020856871
\mathbf{FE}	26.0	-5.3919697225	1.7724114586	0.0816213180
С	6.0	-3.4918419568	0.9933743069	1.1535956960
\mathbf{C}	6.0	-3.8397291622	0.0309630307	0.1759132153
С	6.0	-3.7300118086	0.6644309455	-1.0853271419
\mathbf{C}	6.0	-3.3090857563	1.9970151124	-0.8874008254
С	6.0	-3.1651263368	2.2031902716	0.4980489998
\mathbf{C}	6.0	-4.1825291179	-1.4323827937	0.3857713847
\mathbf{C}	6.0	-4.6449895372	-1.7394846179	1.8187057221
С	6.0	-3.0349096885	-2.3613731012	-0.0101268594
\mathbf{C}	6.0	-1.7026698709	-2.0304007736	0.2382634963
\mathbf{C}	6.0	-0.6910355827	-2.9426404074	-0.0952921515
С	6.0	-1.0193547461	-4.1832452497	-0.6753915221
С	6.0	-2.3402489430	-4.5334346940	-0.9322290570
С	6.0	-3.3305196182	-3.6105304849	-0.5924414059

С		6.0	0 7470956660	$-2 \ 9115041801$	0 0182150217
a		0.0	0.1410550000	-2.3110041001	0.0102100217
U.		0.0	1.2201000834	-4.1389084714	-0.4981827053
Ν		7.0	0.1513307286	-4.8847051730	-0.9158704840
\mathbf{C}		6.0	1.6523068119	-1.9637850016	0.5022732126
\mathbf{C}		6.0	3.0224420195	-2.2410223927	0.4692922306
\mathbf{C}		6.0	3.4716334434	-3.4666583804	-0.0478372440
č		6.0	2 5846011645	4 4286020315	0.5336788540
ä		0.0	4 000000150	1.0056766570	1 00000700040
Č		6.0	4.0289289159	-1.2250700570	1.0066541542
С		6.0	4.0015990509	-1.1082050710	2.5348268861
\mathbf{C}		6.0	3.8103408451	0.0964498039	0.3107374146
\mathbf{C}		6.0	3.8242564936	0.2482520302	-1.1780554296
\mathbf{C}		6.0	3.4917670902	1.5485110280	-1.4896045088
\mathbf{C}		6.0	3.2341390979	2.2392065602	-0.2573239150
č		6.0	3 4014027300	1 3420773883	0.8307480367
		0.0	5.4914027500	1.5420775885	0.0307409307
FE	1	20.0	5.5325414044	1.0799700080	-0.3172045050
C		6.0	7.0703131731	3.3892489570	-0.3373824131
\mathbf{C}		6.0	7.2967943242	2.5446026660	-1.4464127167
\mathbf{C}		6.0	7.6334514836	1.2584859453	-0.9621685910
\mathbf{C}		6.0	7.5948216759	1.3038764328	0.4520676530
\mathbf{C}		6.0	7.2404084175	2.6198243277	0.8351589625
й		1.0	-5.0121480360	-1.6674302799	-0.2756822195
ц		1.0	5 4078501689	1 1999156965	2 0057102062
11		1.0	-0.4976091062	-1.1203130003	2.0957105002
н		1.0	-4.9342894684	-2.7809290929	1.9048210945
Н		1.0	-3.8500363991	-1.5617271633	2.5355328633
Н		1.0	4.7561958075	-0.4164253809	2.8981597299
Η		1.0	3.0313148486	-0.7810689963	2.8910471314
Η		1.0	4.2003350598	-2.0783484152	2.9717129544
Η		1.0	7.2709938506	2.8395873181	-2.4771055680
Η		1.0	7.9207503091	0.4165945825	-1.5607575043
Н		1.0	7.8510648324	0.5032585646	1.1176273446
Н		1 0	$7\ 1659244341$	2 9816591216	1 8418532862
н		1.0	6 838/35805/	4 4352227018	-0.3780433307
ц		1.0	4.0137730017	0.5517477503	
11		1.0	9.9614170700	-0.0011411090	1 0004013014
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Η		1.0	-6.8183835639	4.1489496653	-1.1290736915
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ц		1.0	7.0702855327	0 1018459889	0 1760687250
11 11		1.0	7 6001601705	1.7404272672	1 0719140194
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Η		1.0	0.2005269185	-5.8069564576	-1.2760495704
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Η		1.0	1.2885657454	-1.0329006146	0.9032498809
Н		1.0	5.0177075177	-1.5727519512	0.7142901192
\$F1	ND				
\$V/	EC				
ψv. 1	1	1 5/18	80469E_08	1 01175600E_08	
1	1 0	2 2020	994091-00 93301F 07	1 19039091 07	
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T	3	1.2800	01430E-01	-0.1002030E-08	
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C.3.5 Oxidized with the driver

\$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 \$END \$CONTRL SCFTYP=UHF RUNTYP=ENERGY COORD=CART MAXIT=200 ICHARG=1 MULT=2 \$END \$SCF CONV=1d-5 \$END \$SYSTEM MWORDS=125 \$END \$GUESS GUESS=MOREAD NORB=154 \$END

\$DATA

Title

CI				
\mathbf{C}	6.0	-7.0164138183	3.2976923918	-0.5073977271
\mathbf{C}	6.0	-7.4340771815	2.0256357170	-0.9523252460
\mathbf{C}	6.0	-7.6229874382	1.2066451946	0.1819875317
\mathbf{C}	6.0	-7.3224593538	1.9742121234	1.3288565237
\mathbf{C}	6.0	-6.9471510149	3.2663805178	0.9020856871
\mathbf{FE}	26.0	-5.3919697225	1.7724114586	0.0816213180
\mathbf{C}	6.0	-3.4918419568	0.9933743069	1.1535956960
\mathbf{C}	6.0	-3.8397291622	0.0309630307	0.1759132153
\mathbf{C}	6.0	-3.7300118086	0.6644309455	-1.0853271419
С	6.0	-3.3090857563	1.9970151124	-0.8874008254
\mathbf{C}	6.0	-3.1651263368	2.2031902716	0.4980489998
\mathbf{C}	6.0	-4.1825291179	-1.4323827937	0.3857713847
\mathbf{C}	6.0	-4.6449895372	-1.7394846179	1.8187057221
\mathbf{C}	6.0	-3.0349096885	-2.3613731012	-0.0101268594
\mathbf{C}	6.0	-1.7026698709	-2.0304007736	0.2382634963
\mathbf{C}	6.0	-0.6910355827	-2.9426404074	-0.0952921515
\mathbf{C}	6.0	-1.0193547461	-4.1832452497	-0.6753915221
\mathbf{C}	6.0	-2.3402489430	-4.5334346940	-0.9322290570
С	6.0	-3.3305196182	-3.6105304849	-0.5924414059
С	6.0	0.7470956660	-2.9115041801	0.0182150217
\mathbf{C}	6.0	1.2261565834	-4.1389084714	-0.4981827053
Ν	7.0	0.1513307286	-4.8847051730	-0.9158704840
С	6.0	1.6523068119	-1.9637850016	0.5022732126
\mathbf{C}	6.0	3.0224420195	-2.2410223927	0.4692922306
С	6.0	3.4716334434	-3.4666583804	-0.0478372440
\mathbf{C}	6.0	2.5846911645	-4.4286020315	-0.5336788540
\mathbf{C}	6.0	4.0289289159	-1.2256766570	1.0066541542
С	6.0	4.0015990509	-1.1082050710	2.5348268861
\mathbf{C}	6.0	3.8103408451	0.0964498039	0.3107374146
\mathbf{C}	6.0	3.8242564936	0.2482520302	-1.1780554296
\mathbf{C}	6.0	3.4917670902	1.5485110280	-1.4896045088
\mathbf{C}	6.0	3.2341390979	2.2392065602	-0.2573239150
\mathbf{C}	6.0	3.4914027300	1.3420773883	0.8307489367
\mathbf{FE}	26.0	5.5325414044	1.6799706686	-0.3172045050
С	6.0	7.0703131731	3.3892489570	-0.3373824131
С	6.0	7.2967943242	2.5446026660	-1.4464127167
\mathbf{C}	6.0	7.6334514836	1.2584859453	-0.9621685910
\mathbf{C}	6.0	7.5948216759	1.3038764328	0.4520676530
\mathbf{C}	6.0	7.2404084175	2.6198243277	0.8351589625
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Н	1.0	-6	.6867	410	254	4	.08	907	8566	3	1.53883	74938
Н	1.0	-7	.3979	501	662	1	.64	399	5442	1	2.34650	89600
Н	1.0	-7	.9702	855	327	0	.19	184	5288	2	0.17606	87250
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Н	1.0	-1	.4572	980	942	- 1	1.07	809	3820	14	0.67220	33015
Н	1.0	-4	.3583	551	000	-:	3.86	415	2159	4	-0.78637	71279
Н	1.0	-2	.5977	608	888	-5	5.47	642	9382	6	-1.38005	60981
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H	1.0	5.	01770	075	177	- 1	1.57	275	1951	2	0.71429	01192
\$END												
\$VEC												
1 1	1.54	1804	69E-	-08	• • •	-1.	.011	756	00E-	-08		
1 2	3.38	3233	91E-	07	• • •	1.	129	329	31E-	-07		
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54150	7 9 3	2021	44F	05		7	734	091	19F	04		
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fragn	ame-		E 000		-0/1	UI .						
W101	-5 f	30106	391'	772	411	11						
W1H2	-10	2131	30	376	-0	12						
W1H3	-10	2131	30	763	-0	23						
fragn	ame=	-CO21	00.	100	0	.20						
CICI	5 53	2541	1.6'	799	10	60						
C102	10.3	131	30.3	76	-0.1	2						
C1O2	10.0	131	30.7	63	-0.2	23						
\$END	10.2	101	00.1	00	0.2							
\$WAT	EB1											
WATER	1 as	DR1										
COORD	INATI	ES(BO	HR.)									
W101	-5.5	39196		772	411	11	0.0	0	0			
W1H2	-10	.2131	30	376	-0	.12	0.	0ິ0	.0			
W1H3	-10	.2131	30.1	763	-0	.23	0.	0 0	.0			
STOP	10	.=101	00.	. 55	0		5.	. 0				
MONO	POLE	3										
W101	1.0 ().0										
W1H2 (0.0 ().0										
W1H3	0.0 (0.0										
STOP		-										
REPU	LSIVE	Е РОТ	ENTL	AL								
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```
W1O1
 0\,, 0
  STOP
  $END
  CO21
Carbon as DR2
COORDINATES(BOHR)
C1C1 5.532541 1.6799 10.60 0.0 0.0
C1O2 10.3131 30.376 -0.12 0.0 0.0
C1O3 10.2131 30.763 -0.23 0.0 0.0
STOP
  MONOPOLES
C1C1 0.0 0.0
C1O2 0.0 0.0
C1O3 0.0 0.0
STOP
  REPULSIVE POTENTIAL
C1C1
 0, 0
  STOP
  $END
  $FRGRPL
PAIR=WATER1 CO21
W101 C1C1 0 0
  STOP
  $END
```

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